

# STRUCTURAL AND MORPHOLOGICAL ANALYSIS OF UNTAPPED CLAY DEPOSITS IN ITU LOCAL GOVERNMENT AREA AKWA IBOM STATE, NIGERIA AND ITS POTENTIALS AS RAW MATERIAL FOR PETROLEUM REFINING

## Abstract

This work describes scanning electron microscope and XRD characterization of untapped clay deposits in Itu Local Government Area Akwa Ibom State and its potentials as raw material for petroleum refining. The structure and morphology of the raw kaolin clay and metakaolin were analyzed using XRD and SEM. The XRD data indicates 48.0 % quartz, 30.0% kaolinite, 3.0 % chlorite, 16.0 % illite and 2.0% albite in the raw kaolin and 68.3 % quartz, 17.3 % orthoclase, 5.9 % muscovite, 1.8 % illite , 6.68 %, albite in the metakaolin from Ikot Uso Akpan and 16.2 % kaolinite, 67.0% quartz, 2.5 % muscovite, 5.5 % orthoclase, 1.4 % chlorite, and 7.0 % albite were revealed in EkimItam raw kaolin. 66.0 % quartz, 10.0 % muscovite, 15.0 % orthoclase, 5.0 % illite, 5.0 % albite were recorded in the metakaolin. The XRD results confirmed the presence of quartz and alumina mineral phases from both kaolin deposits. Calcination at 1000 °C increased the crystalline structure of the metakaolin and did not significantly modify surface morphology of the kaolin indicating stability of kaolin structure, one the desired property of a good zeolite. The SEM results indicated heterogeneous size and spongy like porous shape. There was no significant change in the surface morphology of the heterogeneous size calcined kaolin compared with untreated kaolin. The SEM results of the metakaolin kaolin also indicated cubic crystalline structure with well-defined edges. With adequate modification the studied kaolin could be a promising raw material for synthesis of zeolite suitable for bio and fossil fuel processing.

KEY Words: Characterization, Kaolin, Metakaolin, Calcination, kaolinite

## INTRODUCTION

“Kaolin is a white, soft, plastic clay composed mainly of kaolinite,  $\text{Al}_4(\text{OH})_8[\text{Si}_4\text{O}_{10}]$ , and other related clay minerals such as nacrite and dickite (Baker & Uren, 2010). Kaolin deposits are classified as primary or secondary according to their genesis; primary deposits originating in situ by alteration, whereas secondary deposits are of sedimentary origin” (Murray, 2012). “Primary deposits are formed directly from hydrothermal alteration of volcanic and granitic rocks with examples as hydrothermal and volcanic clay types. Secondary clay deposits are products of weathering of pre-existing rocks, and subsequent alteration of the alumino-silicates into clays such as residual and sedimentary clays. Ideal conditions necessary to produce kaolinitic clays by means of chemical weathering are high rainfall, warm temperatures, lush vegetation, low relief and high groundwater table” Cravero & Dominguez, 1999). “Thus, kaolin is eroded and transported by streams to a quiet, fresh or brackish water environment. Post depositional leaching, oxidation, and diagenesis can significantly modify the original clay mineralogy with improvement of kaolin quality” (Hassan, 2014). “Kaolin is an important raw material with widespread industrial applications such as water treatment, as porcelain, cement and ceramics production” (Lima *et al.*, 2017)

“Clays have extensive applications due to their swelling, adsorption and ion exchange properties and high surface areas” (Jankovic *et al.*,2010). “Starting from the beginning of petroleum refining and petrochemical industries, clays have been used as catalysts” (Speight, 2006). “The first hydrocracking process developed over 80 years ago was based on acid-modified clays, but later and still now, zeolites and aluminosilicates were used. However, in few cases such as in treating some heavy fractions, clays are still used. Clays with different acidities can be obtained by thermal treatment before preparing the catalyst. The temperature of the thermal treatment of the clay determines the type and concentration of the hydroxyl groups and thereby its acidity. The most important clay used in the manufacture of catalysts is kaolin and has many applications such as a catalyst substrate in the catalytic cracking of petroleum” (Murray, 2007). “Furthermore, kaolin is used to synthesize zeolite and alumina” (Salahudeen *et al.*,2012). Kaolin or commonly referred as China clay is one type of mineral containing a crystal compound with the main content of kaolinite ( $\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2 \cdot 2\text{H}_2\text{O}$ ). Kaolin can be distinguished from other clay minerals due to its softness, white degree (whiteness) and easily dispersed in water or other solutions. Because of these physical properties, kaolin is widely used in the paper industry as a filler or coating material. According to the theory, pure kaolin contains 46% silicate, 40% alumina and 14 % water. Moreover, it is hardly found in kaolin that meets the composition, since there are other impurities such as titan ( $\text{TiO}_2$ ), iron ( $\text{Fe}_2\text{O}_3$ ), lime ( $\text{CaO}$ ) or potassium ( $\text{K}_2\text{O}$ ).

“The demand for high value petroleum products such as middle distillate, gasoline and lube oil is increasing. Therefore, maximizing of liquid products yield from various processes and valorization residues is of immediate attention to refiners. At the same time, environmental concerns have increased, resulting in more rigorous specifications for petroleum products, including fuel oils. These trends have emphasized the importance of processes that convert the heavier oil fractions into lighter and more valuable clean products” (Silvy, 2004). “A number of technologies have been developed over the years for residual oil upgrading, which include process that are based on the catalyst, carbon rejection and hydrogen addition routes” (Silvy, 2004). “Besides, petroleum products are the basic materials used for the manufacture of synthetic fibers for clothing and in plastics, paints, fertilizers, insecticides, soaps, and synthetic rubber. The uses of petroleum as a source of raw material in manufacturing are central to the functioning of modern industry” (Speight, 2006). Catalyst is the most powerful tool in the petroleum refining industry (Speight, 2006). Global capacity data of catalytic processing units in petroleum refineries collected by Silvy (Silvy, 2004) indicate that hydrotreating boasts the highest global capacity and fastest growth compared to other major catalytic processes such as fluid catalytic cracking (FCC), hydrocracking (HC), reforming, alkylation and isomerization (Samadhi *et al.*, 2011). “Catalyst depends on the rate of deactivation by coke and metal deposits and sintering of the active phases” (Bartholomew, 1994). Information regarding the activity, selectivity and deactivation of the individual catalyst is therefore highly desirable for optimizing the catalyst systems.

