

## **Original Research Article**

### **Green synthesis and characterization of Silver Nanoparticles synthesized using *Piper longum* and its antioxidant activity**

**Running title:** Synthesize of *Piper longum* based silver nanoparticle and to analyse its antioxidant activity.

**Abstract:** Introduction should be concised, add objective after introduction in abstract

Introduction: *Piper longum*, a traditional plant used for regenerative medicine commonly known as the long pepper, is used to treat many pathological conditions. *Piper longum* is used as a source for obtaining the production of various nanoparticles and testing their efficacy. *Piper longum* has many beneficial activities, such as antifungal, anti-amoebic, anti-asthmatic, anti-cancer, anti-oxidant, anti-inflammatory, antidepressant activities. The beneficial activities of *Piper longum* include antioxidant properties which have the ability to treat chronic conditions. The chemical nature of these constituents is the ability to produce antioxidant activity. Nanoparticles are minute particles that range from 1 to 100 nm on a nanometer scale. Different nanoparticles possess different properties; they vary in physical and chemical nature. Silver nanoparticles are widely and commonly used for the conduction of electricity, have a localized resonance effect on the surface plasmon, and have a wide range of antimicrobial activity against various microorganisms. The most important method used to synthesize silver nanoparticles involves the chemical method because of the nature of the silver nanoparticles which have the ease to simply dissolve in the prepared aqueous solution.

Material and methods: *Piper longum* plant extract was prepared and silver nanoparticles extract was synthesized. DPPH assay was then performed to assess the free radical scavenging activity of the *Piper longum* silver nanoparticle extract. The further analysis was that the formation of brownish-red color observed was evident, resulting in the synthesis of silver nanoparticles.

Results: Brown discoloration represents formation of silver nanoparticles and the antioxidant activity was noted to be from 60% to 70% from 10 to 50  $\mu$ L. But when compared to standard vitamin C values the *Piper longum* silver nanoparticles showed less antioxidant activity.

Conclusion: The *Piper longum* mediated silver nanoparticles showed remarkable and considerable antioxidant activity but when compared with the standard sample values the antioxidant activity was low.

Key words: Antioxidant, Innovative technique, eco-friendly, green synthesis, *Piper longum*

**Introduction:**

paragraphs should be more concised and don't exceed 6 rows, references at end of paragraph and grouped in numerical pattern

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*Piper longum*, a traditional plant used for regenerative medicine commonly known as the long pepper, is used to treat many pathological conditions<sup>1</sup>. Various substitutes that are derived from *Piper longum*, one of the most important derivatives, is piperine. It's used to treat diarrhea, chronic bronchitis, stomach pain, asthma, hepatitis, and other respiratory insufficiencies. Advanced technology implying the use of nanoparticles in biomedical research, is gaining its agenda, incorporating almost all natural materials as sources<sup>2</sup>. Drug delivery is the most common use of nanoparticles in the current technological application.

*Piper longum* is used as a source for obtaining the production of various nanoparticles and testing their efficacy. *Piper longum* possesses various pharmacologically beneficial activities, such as insecticide and acaricidal, anti-fungal, anti amoebic, anti-asthmatic, anti-cancer, anti-oxidant, anti-inflammatory, antidepressants, antiulcer activities on routine consumption<sup>3-5</sup>. The fruit of the *Piper longum* has hepatoprotective activity<sup>6</sup>. It is considered as an ayurvedic medicine providing enough treating ability to boost up the immune system to fight against various pathological disorders.

Piperlongumine, sesamin, and sylvatin are important phytoconstituents of *Piper longum*<sup>7</sup>. The chemical nature of these constituents is the ability to produce antioxidant activity. Nanoparticles are minute particles that range from 1 to 100 nm on a nanometer scale. Different nanoparticles possess different properties; they vary in physical and chemical nature. Nanotechnology is now widely used for synthesizing drugs in the pharmaceutical industry. Formulation of nanoparticles, nanospheres, nanocapsules, nanoemulsion, and nanosuspensions are few applications of nanotechnology in the field of nanomedicine. The most common synthesized nanoparticles are plasmonic silver and gold because they possess unique physical and chemical properties when compared to other large surface areas and have strong electronic properties<sup>8</sup>.

