

## Original Research Article

### **Interactive effect of salinity and potassium on maize yield**

#### **ABSTRACT**

A pot culture experiment was undertaken during January 2021, in the greenhouse of Institute of Agriculture Science (IAS), SOA Deemed to be University, Bhubaneswar, Odisha to study the interactive effects of salinity and potassium levels on biomass yield, accumulation and uptake of K and Na by maize. The experiment was conducted in 2-factorial CRD with three replications and twelve treatments consisting of two levels of irrigation water (SW0- normal water EC-0.5 dS m<sup>-1</sup>, SW1-saline water @ 8 dS m<sup>-1</sup>), two sources of K fertilizers (KCl and K<sub>2</sub>SO<sub>4</sub>), and three doses of K (K1-0, K2-60 and K3-120 kg ha<sup>-1</sup>). The results showed that continuous application of saline water (8 dS m<sup>-1</sup>) increased the soil ECe (Electrical Conductivity of a saturated soil Extract) by 182-360 % over normal water treatments. Application of K (120 kg ha<sup>-1</sup>) significantly decreased ECe by 29%, significantly increased the biomass yield (24%), K accumulation (14%), K uptake (41%) and decreased Na content (17%) over K control. The K<sup>+</sup>/Na<sup>+</sup> ratio in maize in saline treatments decreased by 64% over normal water. Application of K @ 120 kg ha<sup>-1</sup> significantly increased the K<sup>+</sup>/Na<sup>+</sup> ratio by 27% over K control. In saline stress environment, K<sup>+</sup>/Na<sup>+</sup> ratio in plant fairly correlated with biomass yield (R<sup>2</sup>=0.73), K content (R<sup>2</sup>=0.79), and Na content (R<sup>2</sup>=0.89) in the plant. K<sup>+</sup>/Na<sup>+</sup> ratio in soil failed to correlate well with biomass yield. K<sup>+</sup>/Na<sup>+</sup> ratio in the plant can be considered as the best indicator in evaluating crop performance in saline soil.

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Keyword: maize, salinity, potassium, biomass yield, K<sup>+</sup>/Na<sup>+</sup> ratio

# 1. INTRODUCTION

Salinity is an agro-environmental problem in arid and semi-arid regions of the world and has a significant impact on crop productivity. It reduces agricultural yields, prevents the use of previously cultivated land, and industrial raw material production. Human-induced salinity, combined with the natural, limits food production in most semi-arid regions of the world [1]. The majority of the world's water is salty containing 30 g of sodium chloride per liter. The availability of saline water is greater than freshwater [2]. Safe use of saline water in agriculture through soil management techniques will be beneficial to improve food security in developing countries.

Salinity affects plants in different ways such as osmotic stress, specific-ion toxicity, and nutritional disorders [3]. Osmotic stress is linked to ion accumulation in the soil solution whereas, nutritional imbalance and specific ion toxicity are connected to ion build-up, mainly sodium and chloride to toxic levels which interferes with the availability of other essential elements such as calcium and potassium [4],[5]. Ionic imbalance and ion toxicity lead to the substitution of potassium with sodium resulting reduction in the  $K^+/Na^+$  ratio in plants. Toxic levels of sodium in plant organs reduce plant growth and affect photosynthesis, respiration, starch metabolism, and nitrogen fixation leading to losses in crop yield [6].

Among the cereals, maize (*Zea mays* L.) is the third most important cereal crop, cultivated in 165 million hectares of land producing 850 million tons of grain with an average grain yield of 5200 kg ha<sup>-1</sup> [7]. Maize is typically classified as a moderately salt-sensitive crop and a salt-sensitive cereal, showing signs of stress including wilting even

when there is adequate soil moisture [8]. Seed germination and early seedling growth stages are more sensitive to salinity than later developmental stages [9]. Hyper-osmotic stress and toxic effects of sodium and chloride ions may delay or inhibit germination, reduce plant growth, and leads to severe nutritional imbalances in maize. Along with other mineral nutrients, potassium is very important for maintaining turgor and membrane potential, balancing osmotic potential, controlling stomatal movement, and activating enzymes [10]. The stress mitigating role of potassium in plants under drought, salinity, and pathogenic infection was reported by [11]. Thus, under saline conditions, potassium may be considered an important management strategy because it competes with sodium in plants.

Keeping in view the beneficial effects of potassium on crops under saline stress conditions, a pot culture experiment was undertaken: to study the interaction effects of salinity and potassium on maize biomass yield, nutrient accumulation, and uptake and to test the possibility that salinity damage can be reduced by elevating the rates of different potassic fertilizers.

## **2. MATERIALS AND METHODS**

### **2.1 Experimental Site**

The interaction effects of salinity and potassium on biomass yield, nutrient acquisition, and uptake of the maize crop were studied through a pot experiment. The experiment was conducted in the greenhouse of the Institute of Agriculture Science (IAS), SOA Deemed to be University, Bhubaneswar, Odisha during January 2021.

## 2.2 Experimental Design and Treatments

The experiment was conducted in a 2-factorial completely randomized design with three replications and 12 treatment combinations consisting of two levels of irrigation water ( SW0-normal water-EC- 0.5- dS m<sup>-1</sup> and SW1- saline water irrigation @8dS m<sup>-1</sup>), three doses of potassium (K1- control, K2- 60 kg ha<sup>-1</sup>, K3- 120 kg ha<sup>-1</sup>) and two sources of potassium fertilizers ( F1- KCl and F2- K<sub>2</sub>SO<sub>4</sub>).

The sandy loam saline soil used in the pot culture study was collected from Brahmagiri village in Puri district during October 2020. The field was mono-cropped with rice during the Kharif season. During Rabi season the land remained fallow because of high salinity and lack of good quality irrigation water. The surface soil (0-15 cm depth) in bulk was collected from the field and brought to the laboratory, air dried, processed, and used for pot culture study. The characteristics of soil are presented in table-1. The saline water used in the study was brought from the Bay of Bengal at Puri seashore. The EC of sea water was 35.5 dS m<sup>-1</sup> which was diluted to 8 dS m<sup>-1</sup> for use as irrigation water in the study.

The polythene-lined earthen pots were rinsed in 0.1N HCl followed by deionized water. Seven kg of air-dried soil was transferred into each pot. Each pot received a common dose of N @ 120 kg ha<sup>-1</sup> (on a weight basis) through DAP and granular urea. The nitrogen fertilizer was applied in two splits as 50% at sowing and 50% at knee high stage(vegetative stage) . A common dose of phosphorous @ 60 kg ha<sup>-1</sup> (on a weight basis) through DAP was applied at seeding time. Potassium through KCl and K<sub>2</sub>SO<sub>4</sub> was

applied in two splits 50% at seeding and 50% at knee high stage as per the treatments. Each pot received a chemical grade of gypsum @ 15 mg pot<sup>-1</sup> through a solution.