“Kaolin catalysts are commercially used catalysts” (Choudhury & Misra, 2011). “Kaolin has extensive applications due to their swelling, adsorption and ion exchange properties and high surface areas” (Jankovič *et al.*, 2010). Kaolin with different acidities can be obtained by thermal treatment before preparing the catalyst. The temperature of the thermal treatment of the kaolin determines the type and concentration of the hydroxyl groups (Herrero *et al.*, 2000), and thereby its acidity. Kaolin has many applications such as catalyst substrates in the catalytic cracking of petroleum (Murray, 2007). Further, kaolin is used to synthesize zeolite (Oghenejoboh & Ohimisor, 2011).

Zeolite is widely used as a catalyst in the petroleum and petrochemical industries because of its high concentration of active acid sites, its high thermal stability and high size selectivity. For example, ZSM-5 is widely used as a catalyst for isomerization, alkylation and aromatization processes (Oghenejoboh&Ohimisor, 2011). However, zeolite Y is the most important component in FCC catalysts (Htay & Oo, 2008). Today, synthetic zeolites are used commercially more often than natural due to the purity of crystalline products and the uniformity of particle sizes (Gonçalves *et al.*, 2009). Zeolites are usually synthesized from low cost silica-alumina sources in alkaline phase under hydrothermal conditions such as clays (Oghenejoboh&Ohimisor, 2011). The synthesis of zeolite from clay material is analogous to natural formation of zeolite from volcanic deposits or other high Si-Al material (Koukoulas, 2007)]. This is because both volcanic ash and natural clay are fine-grained and contain a large amount of aluminosilicate glass. The formation of natural zeolite may take thousands of years; however, laboratory preparation can shorten this period to hours (Oghenejoboh&Ohimisor, 2011). Kaolin is the most versatile industry mineral and is extensively used for synthetic zeolite (Wang *et al.*, 2007). The benefits of using kaolin as an aluminosilicate source in zeolite synthesis widely known (Oghenejoboh&Ohimisor, 2011). The analysis of the natural kaolin clay shows that the clay contains very low concentration of leachable metals (Pb, As, Zn) which negatively affects catalyst performance, thereby making the kaolin clay an excellent raw material for the production of zeolite for catalysis (Oghenejoboh&Ohimisor, 2011). Many variable parameters are strongly influenced by the formation of zeolite from kaolin. The important parameters are starting material composition, amount of added water, aging temperature, aging time crystallization temperature and crystallization time (Htay & Oo, 2008).

Kaolin is the most important clay used in the manufacture of catalysts. Because many catalysts are used at high temperatures and pressures, the refractory character of kaolin is appropriate for many applications. For most industrial applications kaolin must be refined and processed from the crude state to enhance its whiteness, purity and other important commercial characteristics (Naman *et al.*, 2012). The largest use of kaolin is in catalyst substrates in the catalytic cracking of petroleum (Murray, 2007). The purity of the kaolin is critical in this petroleum cracking operation so processed kaolin with low iron, titanium and alkali and alkaline earth compounds is preferred. It is estimated that over 200,000 tons of kaolin are used annually to produce petroleum cracking catalysts (Murray, 2007). Kaolin is the most important raw material in a matrix with zeolite as a catalyst in FCC process (Naman *et al.*, 2012), which kaolin clays modified the properties of catalytic cracking catalysts, such as reducing coke yield, improving attrition resistance, high tolerance to contaminated metal and high thermal stability (Campanati& Vaccari (2001). Kaolins are quite good catalysts for Diels-Alders reactions. They are also excellent supports for Lewis acids or for transition metals, to be applied in various organic reactions (Friedel-Crafts, acylation and alkylation of aromatics, etc.). Depending on the metal, the Brönsted versus Lewis acidity can be tuned, allowing obtaining a family of catalysts (Keith, 2002). In addition, kaolin has been identified as a potential raw material for the production of alumina, one of the most widely used catalyst support materials for hydrotreating (Wang, 2009). Kaolin and halloysite are used to make cracking catalysts, as a polymerization catalyst, peptide bond formation, and others (Murray, 2007). Halloysite and metakaolin are used in the manufacture of molecular sieves used as petroleum cracking catalysts. Kaolin for this application must be low in iron and alkaline earth elements. The dry surface acidic surface of kaolin is used to promote the polymerization of styrene, heterolytic breakdown of organic peroxides, dehydration of alcohols, hydrolysis of esters, and isomerization of alkenes (Murray, 2007,

Emam, 2013, Gawande &Kaware, 2016). It plays an important role in the treatment and finishing of petroleum fractions through desulfurization, deasphalting, bleaching of petroleum fractions. (Dudasova *et al.*, 2008). Naphtha treated with clay prevents gum formation in gasoline by removing diolefins (Speight, 2014). Clay minerals play a significant role in petroleum refining processes, new catalysts are developed from kaolin clay to meet the increasing demand of petroleum products with human and environmental health in view. Despite being blessed with abundant deposits of untapped kaolin, an important raw material used for synthesis of zeolite. Nigeria still imports zeolite and other industrial materials produced from kaolin which is locally available and untapped. Locally available raw materials and indigenous innovation and ingenuity in harnessing our natural resources should be appreciated, regulated, and the products assessed if they meet local and international specifications to mitigate environmental pollution (Udo *et al.*, 2020, Udo *et al.*, 2023). The quest for a sustainable clean environment has led to intensive research into alternative raw materials which will suit different industrial needs (Udo *et al.*, 2020). It is on this note that this research was designed to characterize some untapped clay deposits in Itu Local Government Area Akwa Ibom State, Nigeria using SEM and XRD. And assess its potentials as raw material for petroleum refining, water treatment based on the SEM and XRD results.