Nanoparticles are physically and chemically altered to have high potency to act upon the biological cell structure of the body<sup>9,10</sup>. Silver nanoparticles are widely and commonly used for the conduction of electricity<sup>11</sup>. They also have a localized resonance effect on the surface plasmon and have a wide range of antimicrobial activity against various microorganisms<sup>12</sup>. The synthesis of silver nanoparticles involves 3 complex methods, they include physical synthesis, chemical synthesis, and biological synthesis<sup>13</sup>. The most important method used to synthesize silver nanoparticles involves the chemical method because of the nature of the silver nanoparticles which have the ease to simply dissolve in the prepared aqueous solution<sup>14</sup>. Silver nanoparticles are commonly used in nanotechnology<sup>15</sup>. Antioxidant activity is generally stated as prevention of oxidation of various constituents of the body, thus substances that tend to possess antioxidant activity which will block the oxidation of protein, DNA, and lipids present in our body.

An antioxidant prevents the formation of free radicals and their storage in tissues thus helps to prevent tissue depletion<sup>16,17</sup>. Various naturally occurring plant sources and their products have antioxidant properties, which depresses and overcomes the oxidative stress caused by free radicals<sup>18</sup>. The beneficial activities of *Piper longum* include antioxidant properties which have the ability to treat chronic conditions<sup>16</sup>. Our team has extensive knowledge and research experience that has translated into high quality publications<sup>1920-3334-38, 34-38</sup> (numerical arrangement). Thus

the aim of this study is to green synthesize and characterize the silver nanoparticles using *Piper longum* plant extract and to observe its antioxidant capacity.

## **Materials and methods:**

### **Preparation of plant extract:**

Commercially available dry powder of *Piper longum* was used for this experiment. This experiment was an *in vitro* study which was conducted in Saveetha Dental College, Chennai, Tamilnadu. The experiment was carried out by dissolving 1g of *Piper longum* powder in 100ml of water. The mixture was then boiled in a heating mantle at 70 degrees celsius for up to 10 minutes. The boiled mixture was then filtered using a Whatman number 1 filter paper to obtain the plant extract. Then 40 ml of plant extract was measured using a measuring cylinder and the mixture was added to 60ml of 1 mM silver nitrate (0.0169g) dissolved in 60ml distilled water. The lab technician reconfirmed the synthesized *Piper longum* silver nanoparticles to avoid sampling error for further analysis. The further analysis was that the formation of brownish-red color observed was evident, resulting in the synthesis of silver nanoparticles.

### **Reagent and Chemicals :**

1g of *Piper longum* powder

1 mM silver nitrate (0.0169g)

2-10 µg/ml concentration of *Piper longum* extract at various concentrations (10 µL, 20 µL, 30 µL, 40 µL, and 50 µL)

0.1 mM DPPH in methanol

50 mM of tris HCl buffer

Control: BHT: Butylated hydroxytoluene

### **Antioxidant activity:**

*Piper longum* was the plant used to determine the antioxidant activity by taking only 5 samples in distinct test tubes and adding 10 µL, 20 µL, 30 µL, 40 µL, and 50 µL of the solution respectively using a micropipette. The micropipette usage was dealt with care to avoid methodology error. Random sampling method was performed to eliminate sampling bias and the validation of the procedure was done by principal investigator and by experts in nanotechnology. Other plants and activities were not tested in this present study.

