

SEM & FTIR ANALYSIS OF RICE HUSK TO ASSESS THE IMPACT OF PHYSIOCHEMICAL PRETREATMENT

Abstract

Pretreatment is one of the pivotal processes in utilizing lignocellulosic biomass for producing bioethanol. An ecofriendly system only allows mild pretreatment strategies for industrial bioethanol production. The steam explosion pretreatment process is reported to be efficient **using rice** husk for these procedures with the use of mild acids or bases. In the current work, pretreatment method like steam explosion pretreatment method was used with NaOH and HNO₃ to degrade the complex structures and release the sugars entrapped within lignin. The pretreatment effect on the matrix of husk cell-wall and its constituents are characterized microscopically and spectroscopically by scanning electron microscopy and Fourier Transform Infrared Spectroscopy respectively, in order to comprehend the future possibility of its digestion by cellulase. The crystallinity index of native substrate is very high (0.94 cm⁻¹), which reduced significantly to -0.277 and -0.34 cm⁻¹ when pretreated with 2% HNO₃ and 10% HNO₃ respectively. The steam explosion pretreatment does not support the degradation of the cellulosic fibrillar arrangement, but causes intense re-localization of lignin. The descriptions of scanning electron microscopy were in agreement with the findings of Fourier Transform Infrared Spectroscopy; the ordered structure generally found in native rice husk was missing, suggesting that the structure of the 2% HNO₃ treated rice husk was more amorphous. The fractional removal of hemicelluloses and total removal of wax is the outcome of this research work. Results revealed that steam explosion pretreatment increases the possibility of digestion by enhancing cellulose accessibility through lignin re-localization and a partial elimination of hemicelluloses rather than by cell wall disruption.

Keywords: Bioethanol; rice husk; steam explosion; FTIR; SEM and crystallinity index

Introduction

Growing energy demand for industrial processes, heating and transportation is one of the prominent challenges of 21st century [1]. Substitute energy sources are being extensively investigated worldwide, as energy has always been in demand due to the advanced technologies and population [2]. Challenges like oil dependency are also compelling nations to extend their research for alternative fuels [3]. Potential gaseous biofuels are now available in the form of biohydrogen that can be generated out of waste in microbial electrolysis cells [4]. Bioethanol is a striking unconventional fuel being bio-based renewable resource. Moreover, it supports the reduction in particulate production being oxygenated [5]. Ethanol displays clean burning distinctiveness as its octane number is high and Reid vapor pressure is less [6].

Chemical, physical or biological modes of pretreatment are well explored methods for producing ethanol from lignocellulosic biomass (LB). The pretreatment not only augment the biodigestibility of the lignocelluloses for producing ethanol but also enhance accessibility of the enzymes to the biomass. The complex biodegradable materials get enriched by this process, thereby improving the ethanol yield from wastes [7]. Cellulose (35-50 wt.%, dry basis), hemicelluloses (15-30%), pectin (2-5%), and lignin (12-35%) are the chief constituents of LB. Cellulose and hemicelluloses constitute more than 50% of the entire biomass. Sugars generated out of these components can be converted to ethanol [8]. Pretreatment not only leads to the depolymerizing and solubilizing of hemicelluloses polymers, but also disrupts the matrix of cell wall as well as the association amid carbohydrates and lignin [9]. This ameliorates contact for the saccharifying enzymes and improves mass-transport restriction [8]. Degree of cellulose crystallinity also gets changes momentarily after pretreatment [10].

Alkaline pretreatment employs bases such as NaOH, KOH, NH₄OH, CaOH for the lignocellulosic biomass pretreatment. This process leads to the ester degradation and degradation of glycosidic side chains that consequences in structural alteration of lignin, swelling and fractional decrystallization of cellulose and limited solvation of hemicelluloses [11]. NaOH aids in disruption of lignin assembly of the biomass,

enhancing the approachability of enzymes to cellulose and hemicelluloses [12]. Acid pretreatment causes disjunction of the stringent configuration of the lignocellulosic biomass. Dilute nitric acid (HNO_3) is the most frequently used acid, which has been commercially employed to pretreat a broad spectrum of biomass [13].

Recently, Chhattisgarh Council of Science and Technology (CGCOST), Chhattisgarh, India, funded project on investigation of bio valorization ability of rice husk for bioethanol production has outcome in a steam explosion pretreatment procedure for rice husk that has verified to be efficient at formulating rice husk for enzymatic hydrolysis. Amongst the explored pretreatment methods, the steam explosion (SE) materializes one of the main ensuring approaches as it restricts the usage of chemicals frequently to the usage of saturated steam [14]. Steam explosions employs steam under pressure at the temperature ranging from 160° to 260°C for few minutes to disorganize the structure of lignocellulosic biomass [15].

Microscopic techniques such as scanning electron microscopy (SEM), has lined a pathway headed for an extensive comprehending of the fundamental molecular design of plant cell wall. Various other sophisticated procedures like Fourier transform infrared (FTIR) spectroscopy has also contributed in process upgradation [16]. In the current study, SEM explores of the effect of steam explosion pretreatment on rice husk cell wall disorder, composition, ultra-structure and exterior properties were studied in order better to comprehend the possibilities of its future digestion with cellulase. Chemical disintegration into component polymer classes was conceded out for all sample varieties. Qualitative determination of the structural alterations in rice husk post physiochemical pretreatment were done with the help of FTIR. FTIR is an analytical tool that executes a quick and non-persistent admittance of investigating biomolecules by spawning authentic and unrivaled spectral signatures crop up from diverse endogenic functional groups existing in the biomolecules. FTIR is a reagent-free procedure with minuscule procedure that involves very less sample to figure out the biochemical finger print [17].