All the pots were saturated with normal water up to field capacity and 8 hybrid sweet corn seeds (CV.4226) of 90 days duration were sown on 9th January 2021. The emergence of seedlings was recorded on 16th January 2021. After seedling emergence, 4 seedlings were kept in each pot. First saline water irrigation was applied to SW1 treatments on 25th January 2021 up to saturation point. In SW0 treatments, the maize crop received irrigation through normal water. In SW1 treatments, the same quantity of irrigation water was applied through normal water and saline water alternatively. The crop received 14 saline water irrigation (250 ml in each time) during the crop growth period.

The sea water is alkaline in reaction (pH 7.99) and EC 35.5 dS m<sup>-1</sup>. The content of soluble ion in sea water was Na-5.1 gm l<sup>-1</sup>, K-0.20 gm l<sup>-1</sup>, Ca-0.19 gm l<sup>-1</sup> and Mg -0.56 gm l<sup>-1</sup>.

Plant protection measures were taken as and when required. The crop was harvested at 64<sup>th</sup> day of emergence. Plant samples were collected at harvest, processed, and analyzed for K and Na. Soil samples were collected at 21<sup>th</sup> day of seedling emergence and harvest, processed, and kept for analysis.

### **2.3 Soil and plant analysis**

The soil was analyzed in the laboratory following standard procedures. Particle size was determined by the Bouyoucos hydrometer as given by [12], pH by glass electrode with calomel as standard [13]. Electrical conductivity (EC) of soil was determined in 1:2 soil-water suspension by a conductivity meter (ELICOCM 180

conductivity meter) as suggested by [8]. The bulk density, particle density, and porosity were determined as per the methods outlined by [14]. The organic carbon content of the soil was estimated by the wet digestion method [15]. Available N in soil was determined by the modified alkaline permanganate method [16] and available P by Olsen's method [17]. Water-soluble K and Na were determined in a 1:5 (soil: water) ratio. The available K and Na which include water-soluble and exchangeable forms was extracted with neutral normal ammonium acetate and estimated with a flame photometer (Model: Sytonic128) [18]. The sodium and potassium content in the maize plant was estimated as per the method outlined by [13].

**Table 1. Physio-chemical properties of soil used in pot culture experiment**

<b>Parameter</b>	<b>Value</b>
pH	6.49
ECe (dS m <sup>-1</sup> )	2.57
Sand (%)	65.0
Silt (%)	20.0
Clay (%)	15.0
Texture	Sandy Loam
BD (gm cm <sup>-3</sup> )	1.21
PD (gm cm <sup>-3</sup> )	2.46
Porosity (%)	42.5
OC (%)	1.02
Organic matter (%)	1.75
Available N (kg ha <sup>-1</sup> )	112.0
Available P (kg ha <sup>-1</sup> )	8.50
Available K (mg kg <sup>-1</sup> )	298.37
Exch.K (mg kg <sup>-1</sup> )	270.52
Non exch.K (mg kg <sup>-1</sup> )	1043.63
Water soluble.K (mg kg <sup>-1</sup> )	27.85
Available Na (mg kg <sup>-1</sup> )	56.10
Water soluble Na (mg kg <sup>-1</sup> )	25.50
Exch.Na (mg kg <sup>-1</sup> )	30.60

### 3. Results and Discussion

### 3.1 Effects of saline water irrigation and potassium on soil ECe

Continuous use of saline water leads to the accumulation of neutral soluble salts in soil that impacts crop growth. Long-term application of saline water resulted in higher EC and accumulation of soluble salts in the soil rhizosphere which adversely affect plant growth [19].

Application of saline water significantly increased the ECe values over SW0 (normal water) by 182 % at 21d of seedling emergence and by 360 % at harvest. It varied between 5.82 to 7.39 dS m<sup>-1</sup> at 21 days and 7.40 to 12.33 dS m<sup>-1</sup> at harvest ( Table 2 and Fig.1 ). Similar results were reported by [20] and [21].

Application of K through K<sub>2</sub>SO<sub>4</sub> resulted in lower ECe (6%) as compared to KCl at both stages, although the values were non-significant. A decrease in ECe values in K<sub>2</sub>SO<sub>4</sub> treatments might be due to the presence of associate sulphate ion as against chloride ion. Several findings showed that the ECe of soil increases with increasing the concentration of chloride ions. Similar findings were reported by [22].

Application of different doses of K ( K<sub>0</sub>, K<sub>60</sub>, and K<sub>120</sub> ) significantly decreased the ECe at the harvest stage but was non-significant at 21 days of growth. Application of potassium @ 120 kg ha<sup>-1</sup> significantly reduced the ECe over potassium control treatment by 29% at harvest and 8% at 21 days of growth. This is possible since potassium and sodium although co-exist in soil exchange complex and soil solution, but cations exert an antagonistic effect.

Table 2 . Effect of salinity and potassium on soil ECe (dS m<sup>-1</sup>)

Treatment	21 Days	Harvest
	Tr. Mean	Tr. Mean
Sw0 F1 K1	2.30 <sup>d</sup>	2.04 <sup>d</sup>
SW0 F1 K2	2.40 <sup>d</sup>	2.36 <sup>d</sup>
SW0 F1 K3	2.68 <sup>d</sup>	2.75 <sup>d</sup>
SW0 F2 K1	2.12 <sup>d</sup>	1.75 <sup>d</sup>
SW0 F2 K2	2.30 <sup>d</sup>	2.22 <sup>d</sup>
SW0 F2 K3	2.60 <sup>d</sup>	2.20 <sup>d</sup>
SW1 F1 K1	7.39 <sup>a</sup>	12.33 <sup>a</sup>
SW1 F1 K2	7.10 <sup>ab</sup>	11.37 <sup>ab</sup>
SW1 F1 K3	6.47 <sup>bc</sup>	7.57 <sup>c</sup>
SW1 F2 K1	7.23 <sup>ab</sup>	12.10 <sup>a</sup>
SW1 F2 K2	6.59 <sup>abc</sup>	10.46 <sup>b</sup>
SW1 F2 K3	5.82 <sup>c</sup>	7.40 <sup>c</sup>

Factors	C.D (0.05)	C.D (0.05)
SW	0.336	0.503
F	NS	NS
SW X F	NS	NS
K	NS	0.616
SW X K	0.582	0.871
F X K	NS	NS
SW X F X K	NS	NS

Each value is an average of 3 replications. Figures not showing the same letter(s) in the column differ significantly at a 5% probability level according to Duncan's Multiple Range Test.

