

Internal combustion engine alternative fuels

ABSTRACT

With the increasing concern for environmental protection and sustainable development, alternative fuels for internal combustion engines have become a topic of great concern. Traditional fuels such as oil not only have the problem of emission pollution, but also have the problem of limited quantity and easy to be influenced by political factors. Therefore, finding suitable alternative fuels is an important guarantee to ensure the sustainable development of people's lives and society. In recent years, some new fuels have been developed, such as biomass energy, hydrogen fuel, methanol, etc. The development and use of these alternative fuels can reduce the impact of traditional fuels on the environment and reduce dependence on fossil energy [1].

However, the research and use of alternative fuels also presents a series of problems, such as cost, energy density, sustainability, and so on. Therefore, the performance, feasibility, advantages and disadvantages of alternative fuels need to be fully studied and evaluated [2]. On the basis of existing research, this paper will conduct an in-depth discussion from the aspects of combustion characteristics, emission characteristics, advantages and disadvantages of alternative fuels, so as to provide certain theoretical support and practical experience for the research and application of alternative fuels, and promote the development and application of alternative fuel technology.

Keywords: Alternative fuel; Pollutants; Clean energy

1.INTRODUCTION

As a common mechanical device, the internal combustion engine is an indispensable part of many vehicles and machine equipment. However, with the deterioration of the earth's environment, the research and application of alternative fuels has become an important problem that must be solved. Although traditional fuels, such as oil and natural gas, have the advantages of high energy density and easy storage, they also bring serious pollution problems, which makes the development of alternative fuels attract wide attention [3]. Therefore, in order to protect the environment and promote sustainable development, exploring new alternative fuels has become one of the current research hotspots [11-15]. Starting from the relationship between internal combustion engine and traditional fuels, this paper will discuss the application and development status of various alternative fuels, and analyze the advantages and disadvantages of various alternative fuels, so as to provide some references for the research in related fields.

2 THE IMPORTANCE OF ALTERNATIVE FUELS

2.1 DISADVANTAGES OF CONVENTIONAL FUELS

Traditional fuels, such as oil and natural gas, play an important role in the economy and development of human society. But at the same time, these traditional fuels also have many drawbacks and negative effects. The disadvantages and effects of traditional fuels will be analyzed in this paper.

1. Environmental pollution

The gases and particulate matter produced by the combustion of traditional fuels not only have a serious impact on air quality, but also cause oxidation and damage to the atmosphere and the ecological environment. For example, carbon dioxide emissions are one of the main

causes of global warming and climate change; Emissions of sulfur oxides and nitrogen oxides can lead to problems such as acid rain and light pollution [4]. At the same time, studies have shown that the toxic gases and particles emitted can have adverse health effects on humans and large animals.

2. Waste of resources

The extraction and utilization of traditional fuels is a resource-intensive industry, and the reserves of these fuels are limited. Due to the growing demand for fuel and increasingly scarce reserves, it will lead to the waste and absence of existing energy resources.

3. Economic impact

The economic benefits of reliance on conventional fuels cannot be ignored, but at the same time the prices of these fuels are constrained by supply, demand and geography, and are subject to market and political fluctuations. For example, oil federation monopolies and sharp fluctuations in oil prices can have direct and indirect impacts on global markets and trade [16,17].

To sum up, the disadvantages and effects of traditional fuels are obvious. They destroy the environment, waste resources and have a negative impact on the economy. Therefore, it is essential to explore alternative fuels and develop relevant policies to mitigate the impact of traditional fuels and promote sustainable development.

2.2 THE IMPORTANCE OF ALTERNATIVE FUELS

With the development of the world economy, the increase of the global population and the acceleration of urbanization, the problems of energy demand and environmental pollution are becoming increasingly serious. The emphasis on environmental protection and sustainable development has become a global hot topic [18-20]. Finding alternative fuels to replace traditional fuels has become a strategic goal for many countries, and the importance of this initiative is mainly reflected in the following aspects.

1. Reduce pollution and improve the environment

The burning of traditional fuels releases a large number of harmful gases and particulate matter, leading to increasingly serious environmental problems such as air

pollution and global climate change. The use of new alternative fuels can reduce the emission of these harmful substances, reduce the negative impact on the environment, help to improve the human living environment, protect and maintain the balance of the ecosystem.