Taking all the above facets into contemplation, the intend of the research is to obtain FTIR and SEM profile of native and pulverized rice husk in order to understand its

feasibility for further enzymatic digestion. FTIR spectroscopy and SEM were used to achieve the above aim as analytical means for qualitative determination of structural alteration in native and pretreated rice husk after physiochemical pretreatment.

Materials and Methods

Raw Material and its Collection

The substrate used is rice husk and it was collected from nearby rice mill from Bilaspur district (22.05°N 82.09°E/22.09°N 82.15°E) Chhattisgarh, India.

Processing of lignocellulosic substrates

The collected substrate was dried in a forced air oven at 50.0°C for 24 h and milled in a hammer mill to pass through a 1.30 mm screen. The milled substrate was conserved in sealed plastics bags at 4°C to avoid any potential degradation or spoilage.

Chemicals

Qualitative analytical reagents were procured from Sigma Aldrich Chemicals Pvt. Ltd., Bangalore. Organic solvents were purchased from Sisco Research Laboratories (SRL) and Qualigens Fine Chemicals, India.

Rice husk sample preparation for analyses

Diverse pretreatment procedures such as steam explosion at 160°C for 60 min; dilute nitric acid (2% v/v), concentrated nitric acid (10% v/v), sodium hydroxide (2% w/v), sodium hydroxide (10% w/v) with steam explosion at 160°C for about 20 min, were implemented separately for the pretreatment of rice husk. The pretreated husk was collected and filtered in crucibles followed by a wash with distilled water under suction. The remains left behind was thoroughly washed twice with distilled water to eliminate all associated chemicals and were air dried at 40°C for 48 h, after which they were used for FTIR and SEM analysis.

Fourier- transform infrared spectroscopic analysis

The structural attributes of polysaccharides sample were recorded on a Fourier-transform infrared spectrophotometer (IR Affinity- 1, Shimadzu, Japan). The samples were grounded with KBr powder (spectroscopic grade) and then pressed into 1 mm pellet for FTIR measurement in the frequency range $400\text{ cm}^{-1} - 4000\text{ cm}^{-1}$, with a spectral resolution of 0.5 cm^{-1} . The spectra would be achieved with an average of 64 scans. Analysis was executed on both the native and pretreated samples. The baselines of the spectra were regulated and normalized with the IRsolution software, and the absorption bands at 1427 cm^{-1} and 898 cm^{-1} were employed to calculate the crystallinity index¹⁸.

To assure that the surfaces measured were identical to those investigated by microscopy, the samples were not homogenized preceding to spectral analysis. The risk taken when selecting this procedure was that the surface cells of the native rice husk were not representative for the bulk material. In order to check whether this was the case, some untreated material was ground to a fine powder, and FTIR spectra were obtained from the homogenized material. There were insignificant differences found between these spectra and those from the non-homogenized samples.

Spectra were recorded from three different sub-samples per sample type, and all spectra were corrected according to the standard normal variate (SNV) method [19]. The mean spectrum of the three corrected spectra is presented for each sample type.

Scanning electron microscopy and Elemental X-ray Analysis

Scanning electron micrographs of untreated and treated rice husks was performed with a EVO MA10 (Carl Zeiss) Japan, operated at 20 kv. The samples were coated gold with a SC7620 manual high resolution sputter coater (Quorum Technologies, Newhaven, UK). The sample powder was sprinkled as a thin layer on an adhesive tape placed on the brass sample holder. Surplus amount of the sample was eliminated by blowing with the air spray. The adhered sample was then layered with gold powder employing the sputtering device.

Characteristic X-rays were detected by Oxford instruments attached with above mention SEM model, and further elemental X-ray analysis was done with the help of EDX software.

Results

FTIR spectroscopic analysis

FTIR spectroscopy was employed as an analytical means for qualitative determination of the chemical alterations on the surface of steam explosion pretreated rice husk separately with 2% and 10% each with HNO₃ and NaOH, to accompany and comprehends microscopic explorations. The FTIR spectra of above samples along with their native counterpart are shown in Fig. 1A.

The cellulose spectrum has three distinctive peaks at wave numbers of 1634, 1427 and 899 cm⁻¹. Small peaks or shoulders were there at wave numbers of 1367, 1319, 1337, 1284, 1203, 1161, 1119, 1114 and 999 cm⁻¹. Similarly, hemicellulose (Xylan) had major peaks at wave numbers of 1646, 1563, 1044 and 899 cm⁻¹ and small peaks or shoulders at wave numbers of 1508, 1461, 1420, 1252, 1212, 1164 and 990 cm⁻¹. The lignin spectrum demonstrated characteristic peaks at wave number of 1697, 1603, 1514 and 837 cm⁻¹. A few small peaks were observed at wave numbers of 1457, 1423, 1327, 1281, 1121 and 1034 cm⁻¹ [20].

