

Original Research Article

Melissa officinalis L. Essential Oil: Chemical composition, Antioxidant, Antibacterial and Antifungal Activity- In Vitro Study

ABSTRACT

Melissa officinalis L. plant is widely used in traditional medicines due to its health and nutritional benefits. The present investigated chemical composition of *Melissa officinalis* L. essential oil (MOEO) extracted by hydrodistillation. The MOEO was analyzed by gas chromatography-mass spectrometry (GC-MS), revealing the presence of thirty compounds, representing 98.46% of the oil constituents. The predominant components were 1,8-cineole (39.80%) followed by citronellol (16.66%), geraniol (12.25%), myrcene (5.85%) and geranial (5.45%). The antioxidant potential of MOEO has been summarized using DPPH test (IC₅₀), superoxide anion (O₂⁻) scavenging activity (IC₅₀), β-carotene (IC₅₀) and reducing power (FRAP) (EC₅₀). Results demonstrate strong scavenging superoxide anion capacity and moderate to weaker activity against the other assays. Potent inhibitory effect has been observed towards *Micrococcus luteus* and *Bacillus cereus* as well as the *Candida albicans* ATCC 90028, *Candida tropicalis* 1 and *Candida albicans* 8. Our work provides a view for the further studies on the antioxidant and antimicrobial of the MOEO and their main components.

Keywords: *Melissa officinalis*; essential oil; GS-MS; antioxidant; antibacterial; antifungal.

1. INTRODUCTION

Essential oils are a complex matrix composed of various volatile compounds which have been recognized for a long time as a powerful reservoir of therapeutic and pharmacological effect [1-3]. They have been explored as new alternatives in the development of cosmetics and pharmaceuticals products mainly due to their antioxidant and antimicrobial properties. Those properties are mainly associated with their own chemical composition, which is determined by pedoclimatic conditions and plant genotype [4,5]. The high and spread invasive of bacterial and fungal infections, in parallel with the increasing of drug resistance accelerate the need for new antioxidants and antimicrobials to overcome infectious disease [6-8]. Those from natural products remain the most requested due to their low side effects [9-14].

Melissa officinalis L. (*M. officinalis*) belonging to Lamiaceae family (mint family) is a perennial, subshrub, endemic, herbaceous medicinal plant native to Southern Europe and the Mediterranean regions as well as in Central Asia, Serbia, America, and Africa [15-17]. The germination of the plant accrued naturally in sandy and scrubby areas with a matured stage had an average height in the range 70–150 cm [18,19]. Traditionally this plant has been explored for its large therapeutic effect to cure various diseases, including cardiac,

headaches, carminative flatulence, antispasmodic in digestion, colic, nausea, hypersensitivities, amenorrhea, nervousness, anemia, vertigo, syncope, malaise, asthma, rheumatismbronchitis, failure, arrhythmias, insomnia, epilepsy, depression, psychosis, hysteria, ulcers and wounds [20,21]. Previously studies suggested that the high pharmacological capacity of this plant is mainly related to its richness in secondary metabolites such as flavonoids, phenolic acid, and terpenes [22]. Other secondary metabolites such as eugenol, octinol, octin, octinone, citral, hexenol, and haramin from *M. officinalis* essential oil contribute to its powerful activity [23]. This essential oil of this plant has long been used due to its antioxidant, antifungal and antimicrobial properties [24-26]. The *M. officinalis* essential oil has been demonstrated for its high effect against four bacterial nosocomial infections: *Pseudomonas aeruginosa*, *Klebsiella pneumonia*, *Staphylococcus aureus* and *Citrobacter koseri* [16]. Also, Abdellatif et al. [23] reported the high antimicrobial activity against microorganisms, mainly five human pathogenic bacteria, one yeast, *Candida albicans*, and two phytopathogenic fungi. Moreover, the essential oil obtained by microdistillation of aerial parts of *M. officinalis* (Bulgarian origin) showed potent antioxidant activity. In another study, the low antioxidant properties of EO of *M. officinalis* was mainly explained by attributed to the low contents of volatile phenolic compounds such as camphor and carvacrol [23].

The aim of the study was to evaluate the composition, antioxidant, antibacterial and antifungal properties as well as assessment of the enhancement of the skin penetration, of *M. officinalis* essential oil.