2. Promote sustainable development

The mining, processing and conversion of traditional fuels require a lot of energy and resources, while alternative fuels mainly use renewable energy such as biomass, solar energy, water flow, ocean energy, geothermal energy and so on. Renewable energy can be more sustainable, while also promoting sustainable development, ensuring future energy supplies and meeting human energy needs.

3. Promote economic development

With the continuous improvement of global renewable energy technology and the decline of costs, the use of new fuels will reduce production costs and business operating costs, improve the competitiveness and sustainable development of the industry, and create new opportunities and development space in the process of energy transformation.

4. Strengthen national security

By reducing dependence on conventional fuels, reducing geopolitical risks and dependence on imported fuels in the energy sector, renewable energy can enhance a country's energy security and autonomy.

To sum up, the search for alternative fuels is essential to tackle environmental pollution and promote sustainable development. A truly global deployment of alternative fuels through innovative technologies and policy measures will create a more sustainable, clean and prosperous future for human society.

3 NATURAL GAS

3.1 PHYSICAL AND CHEMICAL PROPERTIES OF NATURAL GAS

Natural gas is a mixed gas that is primarily composed of methane (CH₄), but also contains

small amounts of hydrocarbons such as ethane, propane, and butane, as well as some gaseous impurities such as nitrogen, carbon dioxide, and hydrogen sulfide. Its physical and chemical properties include the following aspects: For air, the density of natural gas is low, so it usually rises and quickly disperses in the air, without depositing near the ground, which also increases the safety hazard after gas leakage. Natural gas has a high calorific value, usually 35-50MJ/m³. The ignition point of natural gas is relatively low, between 650-750 ° C. The boiling point of natural gas varies with the composition, generally between -165 ° C and -60 ° C. Natural gas is a flammable substance that burns easily in the air, and the carbon dioxide and water vapor produced during combustion are harmless to the environment. Natural gas may contain hydrogen sulfide (H₂S), which is a toxic gas that needs to be controlled and disposed of during handling and use to ensure safety. Natural gas is a volatile gas that is gaseous at room temperature but can be compressed into a liquid state for storage and transportation. In addition, natural gas also has the characteristics of clean, efficient, easy to use, colorless and odorless, and has become one of the important energy resources. As shown in Table 1, the physical properties of compressed natural gas are compared with those of other fuels.

Table 1 Comparison of physical properties between natural gas and other vehicle fuels

Physicochemical properties	Compressed natural gas	Gasoline	Diesel oil	Liquefied petroleum gas	Ethyl alcohol
Molecular formula	The main component is CH ₄	C ₅ ~C ₁₂	C ₁₀ ~C ₂₁	C ₃ H ₈	C ₂ H ₅ OH
density /x10 ⁻³ ·kg·m ⁻³	-	0.72~0.78	0.82~0.86	0.506	0.725
Boiling point /°C	-162	30~190	180~370	-	78

Theoretical	16.4	14.7	14.3	15.33	9.05
air-fuel ratio					
Spontaneous					
combustion	650	500	350	504	420
temperature					
Low calorific					
value of fuel /MJ·kg ⁻¹	50	44	42.5	46.42	27.2

3.2 COMBUSTION CHARACTERISTICS OF NATURAL GAS

The main component of natural gas is methane (CH₄). When burned, methane chemically reacts with oxygen to produce water vapor (H₂O) and carbon dioxide (CO₂) while releasing a large amount of heat. This combustion reaction is efficient because methane is a hydrocarbon with a simple molecular structure that is easy to burn and does not produce the large amounts of harmful gases and particles produced by traditional fuels such as coal or oil [5].

3.3 EMISSION CHARACTERISTICS OF NATURAL GAS

In the case of full combustion, the main emissions of natural gas are water vapor and carbon dioxide, which have relatively little impact on the atmospheric environment. In contrast, burning natural gas produces fewer carbon dioxide emissions than coal and oil and is therefore considered a cleaner fuel option. Although the main substances emitted by natural gas combustion are water vapor and carbon dioxide, a small amount of harmful gases such as nitrogen oxides (NO_x) and carbon monoxide (CO), as well as particulate matter [6-7], are also produced in the actual combustion process. The emissions of these substances depend on factors such as combustion conditions, fuel quality and the efficiency of combustion equipment. Natural gas has a higher spontaneous combustion temperature than gasoline and diesel and is safer to use in transportation.