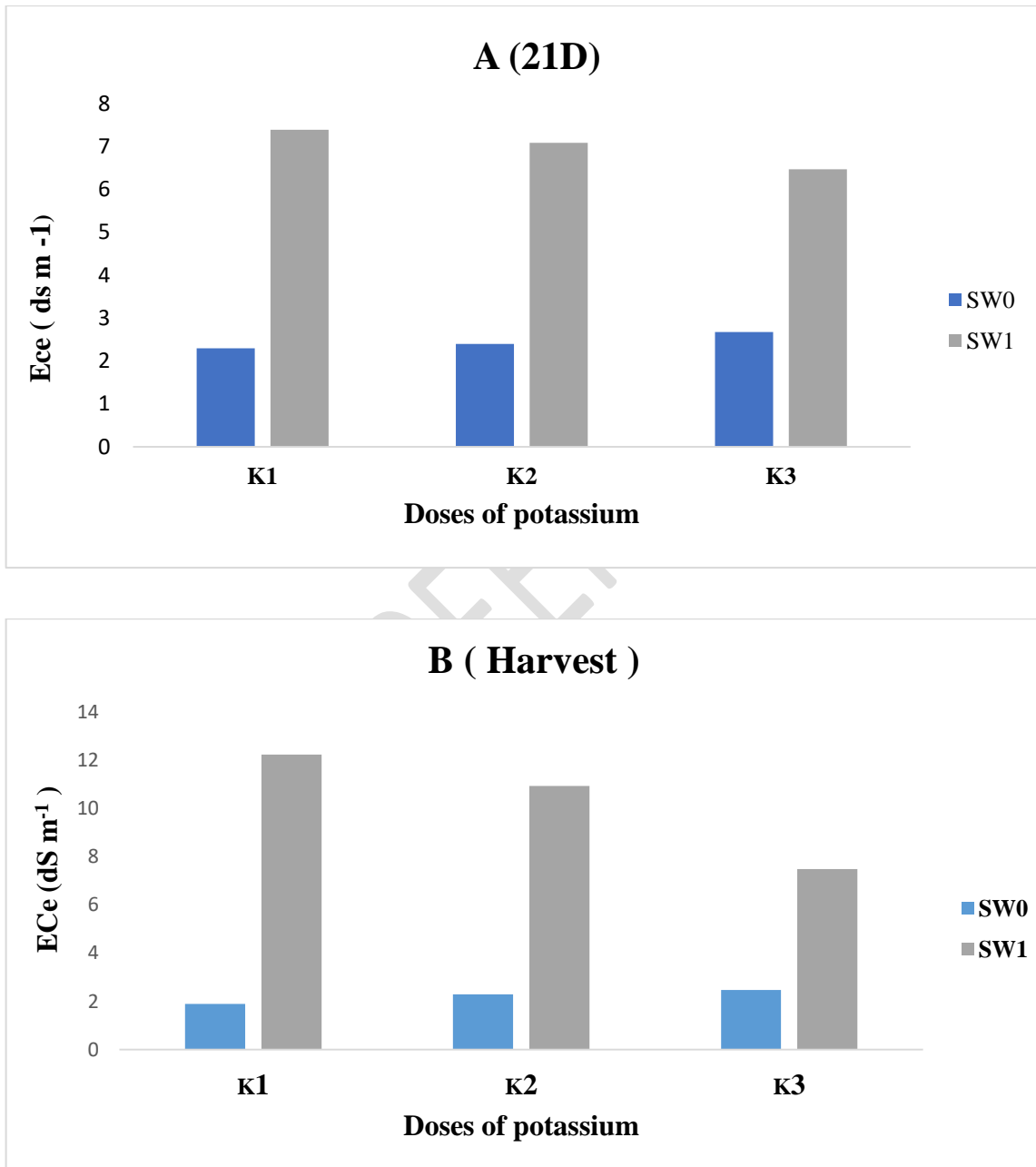


Fig 1. Effect of salinity and potassium on ECe (dS m<sup>-1</sup>) at 21 d (A) and harvest (B)

### 3.2 Effects of salinity and potassium on biomass yield

The biomass yield of maize was significantly influenced by the saline water irrigation and potassium doses (Table 3). Continuous use of saline water irrigation significantly increased the biomass yield by 11% over normal water treatment. Similar findings were reported by [23]. She reported that application of saline water irrigation increased the biomass yield of maize by 25% at 8 dS m<sup>-1</sup> and 17 % at 12 dS m<sup>-1</sup>, but decreased by 29 % at 16 dS m<sup>-1</sup>. Among the two K fertilizers tested, the addition of K<sub>2</sub>SO<sub>4</sub> recorded higher biomass yield (3%) as compared to KCl, because of lower electrical conductivity in K<sub>2</sub>SO<sub>4</sub> treatments than KCl treatments. Further, the presence of SO<sub>4</sub><sup>-2</sup> ion in K<sub>2</sub>SO<sub>4</sub> meets the S requirement of a crop, whereas Cl<sup>-</sup> ion in KCl is detrimental to maize crop.

Significant effect of potassium dose was observed in saline and non-saline treatments (Fig.2). When the K doses were increased from 0 to 60 -120 kg ha<sup>-1</sup>, the biomass yield increased by 34 to 37% in normal water and 11-24% in saline treatments. The reduction in the response of potassium application under a saline environment might be due to the adverse effect of sodium. Further, the data indicated that averaged over SW0 and SW1 treatments, the response to K application over potassium control was 21% in K2 (60 kg ha<sup>-1</sup>) and 24% in K3 (120 kg ha<sup>-1</sup>) indicating that the adverse effect of Na is suppressed at a higher level of potassium application.

Several authors reported that the application of K improved growth and yield under stress ( caused due by salinity ) possibly by resulting in photosynthesis [24]. The adverse effect of Na on plant growth and yield can be reduced by K application [25].

### **3.3 Effect of salinity and potassium on K and Na accumulation and their uptake**

## Accumulation of K and Na

The application of saline water and potassium significantly influenced the accumulation of K and Na in maize crops (Table 3). With the application of saline water, the content of K significantly decreased by 6% as compared to normal water irrigation whereas, Na content increased by 167% over normal water treatments (Table 4 and 5).

Application of a higher dose of potassium has a beneficial effect on K accumulation. It reduced the adverse effect of Na under a saline environment. When the K dose increased from 0 to 60 and 120 kg ha<sup>-1</sup>, the K content in maize increased by 5 and 14%, respectively, over K control. On the other hand, the accumulation of Na decreased by 17%, when the maize was fertilized with 120 kg ha<sup>-1</sup> K. Application of the recommended dose of K (@60 K kg ha<sup>-1</sup> for maize) did not affect much in reducing the adverse effect of Na.

There was no significant difference between the effect of KCl and K<sub>2</sub>SO<sub>4</sub> on K and Na accumulation in maize plants (Table 3). However, the application of K<sub>2</sub>SO<sub>4</sub> reduced Na accumulation by 17% over KCl. But, did not affect potassium accumulation.

## Uptake of K and Na

Application of saline water significantly increased the Na uptake whereas, K uptake was non-significant (Table 3). The uptake of K was increased by 4% and the Na uptake by 186% with application of saline water (SW1) over normal water (SW0) treatments might be due to higher biomass yield in saline treatments.