2. MATERIAL AND METHODS

2.1. Plant material and essential oil isolation

Melissa officinalis L. plants are harvested from a private nursery for aromatic and medicinal plants in the region of Sidi Bouzid (center of Tunisia) and identified according to the flora of Tunisia. Aerial parts were separated and dried at room temperature for about ten days. Once dried, the plant material was extracted for EOs. An amount of 200 g of aerial part was transferred to hydro-distillation for 3 hours with 1000 mL distilled water using a Clevenger-type apparatus. The distilled EO was dried over anhydrous sodium sulfate, filtered, and stored at 4°C. The yield was calculated based on the dried weight of the sample.

2.2. Gas chromatography–mass spectrometry analyses of the essential oil

2.2.1. Gas chromatography analysis.

Gas chromatograph: HP 5890-series II equipped with flame ionization detector (FID), HP-5 (30 m × 0.25 mm i.d., 0.25 µm film thickness) and the HP-Innowax column (polyethylene glycol column, 0.25 mm internal diameter, 30 m length and 0.25 µm film in thickness) fused silica capillary column, carrier gas nitrogen (1.2 ml per min). The oven temperature was programmed from 50°C (1 min) to 280°C at 5°C per minute. The injector and the detector temperatures were 250 and 280°C, respectively. Volume injected: 0.1 µl of 1% hexane solution. The identification of the components was performed by comparison of their retention times with those of pure authentic samples and by mean of their retention indices relative to the series of *n*-hydrocarbons.

2.2.2. Gas chromatography-mass spectrometry analysis.

GC/EIMS analyses were performed with the Varian CP-3800 gas-chromatograph equipped with the HP-5 capillary column (30 m × 0.25 mm; coating thickness 0.25 µm) and the Varian Saturn 2000 ions trap mass detector. Analytical conditions: injector and transfer line temperatures 220 and 240°C respectively; oven temperature programmed from 60°C to 240°C at 3°C per min; carrier gas helium at 1 ml/min; injection of 0.2 µl (10% hexane

solution); split ratio 1:30. The identification of the constituents was based on comparison to retention times with those of authentic samples, comparing their linear retention indices relative to the series of *n*-hydrocarbons, on computer matching against commercial and internal library mass spectra built up from pure substances, components of known oils and MS literature data [27]. Moreover, the molecular weights of all the identified substances were confirmed by GC/CIMS, using methanol as *C*/ionizing.

2.3. Antibacterial activity

The bacterial strains tested in this study belonged to 10 references, which are presented in tables 2. The bacterial species consisted of 6 Gram-positive and 4 Gram-negative bacterial strains. The disc-diffusion assay was performed according to the protocol described by Vuddhakul et al. [28] and slightly modified by Hajlaoui et al [29]. For the experiments, a loopful of the microorganisms working stocks were enriched on a tube containing 9 mL of Mueller-Hinton broth then incubated at 37°C for 18 to 24 h. The overnight cultures were used for the antibacterial activity of the EO used in this study, and optical density was adjusted at 0.5 McFarland turbidity standards with a DENSIMAT (Biomérieux®). The inoculums of the bacteria were streaked onto MH plates using a sterile swab. Sterile filter discs (diameter 6 mm, Whatman paper No. 3) were impregnated with 10 µL of EO placed on the appropriate agar media. Gentamycin (10 µg/disc) was used as positive reference standards to determine the sensitivity of one strain/isolate to each of the tested microbial species. After incubation at 37°C for 18 to 24 h, the diameter of inhibition zone was measured with 1 mm flat rule, and diameters were interpreted according to the Committee of the French Society of the Antibiogram [28]. The dishes were incubated at 37°C for 18-24 h for microbial strains. The diameter of inhibition zones around each of the discs was taken as measure of antimicrobial activity. Each experiment was carried out in triplicate, and the mean diameter of the inhibition zone was recorded.