## **MATERIALS AND METHOD**

The kaolin samples used in this study were collected from clay deposits in Ikot Uso Akpan Itam and EkimItam, Itu L.G.A, Akwa Ibom State, using a clean garden fork. The sample was pulverized to enable its usability stored in a clean dry plastic container. The study area is bounded in the North and North-East by Odukpani in Cross River State and Arochukwu in Abia State, in the West by Ibiono Ibom and Ikono L.G.A, in the South and South-east by Uyo and Uruan Local Government Areas respectively. Itu has a latitude of 5.203624(5o12'13.05oN) and a longitude of 7.968822(7o58'7.76oE) (Bassey *et al.*, 2020, Udo *et al.*, 2023)

### **Hydrothermal Treatment of Kaolin Clay**

In the hydrothermal process, 50 g of dried clay sample was added into a 200 ml conical flask containing 100 ml of deionized water. The mixture was heated at 150 °C for one (1) hour with constant stirring using a hot plate with a magnetic stirrer, the resulting mixture was filtered and oven dried at 120 °C stored in lid tight containers for further use. Hydrothermal methods can transform kaolin into an ultrafine powder under controlled thermal conditions. The benefits of hydrothermal treatment of kaolin include increased reactants reactivity, low energy consumption, reduced air pollution, easy to control the solution, formation of metastable phases, and unique condensed phases (Johnson and Arshad 2014, Udo *et al.*, 2023)

### **Acid Leaching of Mesoporous-Kaolin**

The dried (mesoporous-kaolin) (50 g) prepared in the hydrothermal step was added to a 200 ml conical flask containing 100 ml of 11% hydrochloric acid (HCl). The resulting mixture was heated electrically at 150 °C with constant stirring using a magnetic stirrer for one (1) hour and filtered. The solid residue was washed with deionized water until no chloride was detected using AgNO<sub>3</sub> (Udo *et al.*, 2023)

### **Calcination of the acidified mesoporous –kaolin**

The acidified mesoporous -kaolin was oven dried at 120 °C for one (1) hour and calcined to 650 °C for two (2) hours in a muffle furnace and later cooled. The calcined kaolin was stored in a lid tight container and properly labeled. The heating process eliminates water from the mineral kaolinite ( $\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2 \cdot 2\text{H}_2\text{O}$ ), the main constituent of kaolin clay, and collapses the material structure, resulting in an amorphous aluminosilicate ( $\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2$ ), metakaolinite. The process is known as dehydroxylation  $\text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2 \cdot 2\text{H}_2\text{O} \rightarrow \text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2 + 2\text{H}_2\text{O} \uparrow$ .

## Results and Discussions

The results of the crystalline phases, percentage crystallinity, crystal structure of kaolin deposit from Ikot Uso Akpan Itam and EkimItam, Itu Local Government Area Akwa Ibom State Nigeria are presented Figure1 -Figure 8.

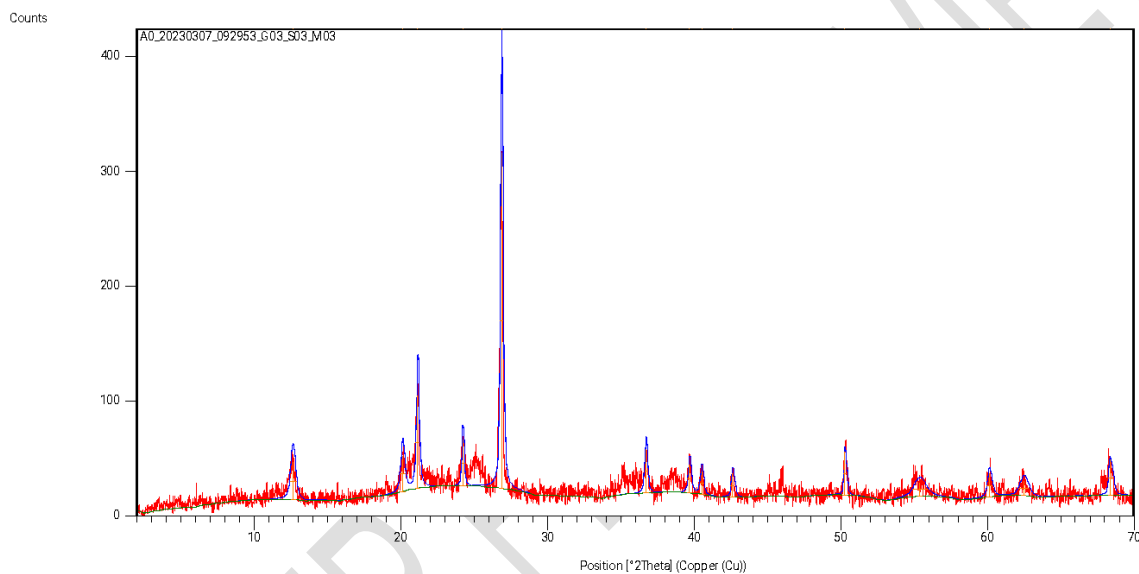


Fig. 1: XRD Pattern of Ikot Uso Akpan Itam Untreated Kaolin

Plot of results

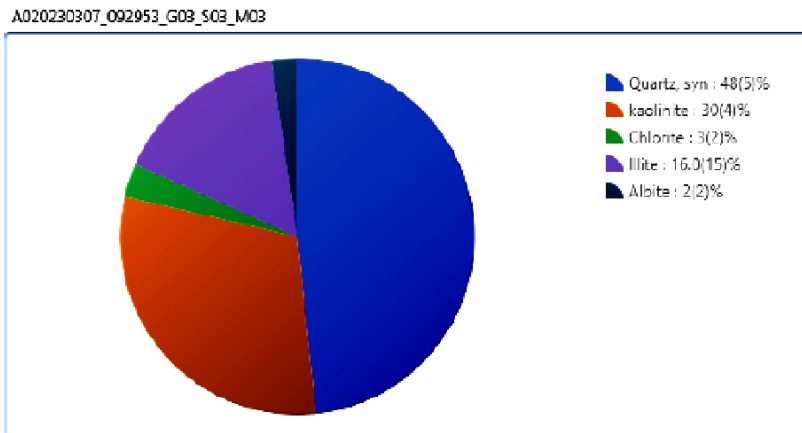


Table of results

Dataset / Weight Fraction, wt%	Value, Unit	Quartz, syn	kaolinite	Chlorite	Illite	Albite
A0_20230307_092953_G03_S03...	0	48(5)	30(4)	3(2)	16.0(15)	2(2)

Fig 2: Percentage quartz, kaolinite, chlorite, illite and Albite for natural (untreated) kaolin sample from Ikot Uso Akpan Itam .