Two interesting features are shown in Fig. 1.A (A2). First it can be seen that the carbonyl band at 1735 cm⁻¹, which has been attributed to hemicelluloses is diminished for all pretreated rice husk. This is anticipated as the pretreatment is known to eliminate a huge portion of hemicelluloses. Second, lignin bands at approximately in 1510 cm⁻¹ (aromatic ring stretch) are strongly enhanced in 10% NaOH pretreated samples of rice husk compared with that of native one, where these peaks are diminished [Fig. 1.A(A2)].

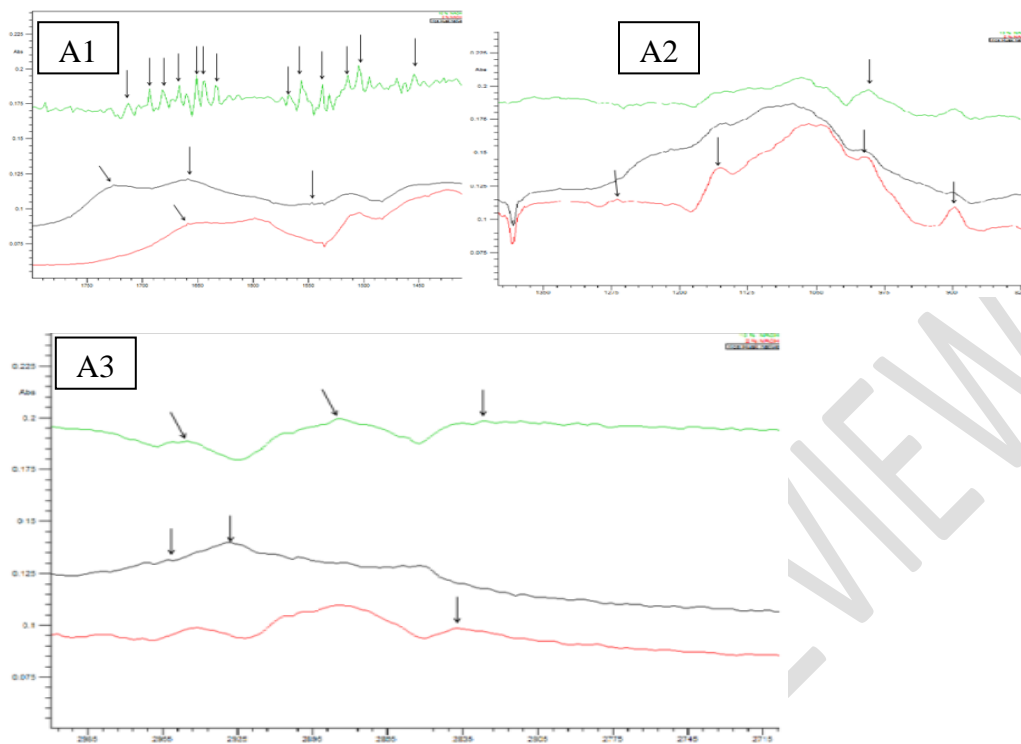
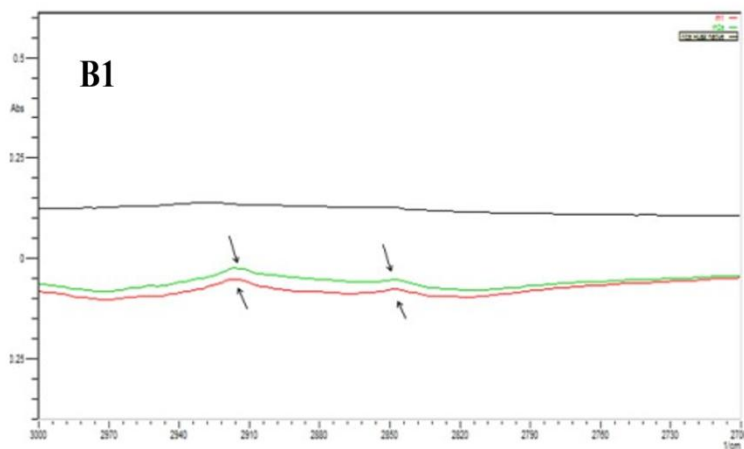


Fig. 1.A (A1-A3) - Overlapped view of FTIR spectra of native RH (black online), 2% NaOH pretreated RH (red online) and 10% NaOH pretreated RH (green online).