Application of K either through KCl or K<sub>2</sub>SO<sub>4</sub> did not affect much on K or Na uptake. Application of different doses of K significantly increased the K uptake. When the K dose increased from 0 to 60 and 120 kg ha<sup>-1</sup>, the K uptake was increased by 27 and 41%, respectively, over potassium control treatments.

On the other hand, K application suppressed the uptake of Na although higher biomass yield was recorded in K treatments. A non-significant increase in Na uptake by 11% in K<sub>2</sub> (60 kg ha<sup>-1</sup>) indicated that despite higher biomass production, application of K suppressed the uptake of Na.

Excessive buildup of Na<sup>+</sup> and Cl<sup>-</sup> ions in the root zone cause several nutritional imbalances in maize due to strong interference of these ions with other essential nutrients. Under saline conditions, nutrient imbalance reduced nutrient uptake including K<sup>+</sup> and ion toxicity is resulted because of high Na<sup>+</sup> and Cl<sup>-</sup> concentrations [26], [27] reported that the yield of wheat decreased with increasing salinity. By increasing salinity, the sodium content of leaf, stem, root, spike, and seed increased but potassium content decreased. Similar results were reported by [28], [29].

Table 3. Effect salinity and potassium on biomass yield, K & Na content %, K & Na uptake, K<sup>+</sup>/ Na<sup>+</sup> ratio in maize plant

Treatments	Treat. Mean					
	Biomass yield(g/pot)	K (%)	Na (%)	K uptake (g/pot )	Na uptake ( mg/pot )	K <sup>+</sup> / Na <sup>+</sup> ratio in maize
Sw0 F1 K1	150.7 <sup>c</sup>	2.25 <sup>bcd</sup>	0.03 <sup>c</sup>	3.39 <sup>e</sup>	45.21 <sup>b</sup>	43.84 <sup>a</sup>
SW0 F1 K2	205.5 <sup>ab</sup>	2.31 <sup>bcd</sup>	0.03 <sup>c</sup>	4.75 <sup>bc</sup>	61.52 <sup>b</sup>	49.19 <sup>a</sup>
SW0 F1 K3	210.0 <sup>ab</sup>	2.55 <sup>ab</sup>	0.03 <sup>c</sup>	5.34 <sup>ab</sup>	62.46 <sup>b</sup>	53.89 <sup>a</sup>
SW0 F2 K1	160.5 <sup>c</sup>	2.28 <sup>bcd</sup>	0.03 <sup>c</sup>	3.66 <sup>de</sup>	48.15 <sup>b</sup>	44.42 <sup>a</sup>
SW0 F2 K2	210.5 <sup>ab</sup>	2.50 <sup>abc</sup>	0.03 <sup>c</sup>	5.23 <sup>ab</sup>	62.08 <sup>b</sup>	51.34 <sup>a</sup>
SW0 F2 K3	215.0 <sup>ab</sup>	2.62 <sup>a</sup>	0.03 <sup>c</sup>	5.64 <sup>a</sup>	64.83 <sup>b</sup>	53.85 <sup>a</sup>
SW1 F1 K1	193.1 <sup>b</sup>	2.21 <sup>cd</sup>	0.09 <sup>a</sup>	4.27 <sup>cd</sup>	173.59 <sup>a</sup>	14.46 <sup>b</sup>
SW1 F1 K2	216.6 <sup>ab</sup>	2.25 <sup>bcd</sup>	0.08 <sup>ab</sup>	4.87 <sup>abc</sup>	173.79 <sup>a</sup>	17.18 <sup>b</sup>
SW1 F1 K3	220.7 <sup>a</sup>	2.50 <sup>abc</sup>	0.07 <sup>b</sup>	5.52 <sup>ab</sup>	154.49 <sup>a</sup>	20.87 <sup>b</sup>
SW1 F2 K1	200.1 <sup>ab</sup>	2.09 <sup>d</sup>	0.08 <sup>ab</sup>	4.18 <sup>cde</sup>	158.8 <sup>a</sup>	15.44 <sup>b</sup>
SW1 F2 K2	218.0 <sup>ab</sup>	2.25 <sup>bcd</sup>	0.08 <sup>ab</sup>	4.92 <sup>abc</sup>	174.4 <sup>a</sup>	16.43 <sup>b</sup>
SW1 F2 K3	225.5 <sup>a</sup>	2.39 <sup>abcd</sup>	0.07 <sup>b</sup>	5.39 <sup>ab</sup>	150.58 <sup>a</sup>	21.11 <sup>b</sup>
Factors	C.D (0.05)					
SW	9.618	0.106	0.006	NS	13.726	6.569
F	NS	NS	NS	NS	NS	NS
SW X F	NS	NS	NS	NS	NS	NS
K	11.779	0.130	NS	0.361	NS	NS
SW X K	16.658	NS	NS	NS	NS	NS
F X K	NS	NS	NS	NS	NS	NS
SW X F X K	NS	NS	NS	NS	NS	NS

Each value is an average of 3 replications.

Figures not showing the same letter(s) in the column differ significantly at a 5% probability level according to Duncan's Multiple Range Test.

Table 4. Interaction effect of salinity and different levels of potassium-on-potassium content (%) in maize plant

Sources	K1	K2	K3	Mean	% Decrease over SW0
SW0	2.27	2.41	2.58	2.42	-
SW1	2.15	2.25	2.45	2.28	5.79
Mean	2.21	2.33	2.52		
% Increase over k1	-	5.43	14.03		

Table 5. Interaction effect of salinity and different levels of potassium on Na content (%) in maize plant

Sources	K1	K2	K3	Mean	% Increase over SW0
SW0	0.03	0.03	0.03	0.03	-
SW1	0.09	0.08	0.07	0.08	166.67
Mean	0.06	0.06	0.05		
% Decrease over k1	-	-	16.67		