To check the antioxidant activity, a DPPH assay was used to test the antioxidant activity of biogenic synthesized silver nitrate nanoparticles. Diverse concentrations (2-10 µg/ml) of

*Piper longum* leaf extract interceded silver nitrate nanoparticle was mixed with 1 ml of 0.1 mM DPPH in methanol and 450 µl of 50 mM of tris HCl buffer (pH 7.4) and incubated for 30 minutes. Later, the reduction in the quantity of DPPH free radicals was assessed dependent on the absorbance at 517 nm. BHT was employed as control. The percentage of inhibition was determined from the following equation:

$$\% \text{ Inhibition} = \frac{\text{Absorbance of control} - \text{Absorbance of sample}}{\text{Absorbance of control}} \times 100$$

The percentage of Inhibition was then calculated and correlated with the standard vitamin C concentration values which were not performed. And the antioxidant activity was correlated using Spearman's correlation in SPSS software version 23.0, the correlation graph was then represented and the p value was less than 0.05 for the non parametric correlation analysis proving statistical significance.

**Results:** correct yellow coloured words as shown, paragraphs more concised, figures and tables at end of paragraph, figure 5 mentioned below wasn't mentioned in introduction of results. The optical observation of the experiment was determined using a UV-visible spectrum providing the characterization of the prepared silver particles. The formation and synthesis of silver nanoparticles were determined by the brownish-yellow discoloration shown (Figure 1). The corresponding formation of silver nanoparticles is (was) shown in peaks at 517 nm in of the UV spectrophotometer graph (Figure 4). The presence of the surface plasmon resonance was the reason for the color change of the solution. Silver nanoparticles are (were) formed by the reaction between the plant extract and the silver ions. Only potential plants are (were) able to synthesize nanoparticles thus the change in color of the solution acts (acted) as a primary identification tool. The DPPH assay performed to test the antioxidant activity of the synthesized *Piper longum* silver nanoparticles is (was) shown (Figure 2).

The percentage of inhibition of the synthesized *Piper longum* silver nanoparticles was noted to be 61.2%, 64.5%, 66.1%, 67.4% and 69.3% at 10 µL, 20 µL, 30 µL, 40 µL and 50 µL respectively and where as the values of the standard vitamin C solution are 76.5%, 78.8%, 84.9%, 89.2% and 92.3% at 10 µL, 20 µL, 30 µL, 40 µL and 50 µL Respectively (Table 1). The correlation analysis performed using SPSS version 23.0 and is (was) shown. There is (was) a positive correlation between the concentration of the sample prepared and percentage of inhibition. Thus the antioxidant activity is (was) increasing with increased concentration of the synthesized silver nanoparticles (Figure 3).

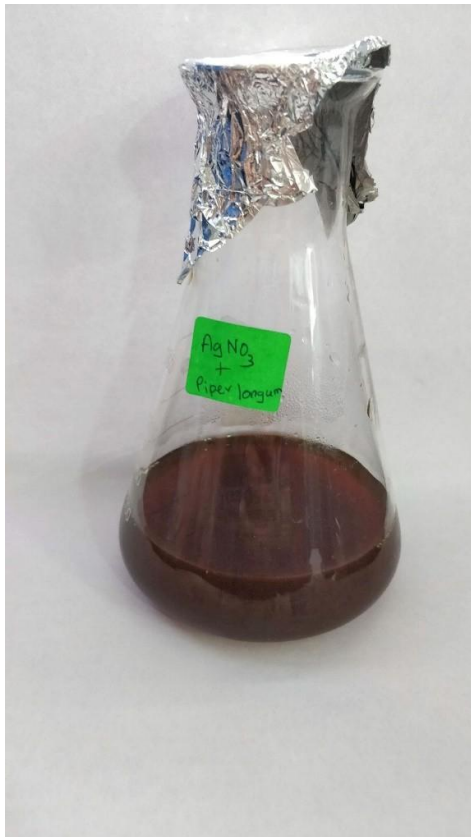


Figure 1: The colour change indicates the synthesis of silver nanoparticles after 72 hrs and its corresponding correlation of the UV spectroscopy (figure 4).