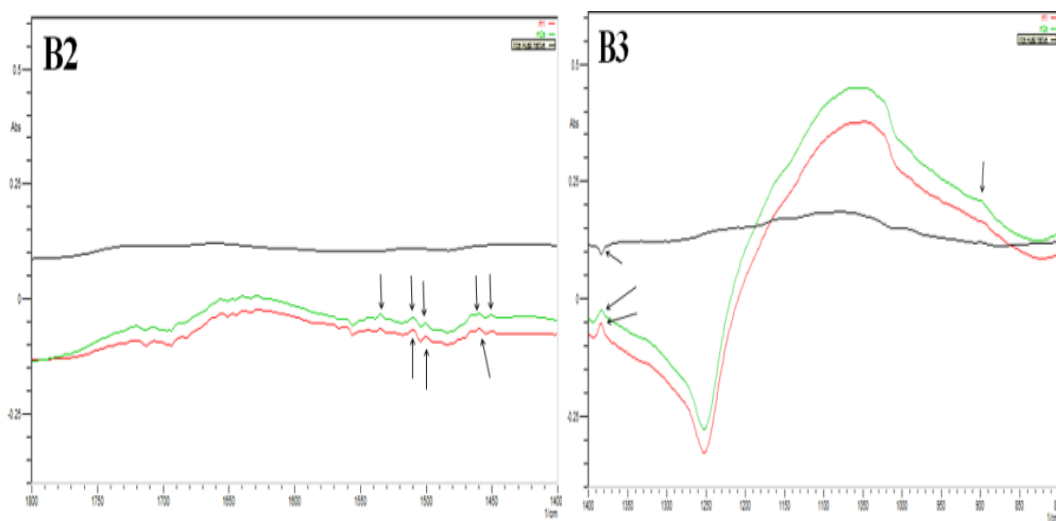


Fig.1.B (B1- B3) - Overlapped view of FTIR spectra of native RH (black online), 2% HNO₃ pretreated RH (red online) and 10% HNO₃ pretreated RH (green online).

Presence of various bands in FTIR spectrum of native and pretreated rice husk are given in Table 1. The CI of native rice husk is very high (0.94 cm⁻¹) while the pretreated substrate with 2% NaOH is 0.01, 10% NaOH is 1.02, 2% HNO₃ is -0.277, 10% HNO₃ is -0.34 cm⁻¹ (Fig. 2). This shows high difference.

Table 1 - Presence of various bands in FTIR spectrum of native and pretreated rice husk

Absorbance at	Native	2% NaOH	10% NaOH	2% HNO ₃	10% HNO ₃	Inferences
3600-3000	0.15	0.13	0.220	-0.003	0.07	This range comprises bands related to crystalline structure of cellulose. It is related to the valence vibration of

						H- bonded OH and intramolecular H-bonds
1463	-	-	+	+	+	CH ₂ + CH ₃ deformation of lignin
1733	+	-	+	-	-	Chemical changes in hemicellulose or lignin
1515	-	-	+	+	+	C=C aromatic skeletal vibration
1300-1000 (1058)	-	+	-	-	-	Penetration of chemical in the amorphous region of the biomass and degrading hemicellulases
1606	-	+	-	-	-	Lignin was also degraded by the action of HNO ₃ . Hence lignin is not composed of % phenyl And aromatic skeletal vibration

2850	-	-	-	+	+	Both are characteristics of cellulose
2918	-	-	-	+	+	Both are characteristics of cellulose
1164	-	+	-	-	-	peak of pure cellulose 1160 – glycosidic linkage
1010	-	+	+	-	-	
CI = 1427/898	0.94	.01	1.02	-0.277	-0.34	
1725	+	-	-	-	-	A simple carbonyl group such as ketone, ester
1545	+	-	-	-	-	C-C=C asymmetric stretching
1630-1660	+	+	-	-	-	Amide C=O stretching
1260	-	+	-	-	-	C-O stretching
900	-	+	-	-	+	anti-symmetric out-of-plane ring stretch of amorphous

						cellulose; C-O stretching
2840/2835	-	+	--	-	-	H-C-H asymmetric and symmetric stretching
2700-3000	-	-	+	-	-	Refers aliphatic compounds
1300-1600	-	-	+	-	-	Simple hydroxyl compound
1510,1460	-	-	-	+	+	semicircle ring stretching (aromatic lignin)
1380,1540	-	-	-	+	+	C-H symmetric and asymmetric deformation
1600	-	-	+	-	-	C-C=C asymmetric stretching, this band associated with lignin
1650	+	-	+	-	-	Due to C=O stretching
1460/1462	-	+	+	+	+	C-H deformation (methyl and methylene)

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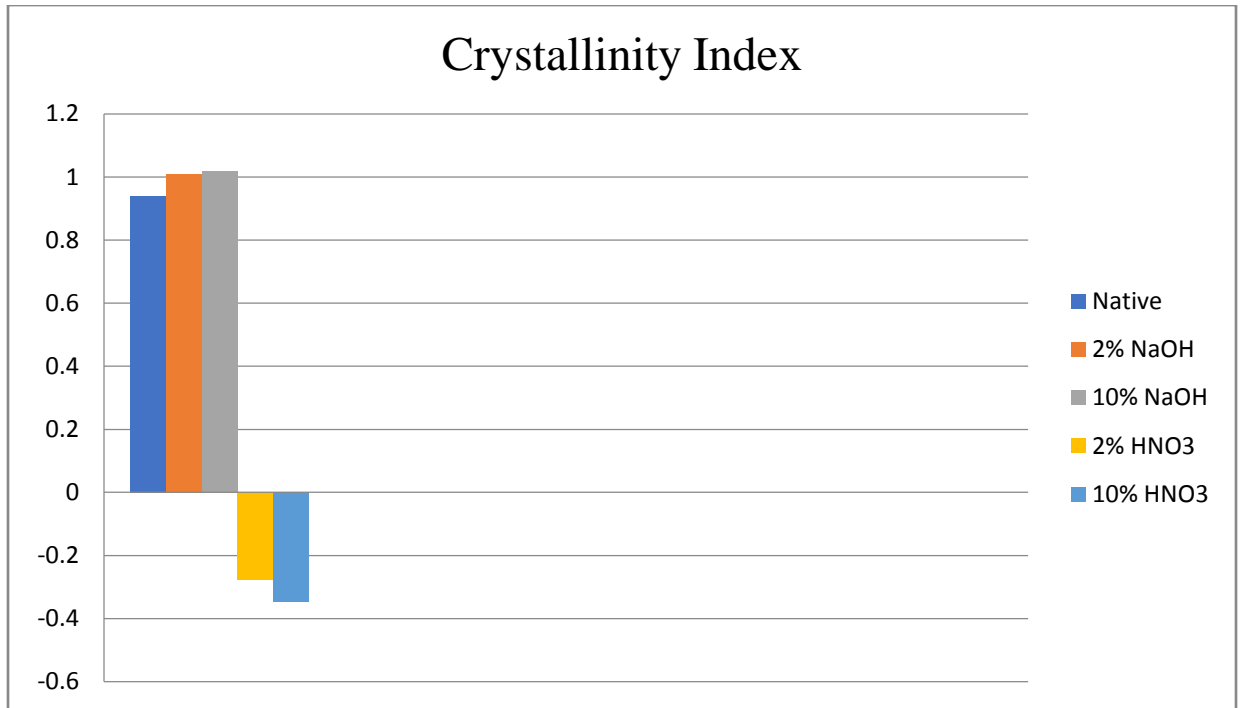