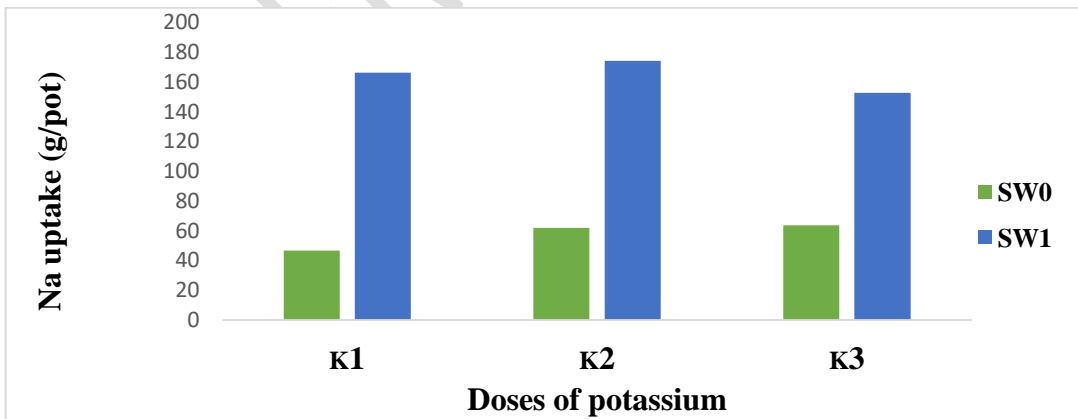
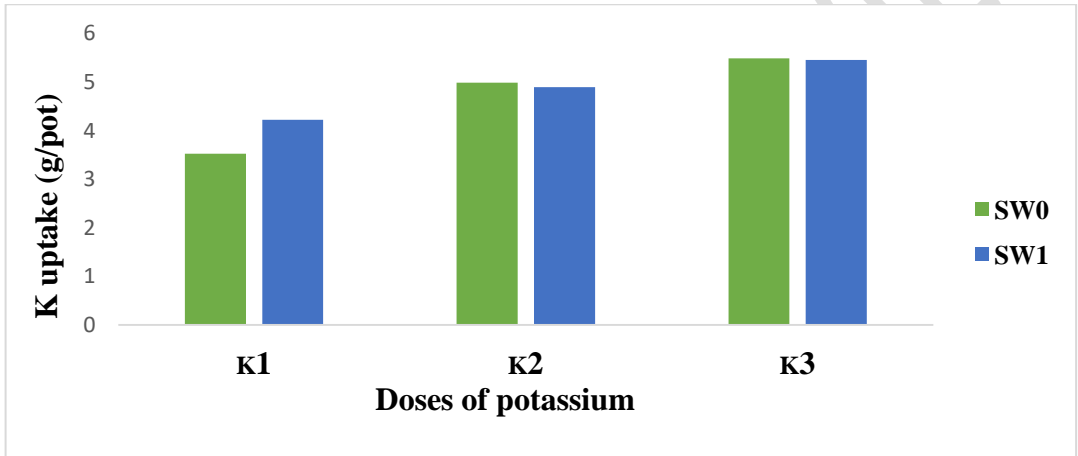
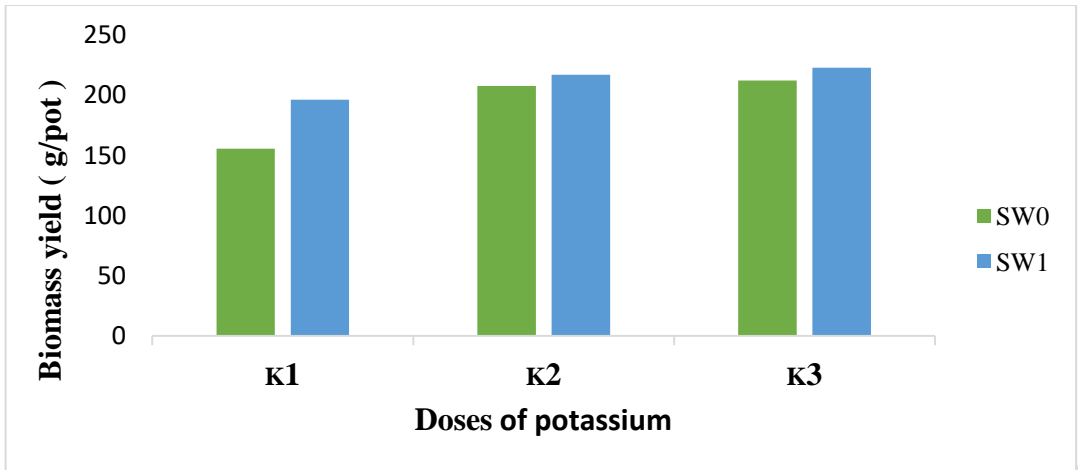


Fig 2. Effect of salinity and potassium on biomass yield, K uptake, Na uptake (g/pot)

\* SW0- normal water, SW1-saline water

### 3.4 Effect of salinity and potassium on $K^+/Na^+$ in maize plant

The  $K^+/Na^+$ , being another indicator (Except ECe) of the plant response to high salinity decreases in many plants after saline water irrigation [30].

The data presented in (Table 3) showed that the application of saline water significantly reduced the  $K^+/Na^+$  ratio in a plant. Under normal water irrigation, the ratio increased with the level of K application which ranged from 43.84 to 53.89 in F1 (KCl) and 44.42 to 53.85 in F2 ( $K_2SO_4$ ), but all the values were at par.

Application of saline water significantly decreased the  $K^+/Na^+$  ratio by 64% over SW0 (normal water) treatments. However, with the application of K fertilizers, there was an improvement in the ratio over potassium control treatments. The  $K^+/Na^+$  ratio increased by 14 and 27% when the K dose increased from 0 to 60 and 120  $kg\ ha^{-1}$ , respectively.

Similar observations were reported by Mohapatra [31] for rice and by Patel [23] for maize crops. Folkard *et al* [32] observed that the leaf  $K^+/Na^+$  ratio predicts salinity-induced yield loss in irrigated rice.

### 3.5 Correlation between soil properties and several plant parameters

A positive relationship exists between ECe at 21 days with biomass yield having an  $R^2$  value of 0.4868 in normal water treatments. However, under saline conditions, the biomass yield of maize significantly decreased with increasing ECe at 21 days with an  $R^2$  value of 0.741 (Fig.3).

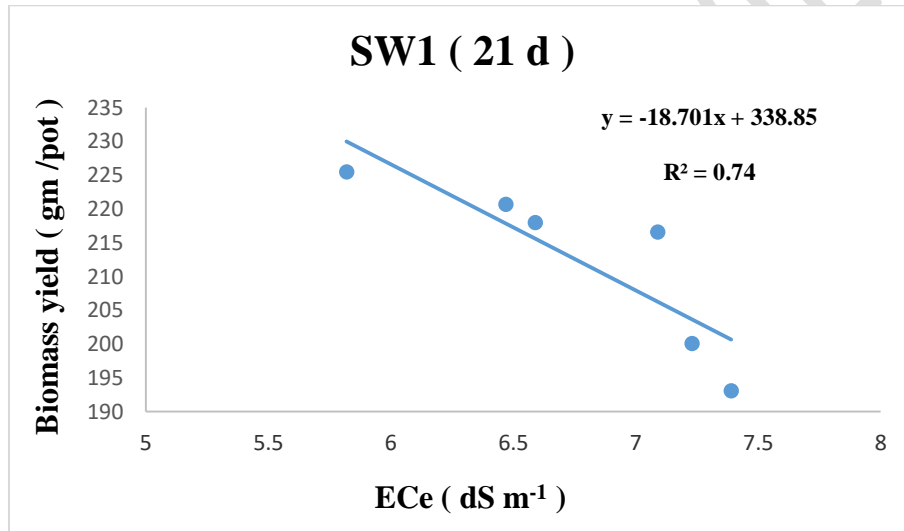
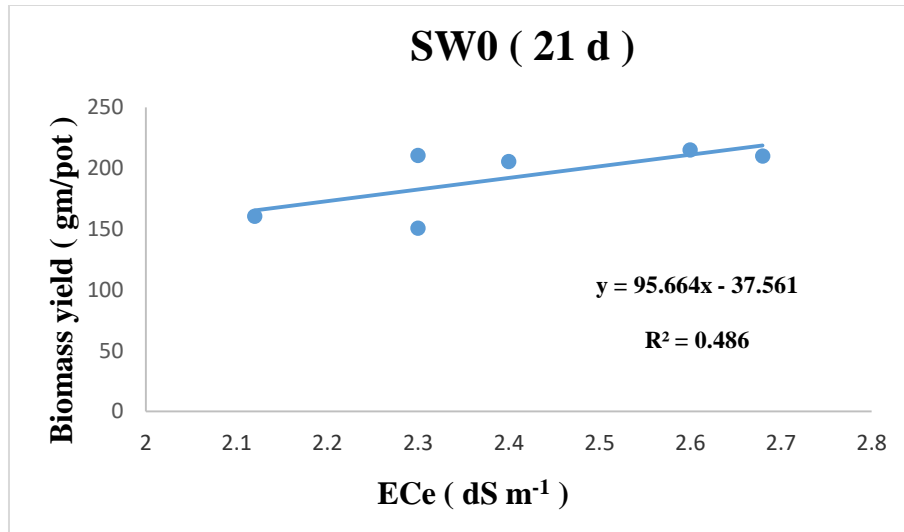