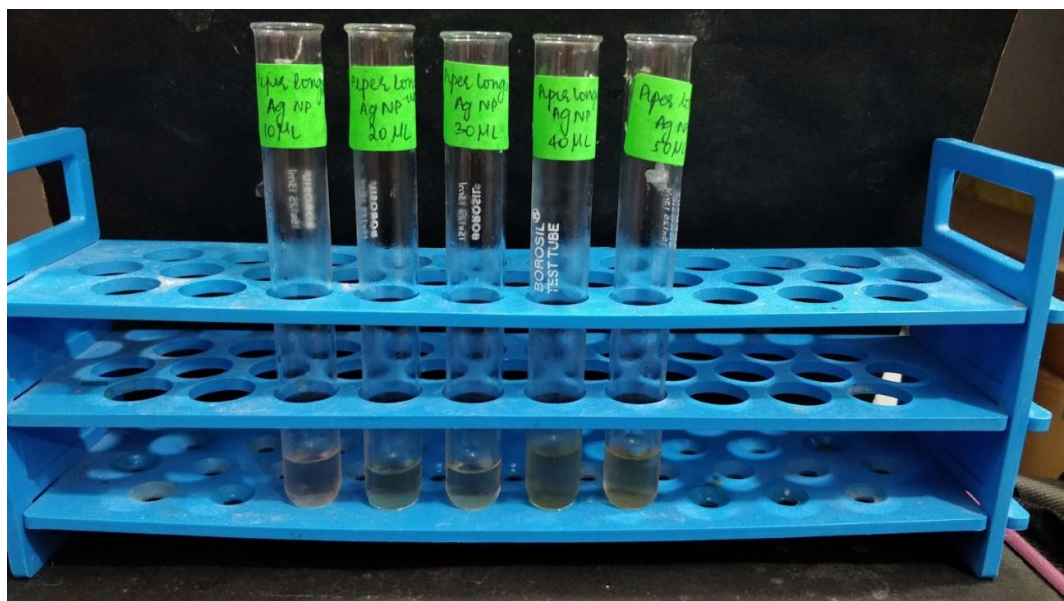


Figure 2: Antioxidant activity of the silver nanoparticles using the plant extract and the colour change is noted.

Percentage of Inhibition (%)		
Concentration ( $\mu\text{L}$ )	Silver nitrate nanoparticles	Standard
10 $\mu\text{L}$	61.2%	76.5%
20 $\mu\text{L}$	64.5%	78.8%
30 $\mu\text{L}$	66.1%	84.9%
40 $\mu\text{L}$	67.4%	89.2%
50 $\mu\text{L}$	69.3%	92.3%

Table 1: The percentage of Inhibition for the different concentrations of the silver nanoparticles (ranging from 10  $\mu\text{L}$  to 50  $\mu\text{L}$ ) in comparison with the standard.