Fig. 2 - Total crystallinity index of native and pretreated rice husk

Scanning Electron Microscopy Analysis

Based on the outcomes from FTIR spectroscopy, scanning electron microscopy was employed to assemble information on the consequence of the steam explosion pretreatment on the ultra-structure and potential disorder of the cell wall. SEM technology was employed for morphological studies of both native and modified rice husk, so to achieve visualized imminent.

As it can be seen in Figures 3 (a & b), outer epidermis of rice husk, is well constructed and has a crumpled arrangement. The structure simulates that of a

amalgamated material with fibers (silica) properly interspaced in the matrix (cellulose, hemicellulose and lignin). The outer epidermis was extremely uneven and highlighted a linear edged conformation with dazzling dome-like constitutions, due to a more concentrated allotment of silica. However, a prior study pertaining field-emission scanning electron microscopy (FE-SEM) and energy dispersive X-ray micro-analysis (EDXA) machinery established that silica also emerged in other parts of rice husk with a comparatively low concentration.

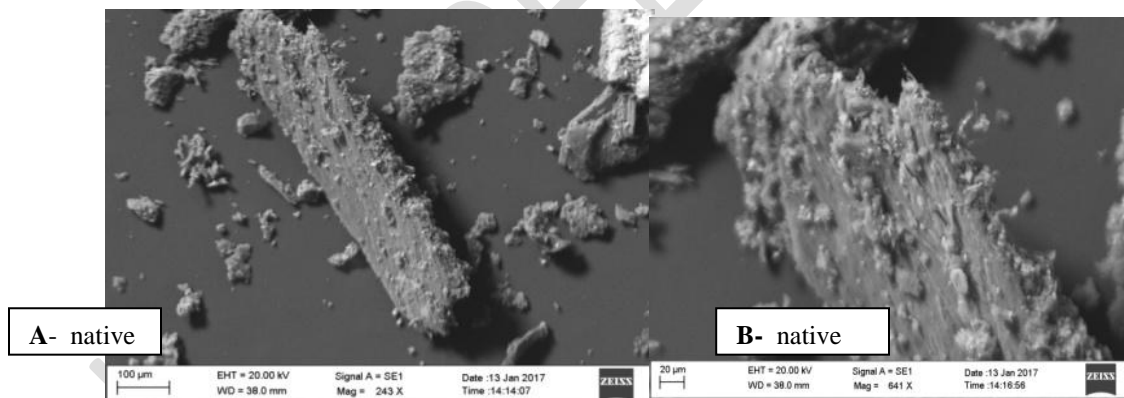
The silica is primarily confined to a small area in the rice husk's sturdy interlayer (epidermis) and filling in the spaces sandwiched between the epidermal cells. The concentration of silica was elevated on the exterior surface of the husk and much feeble on the inner face and virtually missing inside the rice husk. Figure 3 (a & b) shows the standard spherical platelets of approximately equivalent size (40–50 nm) emerging in equivalent rows.

Primarily, the most perceptible consequence of the steam explosion pretreatment distant from a color transform from golden brown into dark brown is the fractional defibration, or separation of individual fibers and cell types of the rice husk (Fig. 3c). As it can be seen in Fig. 3c, the exterior structure of the rice husk residues transformed considerably after 2% HNO₃ under steam explosion. Diminution of roughness on the outer surface and particle cracking, suggests weakening of rice husks due to hike in brittleness. Structural disruptions resulted in the improvement of accessibility of the interior parts of the cell wall matrix. Pits and perforations can be seen (Fig. 3c).

Post treatment by 2% HNO₃, the slight film (wax layer) on the exterior part vanished totally, the consistency was unstructured and fragmented, and some holes emerged on the solids surface. This signifies that the sample structure was dislocated by 2% HNO₃ treatment to a immense level. These alterations were valuable for the proficient biodegradability and the employment of rice husk for later cellulase digestibility.