2.4. Screening for antifungal activity

The human pathogenic yeast used in this study was isolated from patients suffering from candidosis in hospital (Parasitology Department-CHU Farhat Hached and Monastir Dental Clinic, Tunisia). These strains were isolated on Sabouraud chloramphenicol agar plates and identified with Api ID 32 C strips (bioMérieux, Marcy l'Etoile, France) according to the manufacturer's recommendations. The strains tested belong to 4 references strains and 26 isolates: *Candida krusei* (n = 1), *Candida tropicalis* (n = 2), *Candida glabrata* (n = 4), *Candida parapsilosis* (n = 1), *Candida sake* (n = 2), *Candida kefyr* (n = 3), *Candida holmii* (n=2) and *Candida albicans* (n = 11). For screening the antifungal activity of *Melissa officinalis* essential oil, the agar-disc diffusion method was described by [1]. All *Candida* strains were first grown on Sabouraud chloramphenicol agar plate at 30°C for 18–24 h. Several colonies were transferred into Api suspension medium (bioMérieux) and adjusted to 2 McFarland turbidity standard with a densimat (bio- Mérieux). The inocula of the respective yeast was streaked on to Sabouraud chloramphenicol agar plates at 30°C using a sterile swab and then dried. A sterile filter disc, diameter 6 mm (Whatman paper no. 3) was placed in the plate. Ten microlitres of the essential oil was dropped on each paper disc (10 mg disc). The treated Petri dishes were placed at 4°C for 1–2 h and then incubated at 37°C for 18–24 h. The antifungal activity was evaluated by measuring the diameter of the growth inhibition zone around the discs. The susceptibility of the standard drug fluconazole was determined using a disc paper containing 20 µg/disc. Each experiment was carried out in triplicate and the mean diameter of the inhibition zone was recorded.

2.5. Antioxidant activity

2.5.1. Scavenging Ability on DPPH Radical

The DPPH[•] quenching ability of the EO was measured according to the method cited by [30]. 1 ml of the oil, extracted at known concentrations, was added to 0.25 ml of a DPPH[•] methanolic solution. The mixture was shaken vigorously and left to rest at room temperature for 30 min in the dark. The absorbance of the resulting solution was then measured at 517 nm to determine the values corresponding to the ability of the extracts to reduce the stable radical DPPH to the yellow-coloured diphenylpicrylhydrazine. Antiradical activity was expressed as IC₅₀ (µg/mL) values, reflecting the extract doses required to cause a 50% inhibition. A lower IC₅₀ value corresponded to a higher antioxidant activity of plant extract. The ability to scavenge the DPPH radical was calculated using the following equation:

$$\text{DPPH}^{\bullet} \text{ scavenging effect (\%)} = [(A_0 - A_1) / A_0] * 100 \quad (1)$$

Where A₀ refers to the absorbance of the control at 30 min, and A₁ to the absorbance of the sample at 30 min. All samples were analyzed in triplicate.

2.5.2. Superoxide Anion Radical-Scavenging Activity

Superoxide anion scavenging activity was assessed using the method cited by [1] with slight modification. The reaction mixture contained 0.2 mL of EO assayed at different concentrations, 0.2 mL of 60 mM PMS stock solution, 0.2 mL of 677 mM NADH, and 0.2 mL of 144 mM NBT, all in phosphate buffer (0.1 mol/L, pH 7.4). After incubation at ambient temperature for 5 min, absorbance was read at 560 nm against a blank. Antioxidant activity was evaluated based on IC₅₀ values, which was defined as the amount of antioxidant needed to reduce the generation of superoxide radical anions by 50% and expressed as µg/mL (as determined from three replicates per treatment). The inhibition percentage of superoxide anion generation was calculated using the following formula:

$$\text{Superoxide quenching (\%)} = [(A_0 - A_1) \times 100] / A_0$$

Where, A₀ and A₁ had the same references presented in Eq. (1).

2.5.3. Reducing Power

The ability of the EO to reduce Fe³⁺ was assayed using the method described by [31]. Briefly, 1 mL of the EO was mixed with 2.5 mL of phosphate buffer (0.2M, pH 6.6) and 2.5 mL of 1% K₃Fe (CN)₆. Absorbance was measured at 700 nm. The mean of absorbance values was plotted against concentration values, and a linear regression analysis was performed. Increased absorbance of the reaction mixture indicated increased reducing power. The EC₅₀ value (µg/mL) is the effective concentration at which absorbance was 0.5 for reducing power. BHT and ascorbic acid were used as positive control.

2.4.4. β-Carotene-Linoleic Acid Model System (β-CLAMS)

The β-CLAMS method is based on the discoloration of β-carotene by the peroxides generated during the oxidation of linoleic acid at elevated temperature and was performed based on the protocol done by [32]. In this study the β-CLAMS was modified for the 96-well micro-plate reader. In brief, the β-carotene was dissolved in 2 mL of CHCl₃, to which 20 mg of linoleic acid and 200 mg of tween 40 were added. The results are expressed as IC₅₀ values (µg/mL). All samples were prepared and analyzed in triplicate.