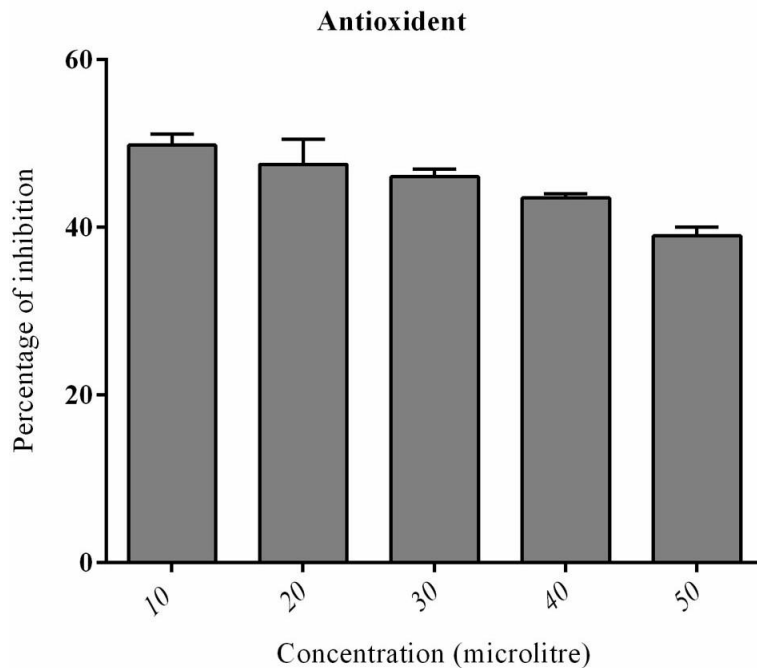


Figure 3: Antioxidant activity of silver nanoparticles. The X-axis represents the concentration of the silver nanoparticles and standard solution and the Y-axis represents the percentage of inhibition. Graph representing the percentage of inhibition observed with the different concentrations ranging from 10  $\mu\text{L}$  to 50  $\mu\text{L}$  in comparison to the standard value, the correlation was found to be positive using spearman's correlation. And the non-parametric correlation was found with  $P$  value less than .05 proving statistically significance.

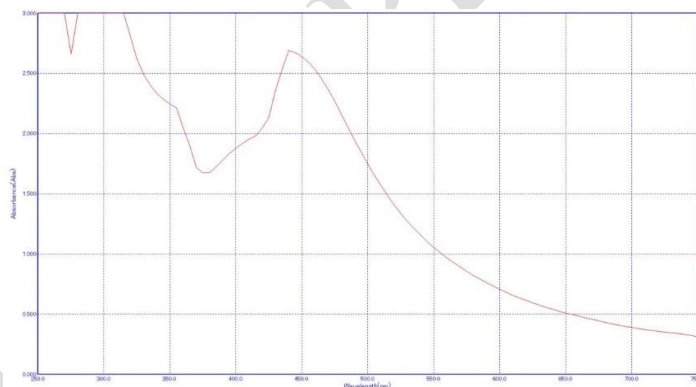


Figure 4: The UV/Visible spectrum of the silver nanoparticles synthesized using the *Piper longum* plant leaf extract.