Although the pretreated material is quite varied and holds larger pieces (up to about 1 cm) that are effortlessly identified as husk, a relevant fraction consists of cells that are either wholly or moderately detached from each other. All individual fibers seem

to be unbroken despite the steam explosion pretreatment, slightly than being damaged or otherwise dislocated (Fig. 3d). Close scrutiny of pretreated fibers reveals the presence of debris covering the surface and deposition of thin layer covering the entire surface (Fig. 3a). These fragments could be parts of middle lamellae. As it can be seen in fig 3d, the exterior micrographs demonstrate domes broken after the treatment step representing that at least part of inorganic fraction (mainly silica) was removed. As it can be seen in Fig. 3 (c-d), significant structural variations occurred on the overall or fibrillar organization of the individual fibers that includes perceptible augment of porosity. Treatment with 2% NaOH resulted in absolute elimination of the superficial surface. Samples appeared lighter, constituting chiefly cellulose with lower thermal constancy than the native one (Fig. 3e and Fig. 3a). As it can be seen in Fig. 3e, the native organization of rice husk was almost damaged and the inorganic portion (mainly silica) there as the conical projections on the native rice husks was eliminated. The removal of the outer surface and the disappearance of the inner surface also implicit the removal of the organic part (Fig. 3f).



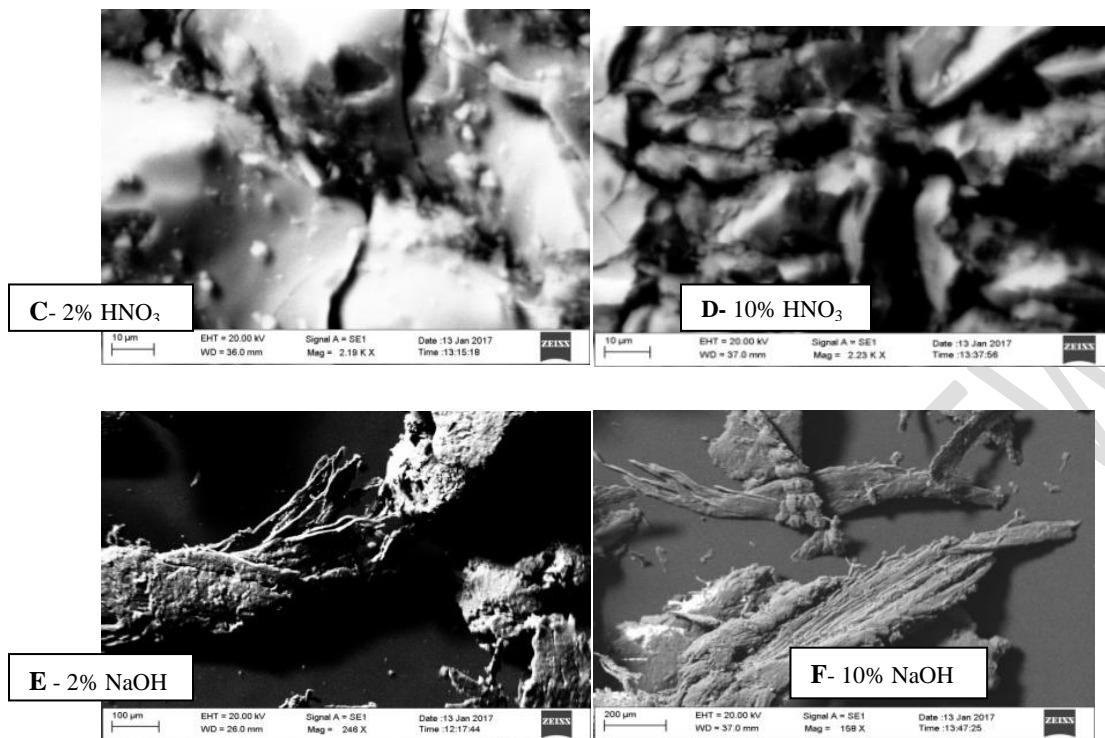


Fig.3 (A-F) - Scanning Electron Microscopic images of native rice husk and pre-treated (2% & 10% HNO₃/NaOH) rice husk.

Discussion

As it can be detected in the Fig. 1A(A1-A3), the untreated rice husk samples obtained various bands at 1650/1655 cm^{-1} which revealed a heterogeneity in the absorption intensity, conform either to carboxylates or the absorbed water; therefore, the difference in the absorption intensity probably expresses various cargo of carboxylates, as all samples were dried following the same protocol [21]. While the band at 1545 cm^{-1} is due to C-C=C asymmetric stretching and band at 1725 cm^{-1} the compound is apparently a simple carbonyl compound, such as a ketone, an aldehyde, an ester, or a carboxylic acid. The band at 1065 cm^{-1} anticipate the in-plane C-H bending vibrations of aromatic compounds symbolically arise in this region, and can endure as complex band structures (multiple, distinct bands). They gravitate not to be diagnostic for various compounds because of clash and convergence with other functional group absorptions, including