3.4 THE ADVANTAGES AND DISADVANTAGES OF NATURAL GAS AS A VEHICLE FUEL

The boiling point and density of liquid natural gas are larger than kerosene, and the technical problems caused by low temperature and volume are easy to solve, so liquid natural gas can be used as a transitional vehicle fuel. The advantages of liquefied natural

gas engine are: it can reduce the emission of nitrogen oxides; Can reduce particulate matter emissions; Can reduce the smoke; Abundant global reserves; The use cost is not high. However, the storage temperature of liquid natural gas is -160°C , which is difficult to use in practice. Liquid natural gas can only be used alone, not mixed with petrochemical gasoline; Due to the limitations of the layout of the filling station network, it is difficult to popularize and can not be large-scale development in the short term.

4 HYDROGEN

4.1 HYDROGEN FUEL

Hydrogen is flammable, and when it mixes with oxygen and comes into contact with an ignition source, it burns violently, releasing a lot of energy. This property makes hydrogen an important fuel. Hydrogen has a melting point of -259.2°C and a boiling point of -252.9°C . These temperatures are much lower than those at normal temperature and pressure, so hydrogen usually exists as a gas at room temperature. Hydrogen can penetrate into many materials, including metals, plastics, and rubber. [8] This property needs to be considered in hydrogen storage and delivery to prevent hydrogen leakage and loss. Hydrogen in the production, transportation and storage are required to have a very high sealing, the use of the process will face great security challenges.

Table 2 Comparison of physical properties between hydrogen and diesel gasoline

Physicochemical properties	hydrogen	Diesel oil	gasoline
Molecular formula	H ₂	C ₁₀ -C ₂₂	C ₅ -C ₁₂
Boiling point / $^{\circ}\text{C}$	-252.77	175~361	40~210
Theoretical air-fuel ratio	34.5	14.3	14.7
Lower calorific value /MJ·kg ⁻¹	121	42.5	43.9
Latent heat of vaporization /kJ·kg ⁻¹	-----	230	290
Spontaneous combustion temperature / $^{\circ}\text{C}$	400	260	350

Flash point /°C		55	-45
Octane number	130	20~30	80~98
Cetane number		40~56	5~25

4.2 COMBUSTION CHARACTERISTICS OF HYDROGEN

Hydrogen is a widely used fuel in the energy and industrial fields, and its combustion characteristics are of great significance for understanding the application and safety of hydrogen. The combustion characteristics of hydrogen are discussed in detail below. Burning hydrogen in the presence of oxygen produces water, releasing a lot of heat. The combustion of hydrogen is an exothermic reaction, releasing about 286 kJ of heat per gram of hydrogen combustion, more than twice that of many common fuels. Hydrogen burns at a higher rate than many other fuels, which means it can release energy faster under the same conditions. Hydrogen has a low ignition temperature of about 500°C, which means it is easily ignited and burns quickly. The main product of hydrogen combustion is water, which does not produce smoke or harmful gases, so it is considered a clean fuel. Hydrogen is flammable, and hydrogen mixed with air can be ignited very easily under the right conditions. Hydrogen can form explosive mixtures under the right conditions, so special care is needed when handling and storing hydrogen. In short, hydrogen as a fuel has unique combustion characteristics, its high combustion speed, high ignition temperature and smokeless combustion characteristics make it have important application prospects in energy and industrial fields. However, the flammable and explosive nature of hydrogen also requires us to maintain a high degree of vigilance when using and storing it to ensure safety.

Because of its excellent physical and chemical characteristics and zero carbon emission advantages, hydrogen is widely used in the field of automotive engine is the most promising clean and efficient alternative fuel. There are two main combustion modes of hydrogen used in engines: one is hydrogen-doped combustion [9]; The

other is pure hydrogen combustion [10]. Due to its physical and chemical characteristics, hydrogen has many abnormal combustion phenomena in internal combustion engine. The common abnormal combustion of hydrogen engine includes premature combustion, tempering, knocking and so on.

4.3 HYDROGEN EMISSION CHARACTERISTICS

Hydrogen emission characteristics are critical to issues such as environmental protection and climate change. Understanding the emission characteristics of hydrogen can help assess its sustainability as an energy source and take steps to reduce its adverse environmental impact.