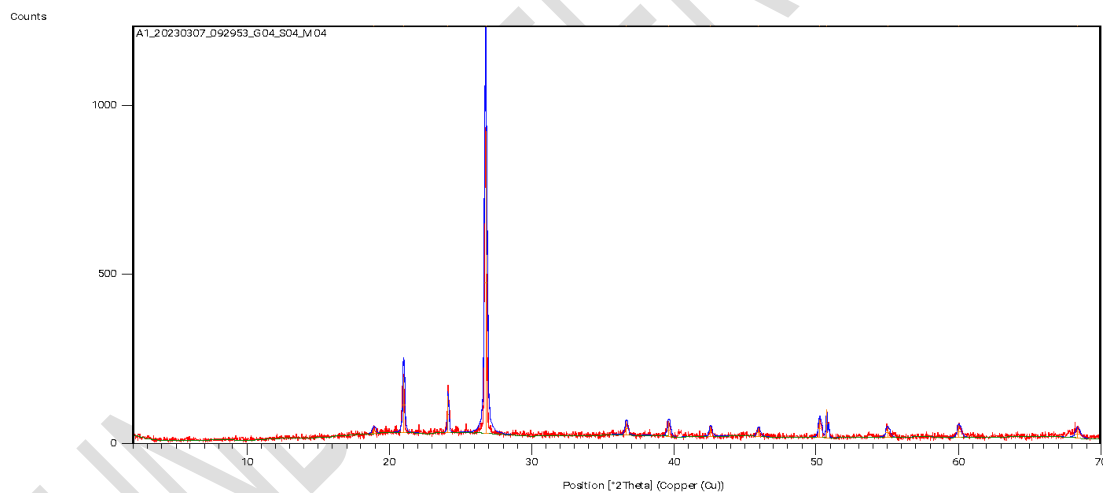


Fig. 3: XRD Pattern of Ikot Uso Akpan Itam Calcined Kaolin

A120230307\_092953\_G04\_S04\_M04

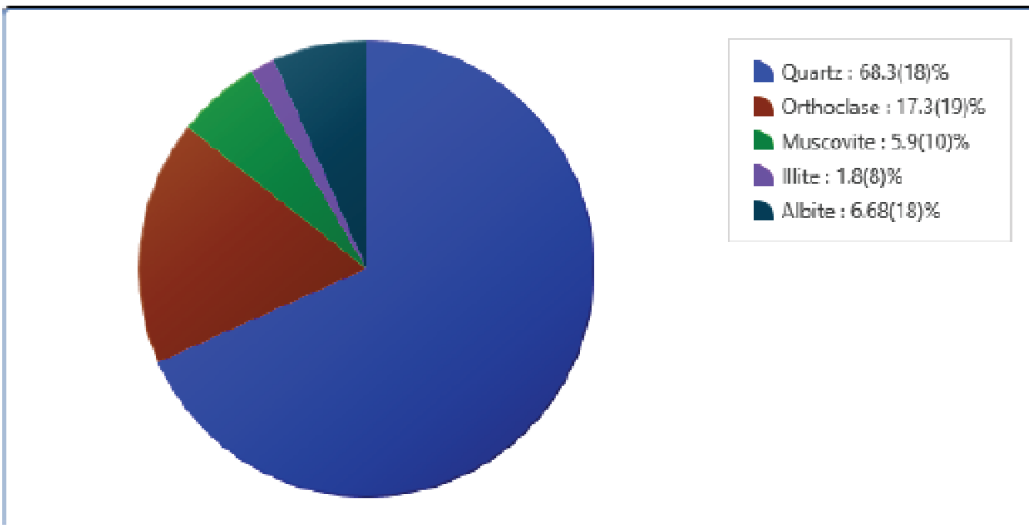


Fig 4: Percentage quartz, Orthoclase, Muscovite, illite and Albite in calcined kaolin from Ikot Uso Akpan Itam Calcined Kaolin

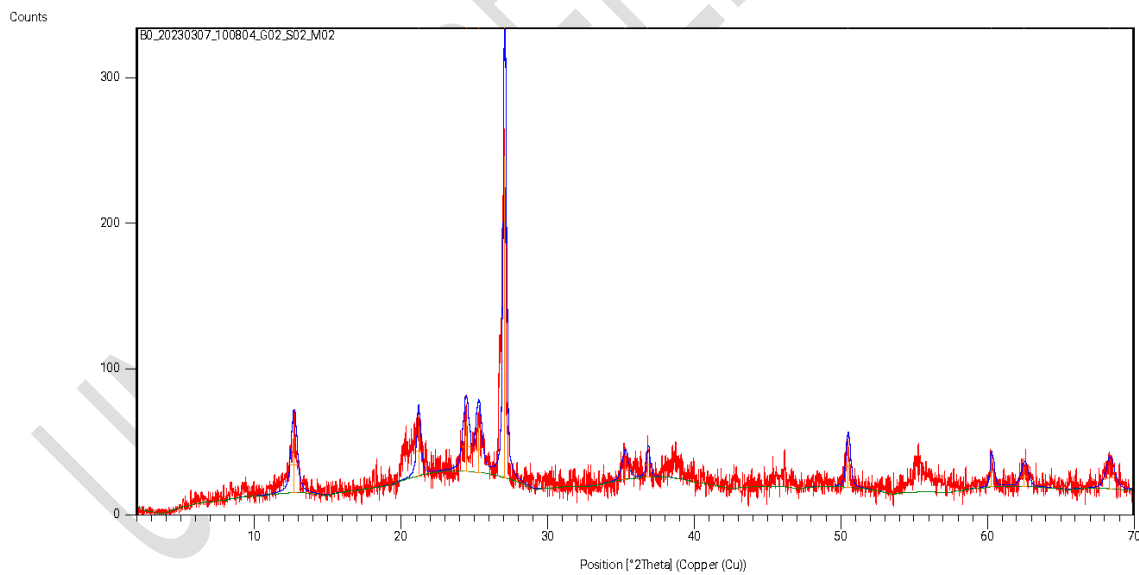


Fig. 5: XRD Pattern of EkimItam Untreated Kaolin

Plot of results

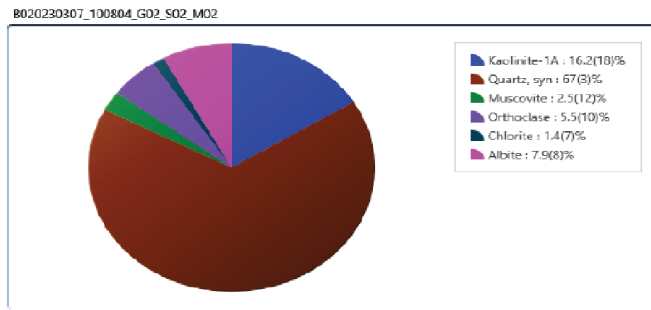


Table of results

Dataset / Weight Fraction...	Value, Unit	Kaolinite-1A	Quartz, syn	Muscovite	Orthoclase	Chlorite	Albite
B0_20230307_100804_G02...	0	16.2(18)	67(3)	2.5(12)	5.5(10)	1.4(7)	7.9(8)