some skeletal (backbone) vibrations. Band at 1200 cm^{-1} speculate that the compound is likely to be a simple hydroxyl compound because the simple hydrogen-bonded OH absorption of a hydroxy (alcohol) function has a very distinctive shape. The band at 2925 cm^{-1} anticipates that they are aliphatic compounds [22]. The peak monitored after pretreatment with 2% NaOH were in range $1630\text{-}1690\text{ cm}^{-1}$ and are due to amide C=O stretching. The carbonyl stretching absorption is one of the robust IR absorptions, and is very useful in structure assurance (chem.ucla.edu/webspectra). The frequency range between $1200\text{-}1000\text{ cm}^{-1}$ has a large contribution of hemicelluloses and cellulose with maxima at 1037 cm^{-1} due to C-O stretching mode and 1164 cm^{-1} due to the asymmetrical stretching C-O-C [23, 24]. Adapa *et al.* [20], stated that the peak at 1164 cm^{-1} refers the existence of cellulose while Robert *et al.* [25], stated the presence of glycosidic linkage. Evident changes were also observed in the lignin-characteristic bands around 1606 cm^{-1} , and therefore, it is feasible to affirm that the lignin was also degraded by the action of the 2% NaOH [26]. Adapa *et al.* [20], also reported the presence of pure lignin and this peak also speaks for the aromatic skeletal vibration [27, 28]. The band monitored at 1260 cm^{-1} is due to the C-O stretching [22]. The peak scrutinizes at 2835 cm^{-1} was due to H-C-H asymmetric & symmetric stretch. Bekiaris *et al.* [21], stated that the band at $900/898\text{ cm}^{-1}$ can be authorized to amorphous cellulose while band at 895 cm^{-1} to $\beta\text{-}1\text{-}4$ linkage [25] and band at 900 cm^{-1} speak for the anti-symmetric out-of-plane ring stretch of amorphous cellulose [29]. Band at 1000 cm^{-1} characterize the pure cellulose [20].

The band observed after pretreatment with 10% NaOH was focused around 1600 and 1500 cm^{-1} , usually crop up as a pair of band structures, often with some crumbling. The display and ratio of these band structures is firmly reliant on the position and nature of substituent on the ring [22]. The absorption in 1733 cm^{-1} is by virtue to a C=O unconjugated stretching of hemicelluloses but also with the allowance of lignin. Absorption around 1733 cm^{-1} specifies chemical changes in hemicellulose and/or lignin. The absorption around 1463 cm^{-1} reveals to CH_2 and CH_3 deformation of lignin [26]. Absorption around 1515 cm^{-1} is linked with C=C aromatic skeletal vibration [27]. The frequency spectrum between $1200\text{-}1000\text{ cm}^{-1}$ has a large input of hemicelluloses and

cellulose with maxima at 1037 cm^{-1} due to C-O stretching mode [23]. The band observed between $1300\text{-}1600\text{ cm}^{-1}$ is likely to be simple hydroxy compound. Coates [22] stated that the band between $2700\text{-}3000\text{ cm}^{-1}$ compound is apparently aliphatic. If the main absorptions are approximately 2935 and 2860 cm^{-1} , and there are also absorptions at 1470 and 720 cm^{-1} , then the compound presumably contains a long linear aliphatic chain.

As it can be seen in the Fig. 1B(B1- B3), the band interpretation of pretreated substrates with 2% HNO_3 observed bands at 1500 cm^{-1} , 1510 cm^{-1} is due to aromatic C=C bending (webspectra.ucla.edu) and are associated with lignin [21] while band at 1460 cm^{-1} is due to N=H bending as well as refers to CH_2 and CH_3 deformation of lignin [26]. Absorption around 1515 cm^{-1} is associated with C=C aromatic skeletal vibration [26]. Band at 1360 cm^{-1} assumes that a simple hydroxyl compound because the simple hydrogen-bonded OH absorption of a hydroxy (alcohol) functions has a very characteristic shape [22]. The small peaks at 2850 cm^{-1} and 2918 cm^{-1} originates from CH_2 and CH_3 symmetric and asymmetric stretching respectively. Both are characteristic of cellulose [23, 30]. The peak at 2920 cm^{-1} and the shoulder at 2850 cm^{-1} correspond to aliphatic. Ciolacu *et al.* [31], observed a variation in this peak from 2900 cm^{-1} for pure cellulose to 2920 cm^{-1} for the amorphous cellulose. The band observed after 10% HNO_3 pretreated substrates were 1380 , 1450 , 1465 , 1500 , 1510 , 1535 and 1580 cm^{-1} which assumed that it is that a simple hydroxyl compound because the simple hydrogen-bonded OH absorption of a hydroxy (alcohol) function has a very distinctive shape ($1300\text{-}1600\text{ cm}^{-1}$) [22]. Peak of 1510 cm^{-1} is due to aromatic C=C bending (webspectra.ucla.edu) and are associated with lignin while 1460 cm^{-1} , tally to xylans. The band at $900/898\text{ cm}^{-1}$ can be accredited to amorphous cellulose [21]. The small peaks at 2850 cm^{-1} and 2918 cm^{-1} originates from CH_2 and CH_3 symmetric and asymmetric stretching respectively. Both are peculiar of cellulose [23, 30].