The main product of hydrogen combustion is water, which makes hydrogen considered a clean fuel. Water discharge is usually harmless to the environment. Compared to conventional fuels, hydrogen combustion does not produce greenhouse gases such as carbon dioxide, so it is considered a zero-emission energy option. This gives hydrogen potential in addressing the challenge of climate change.

Although hydrogen combustion does not produce carbon dioxide, it may generate nitrogen oxides such as nitrogen oxide under high temperature combustion conditions, which have a certain impact on the atmospheric environment. By optimizing the combustion process and controlling the combustion temperature, the generation of harmful substances such as nitrogen oxide can be reduced.

To sum up, hydrogen as an energy source has unique emission characteristics, mainly discharging water and oxygen, and no carbon emissions during combustion. However, its production process may involve the emission of carbon emissions and other harmful substances, so it is necessary to comprehensively consider its life cycle emissions when promoting the hydrogen economy, and take corresponding emission reduction measures.

4.4 ADVANTAGES AND DISADVANTAGES OF HYDROGEN AS A VEHICLE

FUEL

(1) The advantages of using hydrogen energy

1. Reduce local pollution.
2. Reduce global CO₂ emissions.

(3) Resolve the constraints of non-renewable energy such as oil and natural gas on human development.

(2) Technical problems and development status of hydrogen internal combustion engine

1. On-board hydrogen storage: The problem of hydrogen storage has always been a problem in the research of hydrogen energy-related technologies. The small size of hydrogen molecules makes it easy to penetrate into the storage tank materials and hydrogenate the materials, resulting in hydrogen embrittlement. At the same time, the flammable and explosive nature of hydrogen restricts the application scenarios of hydrogen energy. Especially in the vehicle engine, the need to store enough hydrogen energy fuel in a limited space to ensure the endurance is a huge challenge. At present, the research direction of hydrogen storage mainly includes high-pressure gaseous hydrogen storage, low-temperature liquid hydrogen storage and hydrogen storage materials.

2. Deflagellation, early ignition and tempering hydrogen: the problems of internal combustion engines are inseparable from their advantages. Hydrogen combustion propagation speed is very fast (about 9 times that of gasoline combustion), will lead to combustion time is too short, combustion work time is short, can not overcome the compression work, easy to lead to engine stalling, that is, deflagration problem; Secondly, because the hydrogen ignition point is low, the overheating of the spark plug motor in the internal combustion engine, hot deposits, etc., will lead to natural hydrogen and early ignition problems; At the same time, because the combustion propagation speed is fast, the intake valve is not closed at this time, the flame will enter the intake pipe, and the tempering phenomenon will occur.

3. Emissions: Hydrogen internal combustion engines theoretically have five emissions of H₂, HC, CO, CO₂ and NO, of which CO, CO₂ and HC are three pollutants produced by oil combustion, and the emission concentration is small. As the main emission of hydrogen internal combustion engine, it is formed by the reaction of nitrogen and oxygen in the cylinder at high temperature, and the emission can reach up to

0.02g/kW·h. The emission control is the focus of controlling the emission of hydrogen internal combustion engine.

5 GENERAL JUNCTION

The internal combustion engine is a common type of engine, but its use of fossil fuels also brings environmental and resource problems. Therefore, alternative fuels have become one of the important directions that the internal combustion engine can develop. At present, alternative fuels for internal combustion engines are mainly concentrated in hydrogen energy and liquid natural gas.

As an alternative fuel for internal combustion engine, hydrogen energy is widely used in hydrogen fuel cell vehicles and hydrogen-burning engines. The advantages are low pollution, high efficiency, and energy storage can be stored through renewable energy forms such as solar photovoltaic power generation, but there are also security risks and expensive development, storage and transportation costs. Liquefied natural gas (LNG) is an alternative fuel made from natural gas, which is widely used in public transportation, ships, trains and other fields. The advantage of liquefied natural gas is its environmental protection, renewable, low cost, so it is widely used in some fields, but it also contains some challenges, such as storage, transportation, equipment costs are higher.

In general, the alternative fuel of internal combustion engine has a broad prospect, but the application of alternative fuel has certain limitations, involving the cost of each link, production scale and technological innovation. In addition, it is also necessary to strengthen technological innovation and policy support to accelerate the application and development of alternative fuels in the field of internal combustion engines, so as to better meet the needs of society for low-carbon development and environmental protection, and achieve sustainable development.

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