Fig 6: Percentage quartz, Orthoclase, Muscovite, illite and Albite in untreated kaolin from Ikot EkimItam Calcined Kaolin

Phase Data View

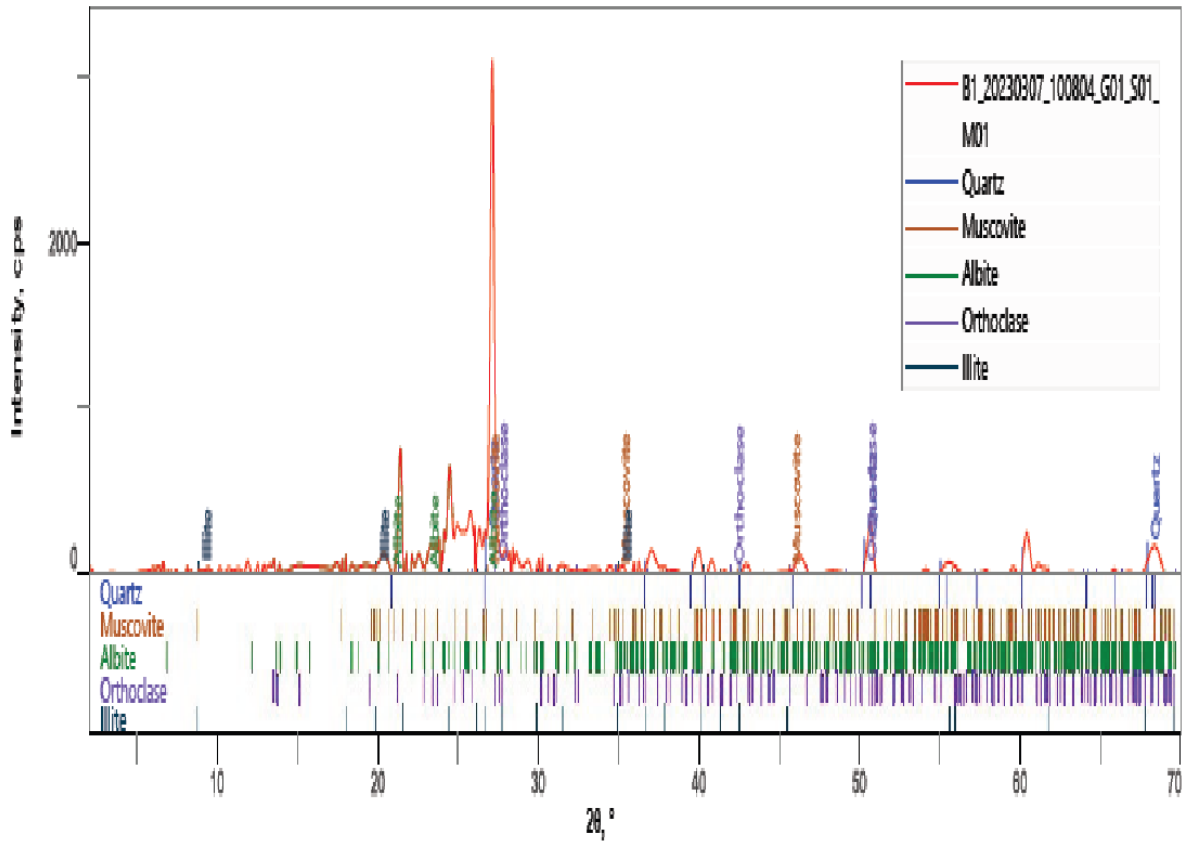


Fig. 7: XRD Pattern of EkimItam Calcined Kaolin

Plot of results

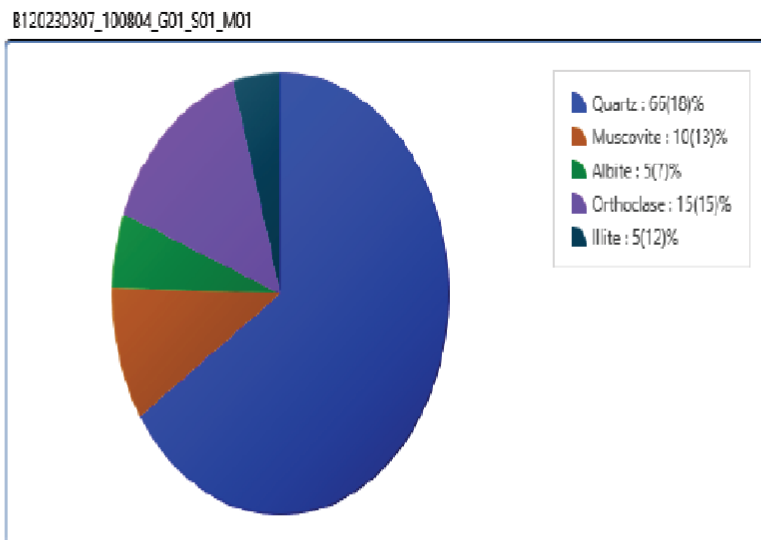


Table of results

Dataset / Weight Fraction, wt%	Value, Unit	Quartz	Muscovite	Albite	Orthoclase	Illite
B1_20230307_100804_G01_S01...	0	66(18)	10(13)	5(7)	15(15)	5(12)

Fig 8: Percentage quartz, Orthoclase, Muscovite, illite and Albite in calcined kaolin from Ikot EkimItam Calcined kaolin.

