

Tissue specific antioxidant response of *Cirrhinus mrigala* (Hamilton, 1822) exposed to lead chloride

ABSTRACT

In the present study, the effect of Lead chloride (PbCl_2) on the peroxidase and catalase activity in the tissues (liver, kidney, gills and muscles) of *Cirrhinus mrigala* was studied. *Cirrhinus mrigala* were observed at sub-lethal concentrations viz. $1/3^{\text{rd}}$ LC_{50} and $1/7^{\text{th}}$ LC_{50} for 28 days. After chronic exposure fishes were dissected and sampled weekly. The activity of peroxidase enzyme was increased significantly ($p < 0.01$) in experimental fish organs as compared to control group. Metal stressed maximum peroxidase activity was found as $0.997 \pm 0.011 \text{ UmL}^{-1}$ (liver), $0.676 \pm 0.016 \text{ UmL}^{-1}$ (kidney), $0.489 \pm 0.005 \text{ UmL}^{-1}$ (gills) and $0.339 \pm 0.006 \text{ UmL}^{-1}$ (muscles) in the fishes exposed to $1/3^{\text{rd}}$ LC_{50} . The catalase enzyme activity decreased significantly in experimental fish organs as compared to control group. Minimum catalase enzyme activity was measured as $592.55 \pm 3.76 \text{ UmL}^{-1}$, $577.32 \pm 8.64 \text{ UmL}^{-1}$, $547.10 \pm 12.65 \text{ UmL}^{-1}$ and $488.21 \pm 28.57 \text{ UmL}^{-1}$ in the metal stressed liver, gills, kidney and muscles respectively, in fishes exposed to $1/3^{\text{rd}}$ LC_{50} . Catalase activity was found significantly higher in the control fish.

Keywords: Fish toxicology, *Cirrhinus mrigala*, Lead Exposure, Antioxidants, Peroxidase & Catalase enzymes.

INTRODUCTION

Aquaculture is a newly developed industry with significant potential for improvement. Water quality management faces greater problems than at any time in its history due to natural pollutants. Varied contaminants exist in the surface waters including multiple chemical compounds and different products of industrial and agricultural revolution (Shahid *et al.*, 2013). In recent decades of extensive urbanization and industrialization, the use of chemical fertilizers and pesticides has increased the concentration of heavy metals in the aquatic ecosystem (Ambreen and Javed, 2015). Heavy metals are major toxicants in the aquatic environment. Because of their toxic influence, heavy metals can severely alter the density and diversity of aquatic biota (Nishida, 2011). Functions of vital enzymes are also affected due to oxidative stress caused by heavy metals (Thangam *et al.*, 2014).

Lead (Pb) is a common, ubiquitous and persistent environmental pollutant with its increased worldwide production of about 2.5 million tons per year (Osfor *et al.*, 2010). Lead acts as most hazardous poison in aquatic climate. When the fish exposed to Pb it accumulates in fish organs and cause renal and hepatic dysfunction with retardation in fish growth. Pb exposure damage the organs of fish and induce changes in pathological, hematological and serum biochemical parameters (Patrick, 2006).

Aquatic species have a protective mechanism to minimize ROS before the harmful effects occur. This system consists of antioxidant enzymes, viz. superoxide dismutase, catalase, glutathione peroxidase, glutathione S-transferase, and glutathione reductase, which are responsible for converting injurious ROS into such products that are less harmful (Goswami and Das, 2016). Superoxide dismutase (SOD) serves as a first line of protection among these antioxidant enzymes, transforming the superoxide radical into hydrogen peroxide, which is then converted by catalase/peroxidase to oxygen and water (Ahmad *et al.*, 2005). Peroxidase is the member of antioxidant family and is considered a major enzyme that is responsible for reducing hydrogen peroxide. Peroxidase enzyme protects the red blood cells from the harm of spoilage and destruction created and enhanced by H₂O₂. Results suggested that this enzyme might guard tissues and defend the body greatly from oxidation troubles produced and propagated by lipid per oxidation. The activity of enzyme may get higher due to environmental toxins (Valavanidis *et al.*, 2006).