Fig 3. Correlation between ECe at 21 days and biomass yield in normal water (SW0) and saline ( SW1 ) treatments.

The  $K^+/Na^+$  ratio in maize fairly correlated with biomass yield both in normal water ( $R^2=0.896$ ) and saline treatments ( $R^2=0.73$ ) (Fig.4). However, the correlation between  $K^+/Na^+$  in soil and biomass yield was non-significant.

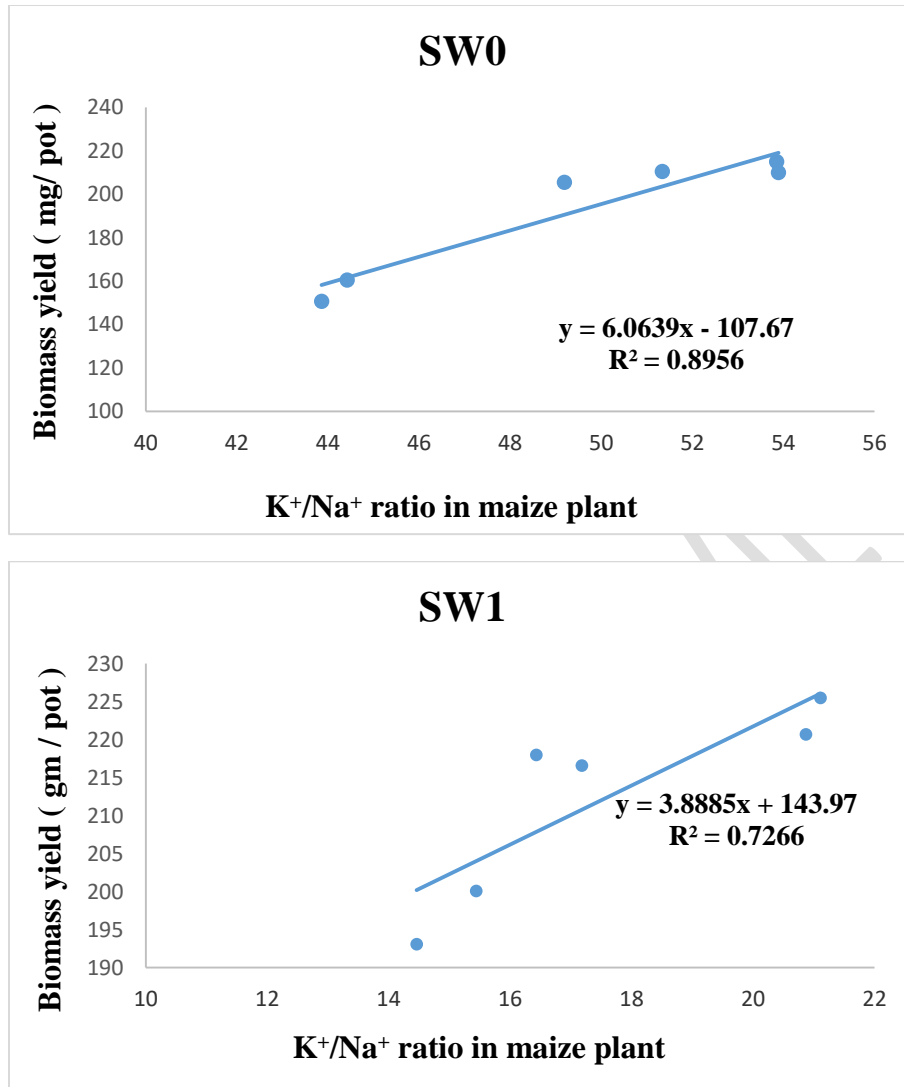


Fig 4. Correlation between  $K^+/Na^+$  ratio in maize plant and biomass yield in normal water (SW0) and saline ( SW1 ) treatments

The K content in the maize plant decreased with ECe at harvest, whereas, Na content increases. The data presented in fig.5 showed that a negative relationship exists between ECe ( at harvest ) and K content in plants with an  $R^2$  value of 0.83. On the other hand, a positive correlation ship exists between ECe (at harvest) and Na content in maize with an  $R^2$  value of 0.83.