3. RESULTS AND DISCUSSION

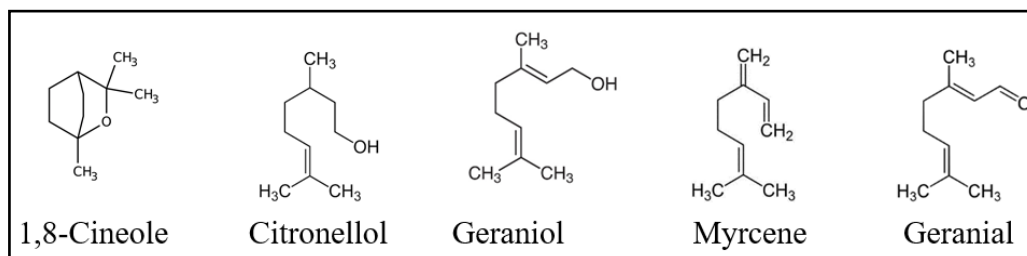
3.1. Essential oil composition of MOEO

The isolated MOEO was a yellow color with the yield of 2.75g/100 g dry weight. The GC-MS analyses led to the identification of 30 different components, representing 98.46 % of the total oil (Table 1).

Table 1. Chemical composition, Kovats indices and percentage composition of *Melissa officinalis* L. essential oil.

N°	Compounds	Kovats index (KI) HP-5	%	Identification
1	α -Thujene	929	0.17	MS, KI
2	α -Pinene	936	1.02	MS, KI
3	Camphene	951	0.06	MS, KI
4	Myrcene	980	5.85	MS, KI
5	Sabinene	992	0.74	MS, KI
6	α -Terpinene	1013	0.05	MS, KI
7	1,8-Cineole	1036	39.80	MS, KI
8	(<i>E</i>)- β -Ocimene	1054	0.34	MS, KI
9	γ -Terpinene	1061	2.87	MS, KI
10	Linalool	1090	0.19	MS, KI
11	<i>cis</i> -Sabinene hydrate	1100	1.06	MS, KI
12	β -Thujone	1129	0.09	MS, KI
14	Camphor	1158	2.91	MS, KI
15	<i>trans</i> -Pinocarveol	1173	0.17	MS, KI
16	Borneol	1180	0.17	MS, KI
17	Terpinen-4-ol	1193	0.21	MS, KI
18	α -Terpineol	1204	0.09	MS, KI
19	<i>trans</i> -Piperitol	1216	0.11	MS, KI
20	Citronellol	1232	16.66	MS, KI
21	Neral	1243	0.68	MS, KI
22	Geraniol	1261	12.25	MS, KI
23	Geranial	1274	5.45	MS, KI
24	Carvacrol	1302	0.09	MS, KI
25	Citronellyl acetate	1353	1.72	MS, KI
26	Eugenol	1361	0.77	MS, KI
27	Geranyl acetate	1384	2.35	MS, KI
28	β-elemene	1397	2.11	MS, KI
29	β -caryophyllene	1426	0.24	MS, KI
30	<i>trans</i> α -bergamotene	1438	0.27	MS, KI
Yield (g/100 g dry weight)			2.75	
Monoterpenes Hydrocarbons			12.16	
Oxygenated monoterpenes			83.71	
Sesquiterpene hydrocarbons			2.62	
Total			98.46	

The main components were 1,8-cineole (39.80%) followed by citronellol (16.66%), geraniol (12.25%), The second components in abundance (1.72-5.85%) were myrcene (5.85%), geranial (5.45%), camphor (2.91), γ -terpinene (2.87%), geranyl acetate (2.35%), β -elemene (2.11%) and citronellyl acetate (1.72%). The major classes of the MOEO were aliphatic



esters (61.02%), oxygenated monoterpenes (83.71%), monoterpene hydrocarbons (12.16%) and sesquiterpene hydrocarbons (2.62%). This composition showed that this MOEO has a richness of compounds known by their important biological activities (Fig. 1)

Fig. 1. Structure of the major components of MOEO

3.2. Antioxidant activity of MOEO

In this study, four assays have been employed to investigate the antioxidant power this EO, and the results were compared to the commercial standard (BHT) ($\mu\text{g/ml}$). The results depicted in Figure 2 shows that MOEO exhibited strongly antioxidant activity towards superoxide anion with $\text{IC}_{50} = 1.5 \pm 0.2 \mu\text{g/mL}$, which is 9.66 times higher than the standard, BHT. Also moderate to weaker antioxidant power has been observed against the other three tests with $\text{IC}_{50} = 11.50 \pm 0.62 \mu\text{g/mL}$ (DPPH), $\text{IC}_{50} = 75 \pm 1 \mu\text{g/mL}$ (β -carotene) and $\text{EC}_{50} = 23 \pm 1 \mu\text{g/mL}$ (FRAP), respectively.