## Discussion

According to XRD data, the calcined kaolin from Ikot Uso Akpan Itam contained 68.3 % quartz, 17.3 % orthoclase, 5.9 % muscovite, 1.8 % illite, 6.68 %, albite and the natural (untreated) kaolin contained 48.0% quartz, 30.0% kaolinite, 3.0 % chlorite, 16.0 % illite and 2.0% albite while the calcined Kaolin from EkimItam contained 66.0 % quartz, 10.0 % muscovite, 15.0 % orthoclase, 5.0 % illite, 5.0 % albite and the natural (untreated) Kaolin contained 16.2 % kaolinite, 67.0% quartz, 2.5 % muscovite, 5.5 % orthoclase, 1.4 % chlorite, and 7.0 % albite, Fig. 1 - 8. The X-ray diffraction patterns of the untreated and calcined kaolin from the two kaolin deposits as presented in Fig 1 - 8 indicate sharp peaks implying that both samples are crystalline at natural and calcined stage. Konne, *et al* 2016 and Maciver, *et al* 2020 reported similar train for untreated and treated kaolin samples from Kono-Boue and Chokocho, Rivers State, Nigeria. The quartz was identified by its characteristics X-ray diffraction peaks at  $20.96^\circ$  and  $54.97^\circ$  2- theta. There was increase in composition of quartz in Ikot Uso Akpan kaolin from 48.0% to 68.3% after calcination at  $650^\circ\text{C}$  for 2 hours indicating higher crystallinity. Increase in calcination temperature increases activity and surface area of the

metakoalin linearly to a maximum temperature of 700 °C. Agglomeration of calcined kaolinite (metakoalinite) and formation of spinel phase occurs above 700 °C. The activation of kaolin should produce structural changes of its minerals composition hence promoting its activities (Gougazeh and Buhl, 2016). This investigation agrees with the work of (Adamu *et al.*, 2022, Udo *et al.*, 2023, Wang & Xie 2008). Calcination above 700 °C may result in disorderliness within the metakoalinite structure (Elimbiet *et al.*, 2011). Activated kaolin and zeolite can increase the productivity of bio-oil. In a related study by (Gandidiet *et al.*, 2018) kaolin proved a better ability to enhance the pyrolysis process with improved yield bio-oil.

The results of SEM micrographs of the untreated and the metakoalin clay samples from Ikot Uso Akpan are presented in Figure 9 and Figure 10 respectively.

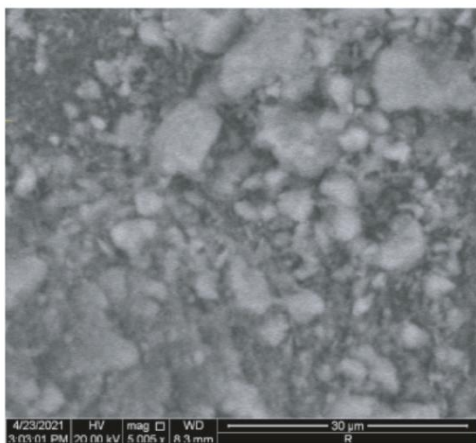


Figure 9: SEM Micrograph of Raw (Untreated) Ikot Uso Akpan Itam kaolin at 500x magnification

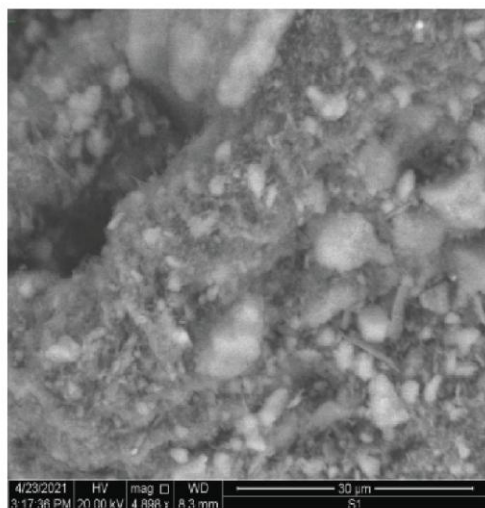


Figure 10: SEM Micrograph of Calcined Ikot Uso Akpan Itam kaolin at 500 x magnification

SEM micrographs of the untreated clay and the metakaolin clay samples are presented in Figure 9 and 10 respectively. Micrograph resolutions were taken at 30  $\mu\text{m}$ . Both the untreated and the calcined surface morphologies were heterogeneous size and spongy-like porous shape. There was no significant change in the surface morphology heterogeneous size of the calcined kaolin from the untreated kaolin. This reveals that thermal treatment at 650°C for 2 hours did not significantly modify surface morphology of the kaolin. In accordance with the results of other analytical methods quartz was identified by SEM in the sample. Cubic crystalline structures with well-defined edges are observed on the micrographs of the calcined kaolin Fig 10. This is in line with the work of other researchers (Yusuf *et al* 2019, Adamu *et al* 2022).

## Conclusion

Kaolin samples from Ikot Uso Akpan Itam and EkimItam, Itu local Government Area, Akwa Ibom State, Nigeria were successfully characterized using X-Ray Diffraction spectrophotometer technique to identify crystalline phase in the samples. This investigation has established the development of metakaolin through hydrothermal process and acid leaching. The characterization using XRD confirmed the presence of quartz and alumina mineral phases in from both kaolin deposits. It was also demonstrated that calcination of kaolin at 800 °C can increase its crystalline structure. Further work is required to synthesize zeolite catalyst from the kaolin for purification of biodiesel and cracking of hydrocarbons. Results of SEM indicated that the Ikot Uso Akpan Itam and EkimItam kaolin can be a promising raw material for zeolite production at appropriate processing conditions that can substitute imported zeolite. This will reduce cost, increase employment and revenue

## REFERENCE

- Atta, A., Ajayi O. and Adefila S. (2007). Synthesis of Faujosite Zeolites from Kankara Kaolin Clay. *Journal of Applied Science Research*, 3(10), 1017-1021.
- Atta, A., Jibril, B., Aderemi B. and Adefila, S. (2012). Preparation of Analcime from Local Kaolin and Rice Husk ash. *Applied Clay Science*, 61, 8-13.