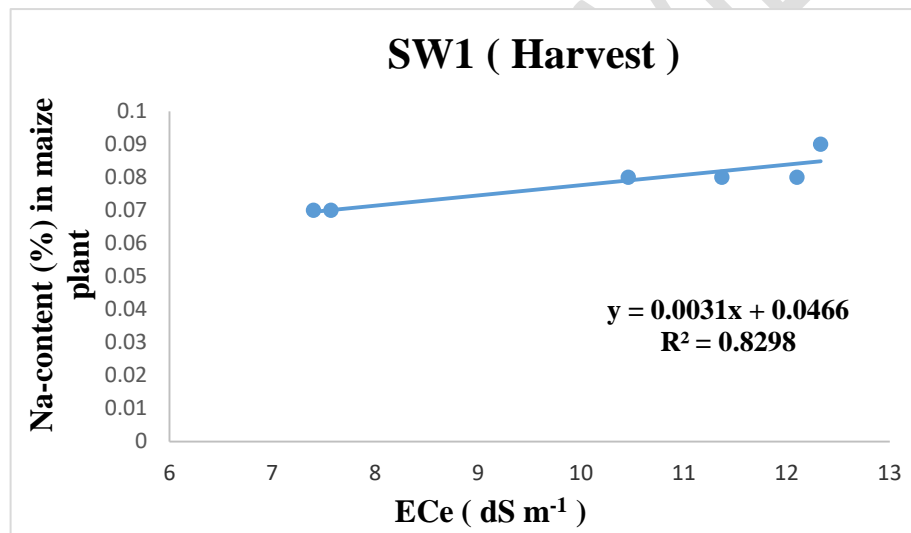
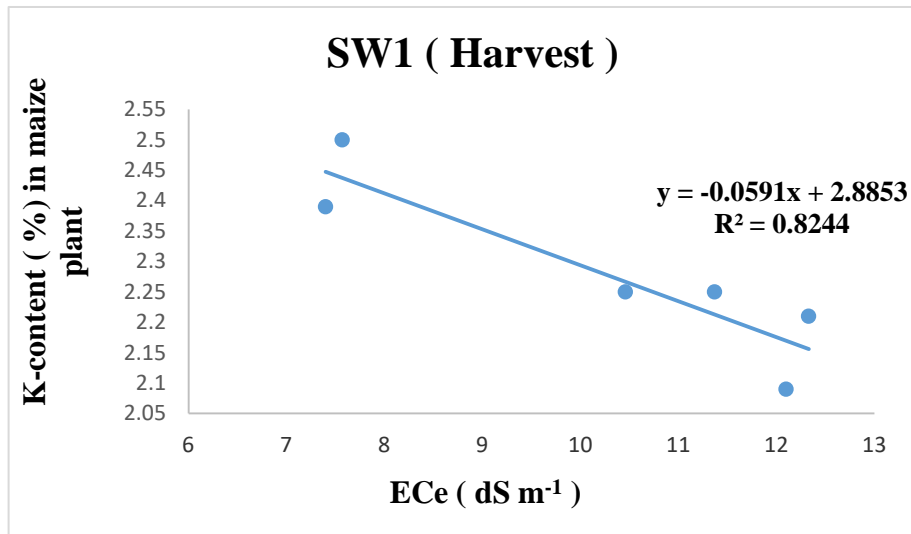


Fig 5. Correlation between ECe at harvest and K or Na content in maize plant in saline treatments ( SW1 )

Potassium content in maize increased with increasing in  $K^+/Na^+$  ratio in maize under saline environment (SW1) whereas, Na content decreases. A fair positive relationship exists between the  $K^+/Na^+$  ratio and  $K^+$  content in a plant (  $R^2$  value of 0.88) and (  $R^2$  value of 0.79) in **normal water** and saline treatments , respectively (Fig.6). On the other hand, the  $K^+/Na^+$  ratio in plants negatively correlated with Na content with an

$R^2$  value of 0.89 in saline treatments (Fig.7), indicating that Na content decreased with increasing the  $K^+/Na^+$  ratio.

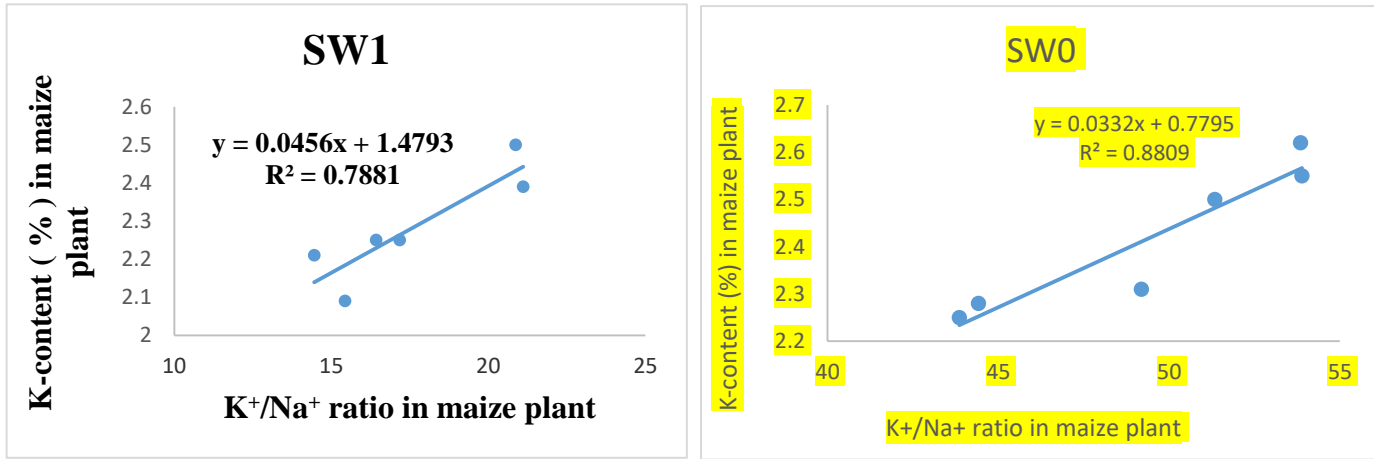


Fig 6. Correlation between  $K^+/Na^+$  ratio in plant and K content in maize plant in normal water (SW0) and saline treatments (SW1).

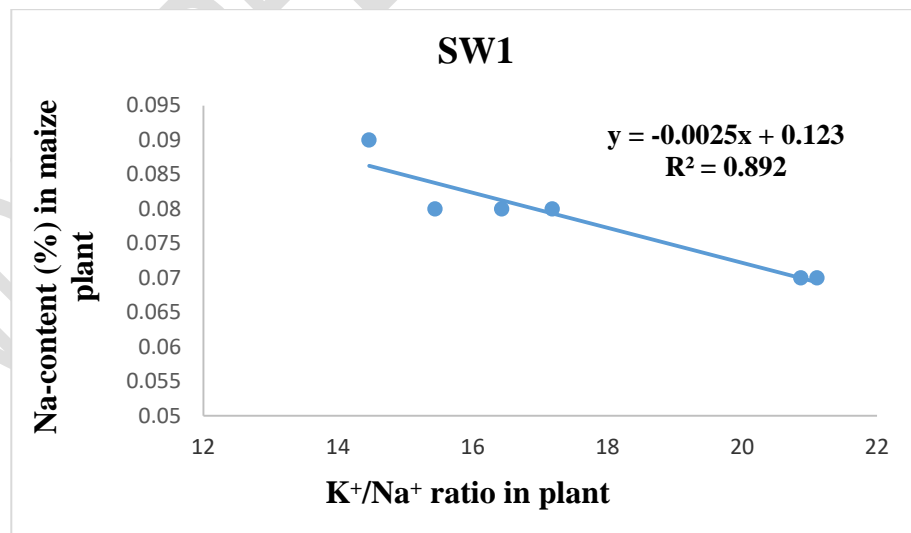


Fig 7. Correlation between  $K^+/Na^+$  ratio in plant and Na content in maize plant in normal saline treatments (SW1).