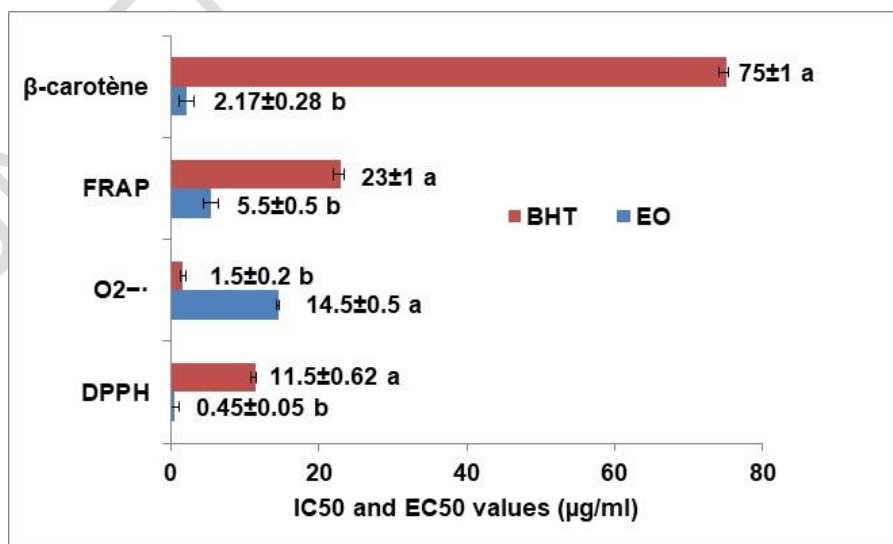


Fig. 2. DPPH test (IC_{50}), Superoxide anion ($O_2^{\cdot-}$) scavenging activity (IC_{50}), reducing power (FRAP) (EC_{50}), and β -carotene of *Melissa officinalis* essential oil (EO) compared to authentic standard (BHT) ($\mu g/ml$). Means (three replicates) followed by at least one same letter within a row are not significantly different at $P < 0.05$.

3.3. Antibacterial activity of MOEO

The *in vitro* antibacterial activity MOEO against ten bacterial strains including six Gram positive bacteria and four Gram negative bacteria, has been assessed in comparison with gentamycin commercial drug, and the data are presented in Table 2. As shown, MOEO affected all microbial strains within inhibition zones (IZ) ranged between 8.00 mm and 23.67 mm and that *Micrococcus luteus* (IZ = 21.33 ± 1.15 mm vs. 27.67 ± 1.5 mm), and the two different strains *Bacillus cereus* ATCC 11778 (IZ = 20.67 ± 0.58 mm vs. 26 ± 1 mm) and *Bacillus cereus* ATCC 14579 (23.67 ± 0.58 vs. 28 ± 0) are the most sensitive pathogens. Besides that, our results indicate that Gram positive bacteria were less resistant to the MOEO than Gram negative bacteria, due to the difficulty of hydrophobic molecules to enter on the cell wall having an outer layer surrounding it, which allowed them to be less able to affect their growth.

Table 2. Zones of growth inhibition (IZ mm \pm SD), showing the qualitative antibacterial activity of *Melissa officinalis* essential oil (EO) against human pathogenic bacteria compared to standard antibiotic (Gentamycin).