Catalase as a primary antioxidant defense component, protects fish from oxidative stress by converting hydrogen peroxide to oxygen and water. This enzyme has important functions such as ion transport, maintenance of electrochemical gradient and regulation of cell volume (Gul *et al.*, 2004). Heavy metals accumulate in fish through various routs like skin and gills which are considered as the main entrance of pollutants, oral intake of water and the use of other polluted organisms as **food. Metallic** ions of high toxicity are known to cause injurious effects on the organs and blood composition of fish. Fish tissues have antioxidant protection mechanisms which consists of superoxide dismutase, catalase, and peroxidase. Superoxide dismutase and peroxidase which guard the tissues against the **damaging** effect of heavy metals (Kousar and Javed, 2015).

Gills are the main organ of fish, assuming a multifunctional part in the complex function's performance, for example, acid base equilibrium, osmoregulation, respiration process, and excretion of nitrogenous wastes. By attaching themselves to

mucous layer of the gills, metals may enter the gills and cause modifications in the ultrastructure and general morphology of fish gills. Gills are the first target of water borne pollutants due to the direct contact with water (Barhoumi *et al.*, 2012).

The organs most associated with the detoxification and biotransformation processes are liver and kidney, and are peak absorption center of heavy metals in diverse fish forms. Fish tissues, specifically the liver and kidney are endowed with an antioxidant defense system to protect them from an oxidative stress caused by metals (Azmat and Javed, 2012). Liver is a major site for detoxification of reactive oxygen species (ROS). One of the most important functions of the liver is to detoxify the body from pollutants. Therefore, it may be considered as an indicator of aquatic pollution (Vesey, 2010). Kidney plays an important role because it maintains the homeostasis and responsible for the excretion of chemical wastes from the body of animals. Impact of heavy metals on aquatic ecosystem can be evaluated by measuring the biochemical factors/parameters in the kidney of the fish that respond specifically to the degree and type of contaminants (Bashir *et al.*, 2018).

Cirrhinus mrigala commonly known as mrigal carp is native to Pakistan riverine system. Because of nutritive quality of fish the Indian major carp *C.mrigala* is economically important to culture in Pakistan. The population explosion of the world is geometrical whereas the increase in food resources is arithmetical, a threat to mankind because of food security (Chavan and Mulley, 2014). *C. mrigala* is not listed in IUCN red list as threatened species.

MATERIALS AND METHODS

The present research was conducted in Toxicology and Limnology Laboratories at Fisheries Research Farm, Department of Zoology, Wildlife and Fisheries, University of Agriculture, Faisalabad.

Cirrhinus mrigala was chosen as an experimental organism. The samples of fish were collected and brought to the laboratory for acclimatization. Fish was acclimatized for 15 days in dechlorinated tap water in glass aquaria having 60-liters. capacity.

Experimental Design:

To carry out the chronic toxicity test, the fishes were randomly selected and

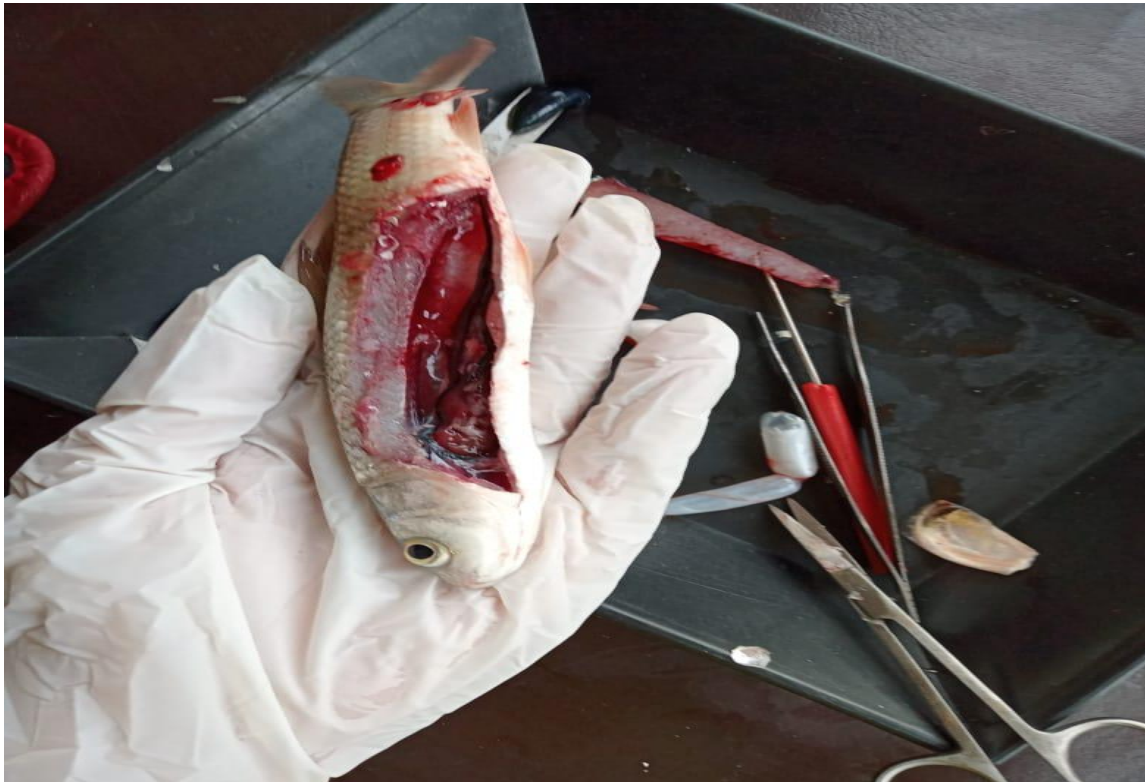


Figure 1 Dissection of *Cirrhinus mrigala*

stocked in 8 nos. in each aquarium with proper oxygenation.

Chronic exposure:

Sub-lethal concentrations ($1/7^{\text{th}}$ LC_{50} and $1/3^{\text{rd}}$ LC_{50}) were given to different groups of fishes for 28 days. After chronic exposure fishes were dissected and the liver, kidney, gills and muscles were isolated and properly stored.

Antioxidant enzyme study:

To examine the antioxidant enzyme activity, organs were first isolated and homogenized. Homogenization was done through following steps:

- To study the antioxidant enzyme activity viz. catalase and peroxidase, organs were extracted and stored at -4°C .
- Weighed the stored organs and mixed with phosphate buffer having a pH of 6.5 (0.2M) to remove the RBCs.
- Pestle and mortar were used to homogenize the tissues in the cold buffer.
- To remove the rubbish, homogenized matter was passed through the muslin cloth.
- Whatman filter paper 1 was used to filter the fluid obtained from muslin cloth.
- Filtrate was centrifuged in a centrifuge machine for 15 minutes at 10,000 rpm.

- After completing the process of centrifugation, supernatants were preserved at -80°C for further examination.



Figure 2 Grinding and Separation of organs through homogenization

Catalase Assay:

Catalase activity was evaluated according to the method of Chance and Maehly (1955). Catalase activity was determined by measuring its ability to decline the hydrogen peroxide concentration per minute at 240nm. 0.224 g NaH_2PO_4 and 0.1632 g Na_2HPO_4 were taken in a flask and dissolved by adding distilled water. Then volume was raised up to 50 ml and adjusted the pH 7.0. 2ml buffer solution was prepared. A cuvette containing the 2 ml of blank solution (buffer) was placed into the spectrophotometer and set it to zero at wavelength of 240 nm. In a cuvette containing buffered substrate solution 0.05 ml of enzyme extract was added and put into the spectrophotometer. The reaction time was 3 minutes and the absorbance were noted after interval of 1 minute.