The  $K^+/Na^+$  ratio in maize increased with increasing the ECe (at harvest) with an  $R^2$  value of 0.57 in normal water treatments (SW0) whereas, the ratio decreases with increasing ECe in saline treatments ( $R^2=0.95$ ) (Fig.8). This showed that continuous addition of saline water to maize crops decreased the K absorption in plants leading to a lower  $K^+/Na^+$  ratio.

Several studies showed that they can be elevated by the addition of  $K^+$  to the substrate. Sodium concentration impairs  $K^+$  nutrition and harms uptake and translocation of  $K^+$  by plants grown in saline environment. Therefore, maintenance of  $K^+/Na^+$  ratio is an important guide under salt stress environment [33] and [29].

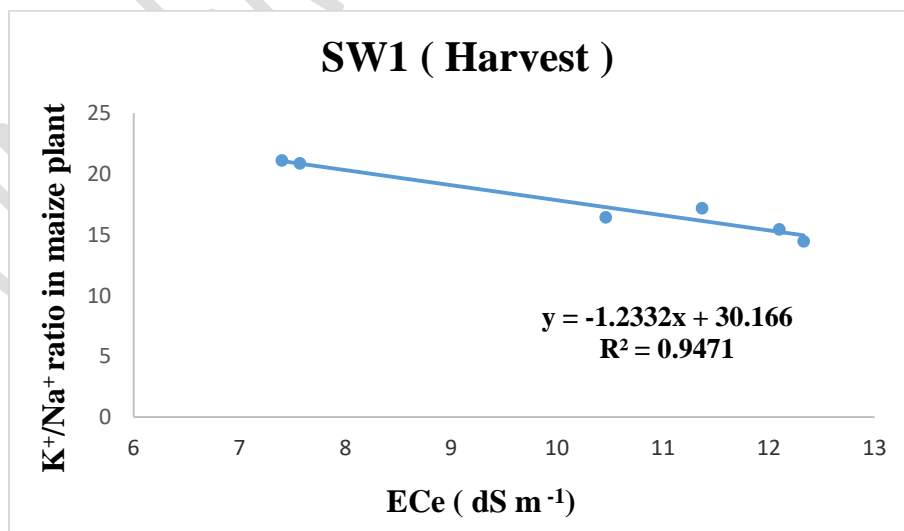
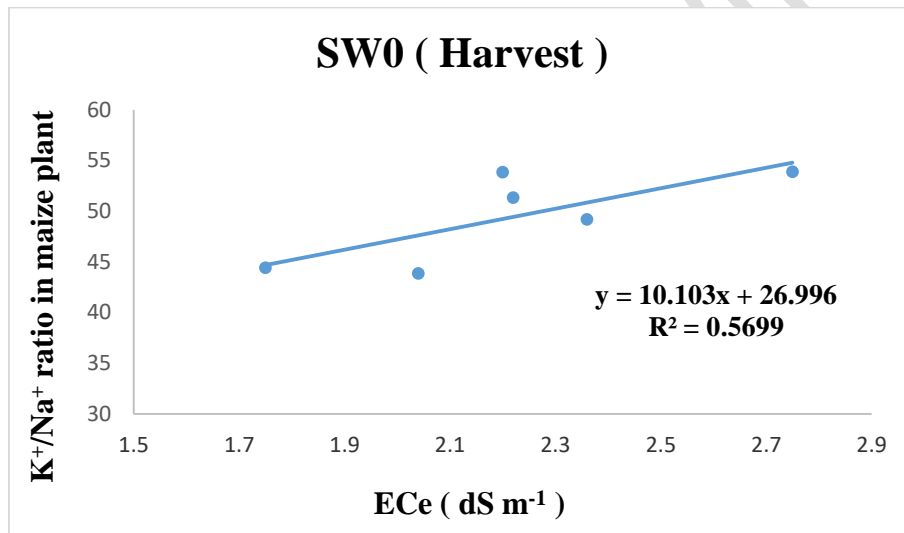


Fig 8. Correlation between ECe at harvest and  $K^+/Na^+$  ratio of maize plant in normal water (SW0) and saline (SW1) treatments

The absorption of  $K^+$  in maize is influenced by the  $K^+/Na^+$  ratio in the soil. Accumulation of K increased with increasing the  $K^+/Na^+$  ratio in the soil. The results of this study showed that a positive relationship exists between the  $K^+/Na^+$  ratio in plants and the  $K^+/Na^+$  ratio in soil (Fig.9).

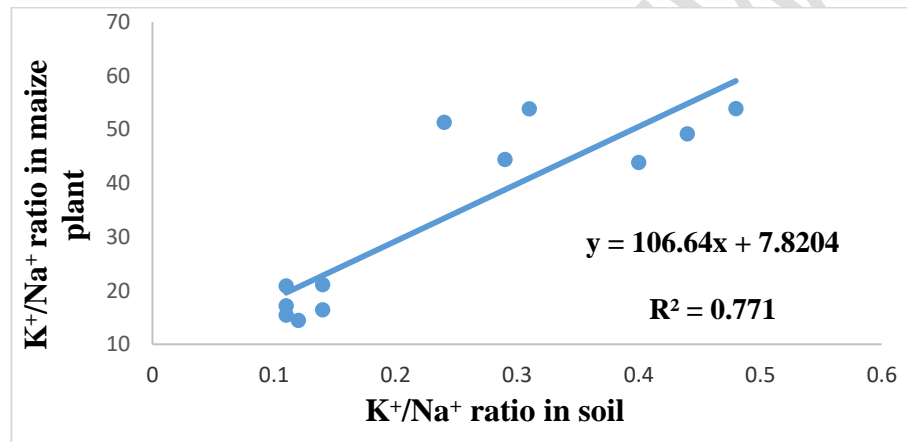


Fig 9. Correlation ship between  $K^+/Na^+$  ratio in soil and plan

#### 4. CONCLUSION

Based on the results cited above, it can be concluded that continuous application of saline water ( $8 \text{ dS m}^{-1}$ ) significantly increased soil Ece above the critical limit ( $4 \text{ dS m}^{-1}$ ). Application of K @  $120 \text{ kg ha}^{-1}$  significantly reduced the toxic effect of Na. The use of KCl fertilizer resulted in higher salinity than  $K_2SO_4$ . Application of a higher dose of K ( @  $120 \text{ kg ha}^{-1}$  ) significantly decreased the deleterious effect of Na by increasing the biomass yield, accumulation, and uptake of K. The Na accumulation and uptake decreased significantly.  $K^+/Na^+$  ratio in a plant is considered a good indicator to evaluate

plant response to K application in a saline stress environment.  $K^+/Na^+$  ratio in plant fairly correlated with biomass yield, accumulation, and uptake of K and Na by maize. Application of K through  $K_2SO_4$  was better than KCl in a saline environment. It decreased soil E<sub>Ce</sub> (6%). The biomass yield (3%), K uptake (3%), and  $K^+/Na^+$  ratio in a plant (2%) were higher in  $K_2SO_4$  than KCl.

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