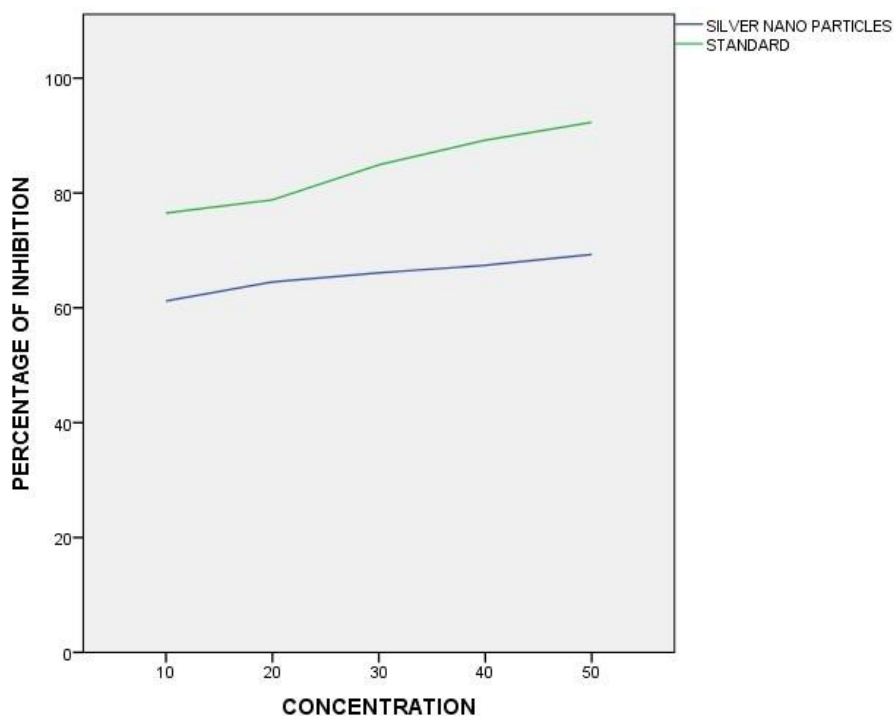


Figure 5: The line graph represents the comparative observation of the antioxidant activity of *Piper longum* AgNPs (blue) and the standard solution antioxidant activity (green). The X-axis represents the concentration of the silver nanoparticles and standard solution and the Y-axis represents the percentage of inhibition. The correlation was found to be positive using Spearman's correlation. And the non-parametric correlation was found with  $P$  value less than .05 proving statistically significance.

Discussion: correct yellow coloured verbs in past simple as shown, references at end of paragraph, paragraphs no more than 6 rows

The present study investigates the optical observation which is the initial and novel technique that reveals the characterization of the synthesized silver nanoparticles as it presents a result of the UV- visible absorption spectra . Figure 1 shows the brownish-yellow discoloration indicative of the formation of the silver nanoparticles. The silver nanoparticle formation showed significant color change because of the occurrence of the surface plasmon resonance<sup>39-40</sup> . As the color change serves as a primary tool, this is useful in the detection of the potential plants for the synthesis of silver nanoparticles.

The colour change in the prepared solution is correlated with its UV spectrophotometer graph in Figure 4, in which the peak of the synthesized *Piper longum* solution with silver nanoparticles production is 517 nm. Similar studies show that the prepared nanoparticles accompanied with the respective natural plant components used in assessing its antioxidant activity was comparative<sup>41-42</sup> . The synthesise and production of the chosen nanoparticles are believed to have formed at the range of 400-520 no in an UVvisible spectrophotometer graph<sup>43</sup> . So compatible values of the UV spectrophotometer graph reveals the silver nanoparticle

produced in 48 hrs.

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2,2-diphenyl-1-picrylhydrazyl is an organic chemical compound which stands as the full form of DPPH, it is an assay performed to determine the antioxidant that is the free radical scavenging activity of a substance<sup>44</sup>. So the DPPH assay was performed to observe the antioxidant activity of the synthesized silver nanoparticles in this present study. **Figure 2 shows** the antioxidant activity of the newly synthesized silver nanoparticles using the plant extract of *Piper longum*. **Figure 3 represents** the bar graph which **represents** the comparison between the percentage of inhibition between the silver nanoparticles synthesized and the standard solution. **Figure 5 represents** the line graph version of the already displayed bar graph of the results in **table1**.

In comparison to various works conducted by different authors, the study conducted by S.N Kharat<sup>45</sup> the DPPH assay scavenging ability was in a purely dose-dependent manner, the scavenging property at 50  $\mu\text{L}$  was 15.23% which was the lowest concentration whereas when compared to the present study the antioxidant activity was 61.2% at the lowest concentration that is at 10  $\mu\text{L}$  shown in **Figure 3** which represents a correlation analysis. The work conducted by Rajeshkumar<sup>39</sup> also concluded that the free radical scavenging activity was concentration dependent as well. Keshari<sup>46</sup> concluded in study that AgNPs **have** the scavenging property of more than 29% stating its high free radical scavenging activity. Comparing works of DPPH radical scavenging with the work done by Zdenka<sup>47</sup> **states** that the activity **is** more than 89% at the highest concentration being 50 $\mu\text{L}$ .

Opposing work conducted by **name of scientist**<sup>48</sup> without the involvement of nanoparticles showed that there **is** potential antioxidant activity exhibited by the synthesized *Piper longum* extract. The *Piper* species **are** accountable and known for their reduction of oxidative stress which **makes** them a good source of an antioxidant substance. They **are** successfully used to treat Alzheimer's disease and Parkinson's disease which are common disorders affecting the central nervous system<sup>49</sup>.