Strains	IZ (EO)	IZ (Gentamycin)
Gram positive bacteria		
<i>Staphylococcus epidermidis</i> CIP 106510	12 \pm 1 ^{dB}	21.33 \pm 0.58 ^{fghA}
<i>Staphylococcus aureus</i> ATCC 25923	13.33 \pm 0.58 ^{CB}	32.67 \pm 0.58 ^{AA}
<i>Micrococcus luteus</i> NCIMB 8166	21.33 \pm 1.15 ^{bB}	27.67 \pm 1.53 ^{bA}
<i>Enterococcus faecalis</i> ATCC 29212	14 \pm 1 ^{CB}	26 \pm 1 ^{CA}
<i>Bacillus cereus</i> ATCC 11778	20.67 \pm 0.58 ^{bB}	26 \pm 1 ^{CA}
<i>Bacillus cereus</i> ATCC 14579	23.67 \pm 0.58 ^{aB}	28 \pm 0 ^{bA}
Gram negative bacteria		
<i>Escherichia coli</i> ATCC 35218	10.67 ^{eB}	22 \pm 1 ^{etgA}
<i>Listeria monocytogenes</i> ATCC19115	13 \pm 0 ^{cdB}	23 \pm 0 ^{deA}
<i>Pseudomonas aeruginosa</i> ATCC 27853	8 \pm 0 ^{fB}	17 \pm 1 ^{IA}
<i>Salmonella typhimurium</i> LT2 DT104	8 \pm 0 ^{fB}	20.33 \pm 0.57 ^{hA}

SD: Standard deviation; IZ: Inhibition zone diameter (mm) around the discs (6mm) impregnated with 10 μl of essential oil and 10 $\mu g/disc$ for Gentamycin (Gent). a,b,c,d,e, f, g, h,i, A,B: Each value represents the average of 3 repetitions. Means followed by the same letters are not significantly different at $P=0.05$ based on Duncan's multiple range test. Small letters are used to compare IZ EO and IZ Gentamycin means between different strains, while capital letters are used to compare means between IZ EO and IZ Gentamycin for the same strain.

3.4. Antifungal activity of MOEO

The results of antifungal activity compared to the standard antibiotic, fuconazole, revealed that MOEO inhibited the growth of all tested fungi with the most remarkable inhibitory effect was obtained towards *Candida albicans* ATCC 90028 (IZ = 27.33 ± 0.58 mm vs. 18 ± 0 mm), *Candida tropicalis* 1 (IZ = 15.66 ± 0.57 mm vs. 16 ± 0 mm) and *Candida albicans* 8 (IZ = 16 ± 1.73 mm vs. 16 ± 0 mm) (Table 3).

Table 3. Antifungal activity of *Melissa officinalis* essential oil against several *Candida* species.

Strains	Origin	Type of samples	IZ (EO)	IZ (Fluconazole)
<i>Candida albicans</i>	ATCC 90028	-	27.33±0.58 ^a	18±0 ^b
<i>Candida glabrata</i>	ATCC 90030	-	14±1 ^b	16.33±0.57 ^a
<i>Candida parapsilosis</i>	ATCC 22019	-	12.67±0.58 ^b	17.33±0.57 ^a
<i>Candida krusei</i>	ATCC 6258	-	12±0 ^b	16±0 ^a
<i>Candida krusei</i> 1	PD-CHU FH	Ear pus	13.33±0.58 ^a	14.33±0.58 ^a
<i>Candida tropicalis</i> 1	PD-CHU FH	Vaginal	15.66±0.57 ^a	14±1 ^b
<i>Candida tropicalis</i> 2	PD-CHU FH	Vaginal	19±1.73 ^a	20±0 ^a
<i>Candida glabrata</i> 1	PD-CHU FH	Oral	20±0 ^a	19.33±0.58 ^a
<i>Candida glabrata</i> 2	PD-CHU FH	CBUE	18.33±1.53 ^b	21.33±0.58 ^a
<i>Candida glabrata</i> 3	PD-CHU FH	Vaginal	11.67±1.15 ^b	14±1 ^a
<i>Candida glabrata</i> 4	PD-CHU FH	Vaginal	11.33±0.58 ^b	13.33±0.58 ^a
<i>Candida Parapsilosis</i> 1	PD-CHU FH	Vaginal	12.66±0.57 ^a	13.33±1.15 ^a
<i>Candida sake</i> 1	MDC	Oral	14±0 ^b	16.33±0.58 ^a
<i>Candida sake</i> 2	MDC	Oral	13.33±0.58 ^b	15±0 ^a
<i>Candida kefyri</i> 1	MDC	Oral	11.66±0.57 ^b	15±0 ^a
<i>Candida kefyri</i> 2	MDC	Oral	13.66±1.15 ^b	17.33±1.15 ^a
<i>Candida kefyri</i> 3	MDC	Oral	17±1 ^a	16.67±0.58 ^a
<i>Candida holmii</i> 1	MDC	Oral	14.83±0.76 ^a	15±0 ^a
<i>Candida holmii</i> 2	MDC	Oral	20±1 ^b	22±0 ^a
<i>Candida albicans</i> 1	MDC	Oral	12.33±0.57 ^b	14.5±0.5 ^a
<i>Candida albicans</i> 2	MDC	Oral	16±1 ^a	17±1 ^a
<i>Candida albicans</i> 3	MDC	Oral	18±1 ^b	20.67±0.58 ^a
<i>Candida albicans</i> 4	MDC	Oral	15.66±0.58 ^b	18.33±0.58 ^a
<i>Candida albicans</i> 5	MDC	Oral	18.67±1.15 ^b	20±0 ^a
<i>Candida albicans</i> 6	MDC	Oral	13±1 ^b	16±0 ^a
<i>Candida albicans</i> 7	MDC	Vaginal	13±1 ^b	15.33±0.58 ^a
<i>Candida albicans</i> 8	MDC	Vaginal	16±1.73 ^a	16±0 ^a
<i>Candida albicans</i> 9	MDC	Vaginal	13±0 ^b	15.33±0.58 ^a
<i>Candida albicans</i> 10	PD-CHU FH	Vaginal	13.33±0.58 ^b	17.33±0.58 ^a
<i>Candida albicans</i> 11	PD-CHU FH	Vaginal	14±1 ^a	14.67±0.58 ^a