- Baker, C.J. and Uren, R.E. (2010). Kaolin in New South Wales. *Geological Survey of South Wales*, 231.
- Bartholomew, H. (1994). *Catalytic hydroprocessing of petroleum and distillates*. In: Oballa M C, Shih S S, editors. New York: Marcel Dekker.
- Bassey, B. E., Opara, K. N. and Usip, L. P. (2020). Morphological and molecular characterization of endophilic *Anopheles gambiae* complex, in Itu Local Government Area, Akwa Ibom State, Nigeria *Nigerian Journal of Parasitology* 41(1): 70-78
- Bergaya, F. and Lagaly, G. (2006). Chapter 1 General Introduction: Clays. *Clay Minerals, and Clay Science, Hand B. Clay Sci.*, 4352(05)01001-9.
- Campanati, M., and Vaccari, A. (2001). *Fine chemicals through heterogeneous catalysis*. In: Sheldon R A, Bekkum H V, Herman H (Eds). Wiley-VCH, Weinheim, Germany, 61.
- Chandrasekhar S. (2006). Influence of meta kaolinization temperature on formation of zeolite 4A from kaolin. *Journal of Clay mineral*, 31 (2), 253-261.
- Choudhury, T. and Misra, N. (2011). Synthesis and characterization of zeolite Y and ZSM-5 from Nigerian Ahoko Kaolin using a novel, lower temperature, meta kaolinization technique. *Bull Mater Sci*, 34(6): 1273.
- Christidis, G. and Papantoni, H. (2008). Synthesis of FAU Type Zeolite Y from Natural Raw Materials. *The Open Mineralogy Journal*, 2, 1-5.
- Churchman, J. (2000). The definition and nomenclature of halloysites. *Clay Clay Miner.* 23 382-388,
- Cravero, E. and Dominguez, F. (1999). Origin of Sedimentary Kaolin in the Neuquen Basin, Argentina as Determined by Oxygen Isotopes. *Periodico di Mineralogia*, 68, 213-222.
- Dudasova, D., Simon, S., Hemmingsen, V., and Sjöblom J. (2008). *Colloids and Surfaces Physicochemical and Eng. Aspects*, 317, 1-9.
- Emam, A., (2013). Synthesis and characterization of zeolite Y and ZSM-5 from Nigerian Ahoko Kaolin using a novel, lower temperature, meta kaolinization technique. *American J. of Environmental Protection*, 2 (6), 161-169.
- Farag, H. and Zhang, J. (2012). Synthesis and characterization of zeolite Y and ZSM-5 from Nigerian Ahoko Kaolin using a novel, lower temperature, meta kaolinization technique. *Journal of Material Chemistry*, 19(34), 6207-6212.
- Elimbi, A., Tchakoute, H.K. Njopwouo, D. (2011) Effects of calcination temperature of kaolinite clays on the properties of geopolymer cements. *Construction and Building Materials* 25 (2011): 2805-2812

- Mousa Gougazeh and J.-Ch. Buhl (2014). Synthesis and characterization of zeolite A by hydrothermal transformation of natural Jordanian kaolin. *Journal of the Association of Arab Universities for Basic and Applied Sciences* (2014) 15, 35–42
- Gougazeh, M., Buhl, J.-C.h., 2010. Geochemical and Mineralogical Characterization of the Jabal Al-Harad Kaolin Deposit, Southern Jordan for its Possible Utilization. *Clay Miner. Mineral. Soc. Great Britain Ireland* 45 (4), 281–294.
- Gawande, P., and Kaware, J. (2016). International Research. *J. of Eng. and Technology*, 3 Issue 12, 337-340.
- Georgiev, D., Bogdanov, B., Markovska, I. and Hristov, Y. (2012). A Study on the Synthesis and Structure of Zeolite NaX. *Journal of Chemical Technology and Metallurgy*, 48(2), 168-173.
- Godwin J. Udo, Joachim J. Awaka-Ama, Emaime J. Uwanta, Ifiok O. Ekwere and Igwe R. Chibueze (2020). Comparative Analyses of Physicochemical Properties of Artisanal Refined Gasoline and Regular Automotive Gasoline. *Front. Chem.* 8(753):1-7,doi: 10.3389/fchem.2020.00753.
- Godwin J. Udo, Emmanuel E. Edemumo, Joachim J. Awaka-Ama, Emmanuel E. Ubuo, Emmanuel J. Ukpogon, Emaime J. Uwanta and Aniedi E. Nyong (2023). Synthesis and Characterization of Di-Metallic Zeolite Pt-Co Composite as Potential Catalyst for Reforming and Hydro-Desulphurization of Gasoline. *Mater. Sci. Res. Rev.*, 6(4):720-729
- Godwin J. Udo, Usoro M. Etesin, Joachim J. Awaka-Ama, Aniedi E. Nyong, Emaime J. Uwanta (2020). GCMS and FTIR Spectroscopy Characterization of Luffa Cylindrica Seed Oil and Biodiesel Produced from the oil. *Communication in Physical Sciences* 2020, 5(3): 378-390.
- Gandidi, I.M, Susila, M. D and Rustamaji, H. (2018). Effect of natural zeolite and kaolin as a catalyst in the isothermal-catalytic cracking of real municipal solid waste (MSW) for bio-oil production. *IOP Conf. Series: Earth and Environmental Science* 160 (012018):1-9 doi :10.1088/1755-1315/160/1/012018
- Haldar, K. (2020). Basic mineralogy. *Introd. To Mineral. Petrol., Elsevier*, pp. 109
- Haq, A., Iqbal, Y., and Khan, M. (2008). Historical development in the classification of kaolin subgroups. *J. Pakistan Mater. Soc.* 2 44–49.