Generally the silver nanoparticles **are** remarkable for its significant antioxidant activity<sup>50</sup>, they also **show** numerous beneficial activities such as antimicrobial<sup>51,52</sup> and anti-inflammatory<sup>53</sup>, so silver nanoparticles **are** of the most commonly used metal nanoparticles in nanomedicine<sup>54</sup>. So silver nanoparticles **play** a crucial role in nanomedicine involving nano technological advances and also for biomedicine using genetic engineering<sup>55</sup>. The use of silver nanoparticles made the present study economical, less cost effective, and easily available<sup>56</sup>. But not so on the bright side, recent researchers **have** found that silver nanoparticles consumption **has** caused some adverse effects on human metabolism. The toxic adverse effects **include** suppressing activity of the cell growth, division and multiplication<sup>57</sup>, which was found to be in a pure concentration consumption and exposure manner.

The present study was based on the silver nanoparticles having the property of possessing equally accessible antioxidant activity when compared to the standard scavenging property of the vitamin C solution<sup>58</sup>. Vitamin C was chosen for this present study from the other two free radical scavenging vitamins that are A and E because of the slightly higher values of the antioxidant activities shown by them on various biomedical facilities<sup>59</sup>. Vitamin C free radical scavenging property is due to its capacity to donate an electron which **provides** the basic properties of a vitamin which **is** potent enough to show strong antioxidant activity. An antioxidant thus **works** by preventing the oxidation of the cells and tissues present in the human body and thereby neutralizing the free radicals produced<sup>60</sup>. Substances which **possess** strong antioxidant activity **have** a great potential to treat and cure many high risk human

metabolic problems like atherosclerosis and cardiovascular disorders <sup>61</sup>. In vitro studies have clearly ruled out the antioxidant activity of vitamin c in biological systems <sup>62</sup>.

The present study deals with the comparison of the prepared *Piper longum* plant extract nanoparticles with standard Vitamin C values which shows significant values of free radical scavenging property. But the plant extract nanoparticle prepared had less antioxidant activity than the standard values. Table 1 shows clear and accurate values of the antioxidant activity of the two solutions used to check its antioxidant activity.

Limitations of the present study were less sample size taken for the study, different plants could be taken as they all show different antioxidant capacity and other beneficial properties. Only antioxidant activity of the *Piper longum* plant was assessed, and the specification in using silver nanoparticles may also be limitations of the study. Value changes may be formed due to errors in experiment performance.

According to the present study conducted, the free radical scavenging property of the prepared *Piper longum* plant extract silver nanoparticles was from an average range of 60% to 70% of the antioxidant activity. The maximum antioxidant activity was shown at the highest concentration that is at 50 µL it showed nearly 70% of inhibition. Thus the prepared *Piper longum* silver nanoparticles have good antioxidant activity and can be potentially used in nanomedicine to treat various biological conditions. The future scope of this study is the properties of synthesizing nanoparticles which are further implied in nanotechnology which may be useful for nanomedicine, biomedicine and other sourcing fields. Furthermore research works are needed to potentiate the various aspects of the use of nanoparticles in different fields.

Conclusion: correct yellow coloured verbs in past simple as shown,

The *Piper longum* mediated silver nanoparticles show remarkable and considerable antioxidant activity when compared with the standard values of vitamin C. The potential free radical scavenging property of the plant leaves extract mediated silver nanoparticles, is applicable in nanomedicine through the help of advanced technologies to support the medical voided fields. The properties of the synthesized nanoparticles are further implied in nanotechnology which may be useful for other sourcing fields. Further research studies are required to assess the full potential of the plant mediated nanoparticles.

**ACKNOWLEDGEMENTS if present**

**COMPETING INTERESTS**

**AUTHORS' CONTRIBUTIONS if present**

**ETHICAL APPROVAL (when applicable)**

**References: et al is added after 3 references as shown below**

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