$$\text{Catalase activity (Units/mL)} = \frac{\Delta A/\text{min} \times \text{dilution} \times 2\text{ml}}{(0.04 \text{ M}^{-1}\text{cm}^{-1} \times 0.05\text{ml})}$$

2) Peroxidase Assay:

Peroxidase enzyme action was concluded by assessing its capability to reduce the hydrogen peroxide concentration at 470nm. Prepared 0.2 M phosphate buffer solution. 4g NaH_2PO_4 and 1g Na_2HPO_4 was taken in a flask and dissolved by adding distilled water. Then volume was raised up to 200 ml and adjusted the pH at 6.5. Also prepared buffered substrate solution 3 ml.

A cuvette containing the 3 ml of blank solution was placed into spectrophotometer and set it to zero at wavelength of 470 nm. In a cuvette containing buffered substrate

solution, 0.06 ml of enzyme extract was added and put into the spectrophotometer. The reaction time was 3 minutes and so absorbance was noted after 3 minutes.

$$\text{Peroxidase activity (Units/mL)} = \frac{\Delta A/3\text{min}}{26.6 \times 60 \mu\text{l} / 3000 \mu\text{l}}$$

RESULTS AND DISCUSSION

The laboratory experiments were performed to evaluate the effect of lead chloride on peroxidase and catalase enzyme activity in the tissues (liver, kidney, gills and muscles) of *Cirrhinus mrigala*. *Cirrhinus mrigala* was exposed to various sub-lethal concentrations of lead chloride (PbCl_2).

Liver:

Peroxidase activity in fish liver during chronic exposure of lead chloride at sub-lethal concentrations was $1/7^{\text{th}} \text{LC}_{50}$ was $0.648 \pm 0.007 \text{ U mL}^{-1}$ and in $1/3^{\text{rd}} \text{LC}_{50}$ was $0.997 \pm 0.01 \text{ U mL}^{-1}$. Group treated with $1/3^{\text{rd}} \text{LC}_{50}$ recorded higher peroxidase activity in fish liver than other group $1/7^{\text{th}} \text{LC}_{50}$.

Catalase activity in the liver of the fish during chronic exposure of lead chloride at sub-lethal concentration of $1/7^{\text{th}} \text{LC}_{50}$ was $619.80 \pm 4.51 \text{ U mL}^{-1}$ and at $1/3^{\text{rd}} \text{LC}_{50}$ it was $592.55 \pm 3.76 \text{ U mL}^{-1}$. The activity of catalase was lower in both the treated groups as compared to the control group and the activity of catalase is lower in $1/3^{\text{rd}} \text{LC}_{50}$ as compared to $1/7^{\text{th}} \text{LC}_{50}$.

Kidney:

In this study it was observed that peroxidase activity was high in the kidney at $1/3^{\text{rd}} \text{LC}_{50}$ as compared to $1/7^{\text{th}} \text{LC}_{50}$ in the fishes when exposure to lead chloride. The activity of peroxidase in control group was $0.266 \pm 0.01 \text{ U mL}^{-1}$, and in $1/7^{\text{th}} \text{LC}_{50}$ exposed was 0.485 ± 0.01 and $1/3^{\text{rd}} \text{LC}_{50}$ was $0.676 \pm 0.01 \text{ U mL}^{-1}$ which was higher than the control group.

Catalase activity in fish kidney during exposure to lead chloride at sub-lethal concentrations $1/7^{\text{th}} \text{LC}_{50}$ was $577.06 \pm 1.88 \text{ U mL}^{-1}$ and at $1/3^{\text{rd}} \text{LC}_{50}$ was $547.10 \pm 12.65 \text{ U mL}^{-1}$ and in case of control group it was $605.19 \pm 0.65 \text{ U mL}^{-1}$. Activity of catalase was low in group with $1/3^{\text{rd}} \text{LC}_{50}$ as compared to $1/7^{\text{th}} \text{LC}_{50}$ and control group.