SD: Standard deviation; IZ: Inhibition zone diameter (mm) around the discs (6mm) impregnated with 10 µl of essential oil and 20 µg/disc for Amphotericin B (Amp. B); Each value represents the average of 3 repetitions. Means followed by the same letters are not significantly different at $p= 0.05$ based on Duncan's multiple range test. Small letters (a and b) are used to compare means between IZ EO and IZ Gentamycin for the same strain; PD-CHU FH: Parasitology Department-CHU Farhat Hached; MDC: Monastir Dental Clinic; CBUE: Cytobacteriological Urine Exam.

Mechanistically, EOs penetrate into bacterial cell via the membrane inducing a loss of ions, reduction of membrane potential, destruction of proton pump, disruption of bacterial enzyme systems and finally, lysis of the cells [33]. We outlined also that the antimicrobial activity of EOs cannot be easily ascribed to a specific compound, but it depends essentially on the synergistic or antagonistic effects of different compounds in EOs cause antimicrobial activity. EOs composition comprises more than one main compounds with a high concentration that estimate their biological potential. In our study, the biological activity MOEO was significantly related the presence of high level of oxygenated monoterpenes with the main components this oil, are 1,8-cineole (39.80%) followed by citronellol (16.66%), geraniol (12.25%). 1,8-Cineole as the most active compound, plays an important role as adjuvant for antimicrobial

purposes since it increases antimicrobial activities when it was combined together with other chemical antibacterial agents were such as mupirocin [34] and chlorhexidine digluconate [35]. Geraniol and citronellol, as two monoterpenes' alcohols produced from combination of two isoprene units have been proven for their efficiency against *Trichophyton rubrum* involves inhibition of ergosterol biosynthesis [36].

4. CONCLUSION

This study demonstrated that MOEO has a significant antioxidant and antimicrobial properties. The EO explained a significant antioxidant capacity better than that obtained by synthetic antioxidant (BHT). These potentials are closely related to the chemical composition showed a wealth of compound interest such as 1,8-Cineole, citronellol geraniol, myrcene and geraniol. The richness in oxygenated monoterpenes makes this EO widely used in various applications. In fact, MOEO represents a source of important bioactive compounds with antimicrobial properties which merit further study for possible application in the food industry, where they may be considered as natural preservatives to replace the synthetic ones which consumers are increasingly distrustful. The important actions of the EO against a wide range of strains known for their pathogenicity, let us to suggest that *M. officinalis* can provide an important source of antiseptic agents which can be deserved to be used in pharmaceutical and cosmetic industries.

CONSENT

It is not applicable.

ETHICAL APPROVAL

It is not applicable.

NOTE:

The study highlights the efficacy of "traditional medicine" which is an ancient tradition, used in some parts of India. This ancient concept should be carefully evaluated in the light of modern medical science and can be utilized partially if found suitable.

COMPETING INTERESTS DISCLAIMER:

Authors have declared that no competing interests exist. The products used for this research are commonly and predominantly use products in our area of research and country. There is absolutely no conflict of interest between the authors and producers of the products because we do not intend to use these products as an avenue for any litigation but for the advancement of knowledge. Also, the research was not funded by the producing company rather it was funded by personal efforts of the authors.

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