- Hassan, M. (2014). Geochemistry and Origin of the Cretaceous Sedimentary Kaolin Deposits, Red Sea, Egypt. *Geochemistry*, 74, 195-203.
- Herrero, J., Blanco, C., Esteruelas, A., and Oro, A. (2000). Preparation of type 4A zeolite from Iraqi kaolin: Characterization and properties measurements. *Appl. Organometal Chem*, 4: 157.
- Hofmann, U., Endell, K., and Wilm, D. (2005). Röntgenographische und kolloidchemische Untersuchungen über Ton, *Angew. Chem.* 47 539–547.
- Hosseini, A., Niaei, A., and Salari, D. (2011). Preparation of type zeolite from kaolin: Characterization and properties measurements *Open Journal of Physical Chemistry*, 2011, 1: 23.
- Htay, M. and Oo, M. (2008). World Academy of Science. *Engineering and Technology*, 48: 114.
- Imerys, P. (2013). Calcined Clays in Light Weight Coatings Technical Guides. *Journal of Sciences and Technology*, 34 75-80.
- Jankovič, Ľ., Dimos, K., Bujdák, J., Koutselas, I., Madejová, J., Gournis, D., Karakassides, A., and Komadel, P. (2010). *Physical Chemistry Chemical Physics*, 12: 14236.
- Joussein, E., Petit, S., and Churchman, J., Theng, B., Righi, D., and Delvaux, B. (2005). Preparation of type 4A zeolite from Iraqi kaolin: Characterization and properties measurements *Halloysite clay minerals — a review, Clay Miner.* 40 383–426.
- Kerr, F. (1999). A decade of research on the nature of clay. *J. Am. Ceram. Soc.* 21 267–286.
- Kohyama, N. (1999). Observation of the hydrated form of tubular halloysite by an electron microscope equipped with an environmental cell. *Clay Miner.* 26 25–40.
- Kotal, M. and Bhowmick, A.K. (2015). Polymer Nano-Composites from Modified Clays: Recent Advances and Challenges. *Progress in Polymer Science*, 51, 127-187.
- Koukouzas, N. (2007). Ptolemais, Northern Greece. *Internal Journal of Coal Geology*, 71( 2-3): 276.
- Kovo, S. (2009). Synthesis and characterization of zeolite Y and ZSM-5 from Nigerian Ahoko Kaolin using a novel, lower temperature, meta kaolinization technique. *Journal of Material Chemistry*, 19(34), 6207-6212.
- Lima, P.A., Angélica, R. and Neves, R. (2017). Dissolution Kinetics of Amazonian Meta-Kaolin in Hydrochloric Acid. *Clay Minerals*, 52, 75-82.

- Liu, P., Farzana, R., Rajarao, R. and Sahajwalla, V. (2017). Lightweight Expanded Aggregates from the Mixture of Waste Automotive Plastics and Clay. *J. Construction and Building Materials*, 145, 283-291.
- MacEwan, D. (2010). The nomenclature of the halloysite minerals, *Mineral. Mag. J. Mineral Soc.* 28 36–44.
- Menzi, M., Hamzaoui A., Hamdi N. and Srasra E. (2011). Synthesis of Zeolites from the Low-Grade Tusisian Natural Illite by Two Different Methods. *Journal of Clay Science*, 52(3), 209-218.
- Mercier, Y. and Le, G. (2011). Rational ab initio modeling for low energy hydrogen bonded phyllosilicate polytypes, *Eur. J. Mineral* 23 401–407.
- Miller, W. (2013). Differentiation between endellite-halloysite and kaolinite by treatment with potassium acetate and ethylene glycol. *Clay Miner.* 10 244–253.
- Morsy, F., El-Sherbiny, E., Hassan, S. and Mohammed, H. (2014). Modification and evaluation of Egyptian kaolinite as pigment for paper coating. *Powder Technol.* 264 430–438.
- Murray, H. (2007). Applied clay mineralogy. 1<sup>st</sup> edition. *Journal of Clay Science*, 13 78-90.
- Murray, H.H. (2012). Kaolin Minerals: Their Genesis and Occurrences. In: Bailey, S.W., Ed., *Hydrous Phyllosilicates (Exclusive of Micas). Reviews in Mineralogy, Mineralogical Society of America, Washington DC, Vol. 19, 67-90.*
- Naman, A., Maher, K., and Ibrahim P. (2012). Applied clay mineralogy. *Journal of Petroleum and Gas Exploration Research*, 2(4): 80.
- Newnham, E. (1999). A refinement of the dickite structure and some remarks on polymorphism in kaolin minerals, *Mineral. Mag. J. Mineral Soc.* 32 683–704.
- Oghenejoboh, M. and Ohimsor, O. (2011). Applied clay mineralogy. *European J Sci Res*, 56(4): 445.
- Okada, K., Kawashima, H., Saito, Y., Hayashi, S., and Yasumori, A. (2011). Applied clay mineralogy. *J Mater Chem*, 5(8): 1244.
- Parker, S. (2010). McGraw-Hill Encyclopedia of the Geological Sciences. *Journal of Material Chemistry*, 33, 69-80.
- Saceda J., Rintramee K., Khabuanchalad S., Prayoonpokarach S., De R.L. and Wittayakun J. (2012). Properties of Zeolites as Promising Materials for Value Upgradation of Jatropha Oil. *Journal of Material Science*, 4(8), 7949-7959.

- Samadhi, T., Subagjo, W., Lismana, K., and Fuadi, K. ,(2011) Influence of meta kaolinization temperature on formation of zeolite 4A from kaolin. *Journal of applied Science*,43(2): 113.
- Schuman, T., Karlsson, A., Larsson, M. and Rigdahl, M. (2005). Characteristics of pigment-filled polymer coatings on paperboard, *Prog. Org. Coating* 54 360–371.
- Silvy, R. (2004). Applied clay mineralogy. *ApplCatal A*, 261:247.
- Speight, G. (2014). The Chemistry and Technology of Petroleum. *Journal of Science and Technology*, 48, 56-76.
- Speight, G. (2006). The chemistry and technology of petroleum. *Journal of Science and Technology*, 48, 56-76.
- Sun, H., Ma, T., Pang, M., Sun, B. and Wang, L. (2005). Natural Zeolite, Mineral and rock. *Bull CatalSoc of India*, 4: 72.
- Thomas, R. (2010). *High Temperature Processing of Kaolinitic Materials*. Univ. Birmingham Dr. Thesis., p. 4.
- Ugal, J., Hassan, K. and Ali, I. (2010). Preparation of type 4A zeolite from Iraqi kaolin: Characterization and properties measurements. *Journal of the Association of Arab Universities for Basic and Applied Sciences*, 9(1), 2-5
- Wang, P., Shen, B., Shen, D., Peng, T., and Gao, J. (2007). The active site of Acidic Aluminosilicate Catalyst. *Journal of Catalysis Communications*, 8: 1452.
- Wang, S., Li, X., and Zhai, Y. (2009). Characterization of kaolin. *Mater Lett*, 62(20).