Gills:

Durations	Tissues	Treatments
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Activity of peroxidase was high in the gills of fish which were treated with 1/3rd LC₅₀ as compared to 1/7th LC₅₀ and control group during exposure of lead chloride. The activity of peroxidase in 1/3rd LC₅₀ was 0.489±0.005 UmL⁻¹, in 1/7th LC₅₀ was 0.296±0.022 UmL⁻¹ and whereas in control it was 0.197±0.004 UmL⁻¹

Catalase activity was lower in the gills of fish which was treated with 1/3rd LC₅₀ as compared to at 1/7th LC₅₀ and control group during exposure of lead chloride. The activity of peroxidase in 1/3rd LC₅₀ was 577.32±8.64 UmL⁻¹ and at 1/3rd LC₅₀ was 608.01±2.40 UmL⁻¹ and in control group is 621.04±0.74 UmL⁻¹.

Muscles:

Peroxidase activity in the muscles of the fish during chronic exposure of lead chloride sub-lethal concentrations 1/7th LC₅₀ was 0.206±0.006 UmL⁻¹ and at 1/3rd LC₅₀ was 0.339±0.006 UmL⁻¹. Peroxidase activity was higher in the muscles of fish group treated with 1/3rd LC₅₀ as compared to 1/7th LC₅₀ and control group.

Catalase activity in the muscles of the fish during chronic exposure of lead chloride sub-lethal concentrations 1/7th LC₅₀ was 526.24±7.66 UmL⁻¹ and at 1/3rd LC₅₀ was 488.21±28.57 UmL⁻¹. Activity of catalase was low in the muscles of fish I group treated with 1/3rd LC₅₀ as compared to 1/7th LC₅₀ and in control group.

Table showing Peroxidase activity (UmL⁻¹) in the tissues of *Cirrhinus mrigala* after chronic exposure of lead chloride.

		Control	1/7th LC₅₀	1/3rd LC₅₀
After 7 days	Liver	0.275±0.005	0.394±0.014	0.649±0.021
	Kidney	0.19±0.003	0.279±0.017	0.436±0.010
	Gills	0.189±0.006	0.239±0.007	0.419±0.006
	Muscles	0.085±0.006	0.167±0.006	0.219±0.006
After 14 days	Liver	0.278±0.004	0.562±0.007	0.871±0.007
	Kidney	0.223±0.004	0.491±0.009	0.628±0.032
	Gills	0.19±0.001	0.298±0.002	0.481±0.003
	Muscles	0.110±0.007	0.189±0.006	0.318±0.004
After 21 days	Liver	0.283±0.005	0.732±0.004	1.114±0.007
	Kidney	0.291±0.051	0.549±0.008	0.749±0.013
	Gills	0.198±0.004	0.286±0.071	0.514±0.004
	Muscles	0.141±0.007	0.222±0.006	0.391±0.009
After 28 days	Liver	0.287±0.006	0.903±0.003	1.354±0.009
	Kidney	0.358±0.002	0.622±0.013	0.891±0.008
	Gills	0.209±0.004	0.362±0.008	0.543±0.006
	Muscles	0.172±0.016	0.246±0.008	0.432±0.005
Overall Mean±S.D	Liver	0.281±0.005	0.648±0.007	0.997±0.011
	Kidney	0.266±0.015	0.485±0.012	0.676±0.016
	Gills	0.197±0.004	0.296±0.022	0.489±0.005
	Muscles	0.127±0.009	0.206±0.006	0.339±0.006

NS= Non-significant (P>0.05); *=Significant (P<0.05); **= Highly Significant (P<0.01)

Conclusion

It was observed that peroxidase enzyme activity was found significantly increased (p<0.01), and catalase enzyme activity was found significantly lower in the lead stressed fishes as compared to the control fishes. Among the tissues, fish liver exhibited significantly higher activity of peroxidase and catalase show decline trend Liver > gills > kidney > muscles.

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