Two interesting features are shown in Fig. 1.A (A2). First it can be seen that the carbonyl band at 1735 cm^{-1} , which has been attributed to hemicelluloses is diminished for all pretreated rice husk. This is anticipated as the pretreatment is known to eliminate a huge portion of hemicelluloses. Second, lignin bands at approximately in 1510 cm^{-1}

(aromatic ring stretch) are strongly enhanced in 10% NaOH pretreated samples of rice husk compared with that of native one, where these peaks are diminished [Fig. 1.A(A2)]. One elucidation for this could be a comparative hike in the lignin amount due to the elimination of hemicelluloses. Another basis could be release of lignin and its re-deposition on the surface. The lignin hike is considered too considerable to be only due to the hemicelluloses exclusion [32].

Reduction in cellulose crystallinity is one of the best strategies engaged to enhance the enzymatic convertibility. Infrared peak ratios describe the comparative amounts of amorphous and crystalline cellulose which differentiates the samples. The crystallinity variation can be studied by accrediting the 1427 and 898 cm^{-1} absorption bands with respective to cellulose I and cellulose II. The absorbance ratio A_{1427}/A_{898} is acknowledged as crystallinity index (CI). Crystallinity of cellulose is one of the crucial factors influencing enzymatic hydrolysis. The lower crystallinity index signifies a higher quantity of amorphous cellulose present in the regenerated cellulose [33].

Silica throughout composition augmented the strength of the rice husk epidermis [34]. Park *et al.* [35], have revealed from field-emission SEM (FE-SEM) and energy dispersive X-ray micro-analysis (EDXA) trials, that silica appears to be there all through the outer face of rice husk (domes and their shoulder). The elevated silica substance on outer epidermis offers tenacity and rigidity to the husk.

Figure 3 (a & b) shows the standard spherical platelets of approximately equivalent size (40–50 μm) emerging in equivalent rows. As reported, the majority of the silica prevailed in the superficial epidermal cells, being predominantly cluttered in the dome-formed protuberance [36].

The surface of rice husk residues emerged to be rough and had cracks. The crystallinity index of 2% HNO_3 treated rice husk is -0.2777. This illustrates that the disintegration and consecutive reclamation of cellulose is dependent on the extent of swelling on the biomass. The disintegration procedure was presage by considerable swelling of the rice husk matrix which could be monitored from the swollen display of the rice husk residue. This rice husk residue illustrated meticulously disordered structure

with diminished crystallinity. Rice husk residues from steam explosion pretreatment could be a prospective substrate for bioconversion into value compounds. The disrupted surface configuration of the rice husk residues makes them complimentary for solid state fermentation, where it aids in growth of microorganism by permitting entry of microorganism to the lignocellulosic matrix [33].

The images of SEM were in conformity with the conclusion of FT-IR; the ordered structure frequently present in indigenous lignocellulosic biomass was missing, suggestive of that the structure of the 2% HNO₃ treated rice husk was more amorphous. This also indirectly specifies that, with 2% HNO₃ pretreatment, results in reduction of crystallinity of cellulose in comparison to the native rice husk.

As it can be seen in Fig. 3 (c-d), significant structural variations occurred on the overall or fibrillar organization of the individual fibers that includes perceptible augment of porosity. These variations are considered to be correlated with thermal pretreatments. Holes or cracks were also observed in the fibers.

The complete vanishing of 1732 cm⁻¹ absorbance was revealed in the FTIR spectrum, signifying the absence of hemicelluloses and waxes in this material. Presence of peak at 1606 cm⁻¹ in the FTIR spectra with diminished intensity indicates the existence of un-extracted aromatic compounds [37]. The chief consequences of alkali pretreatment include the reduction in content of lignin and hemicellulose in natural fibers, permitting diffusion of water molecules to the inner layers and cleavage of bonds joining lignin-carbohydrate and hemicelluloses. Along with the removal of hemicelluloses, alkali treatment develops the fiber surface bond distinctiveness and generating rough surface. This type of surface presents better fiber matrix interface bond and an augment in mechanical properties [38].

The light color of the sample indicated that the consequential rice husk after pretreatment is mostly cellulose, as the cementing matter of the lignocellulosic matrix i.e. lignin and hemicelluloses were preferentially degraded. The dark color of raw rice husk is basically due to lignin concentration [34].

The SEM micrographs clearly indicate the significant changes in the morphology of rice husk when the later was treated in alkaline conditions. Both the inner and outer surface of rice husk were affected as an outcome of elimination of silica (inorganic component), hemicellulose and lignin (organic component). The resultant rice husk presented an intense color and a more consistent fiber allocation [34]. The skeletal arrangement is integral with insignificant changes in cellulose crystallinity. Consequently, the pretreatment is effective due to removal of hemicellulose and lignin re-localization. Pretreatment does not support complete removal of lignin leading to fruitless cellulases adsorption, as lignin enclose the cellulose in the cell-wall matrix, obstructing cellulases from accessing cellulose fibrils.

Conclusion

Attention was made among global researchers and academicians towards the bioconversion of rice straw and husk for production of bioethanol being source for sustainable and green fuel. Having bestowed in cellulosic biomass with great interest in converting them to bioethanol, rice husk is used as global source for biorefinery purpose. However, due to much recalcitrant than rice straw and producing many fermentation inhibitors efficient treatment processes are to be explored towards healthy saccharification and fermentation. **These studies ensure the accessibility of substrate to the future exposure with cellulase which will further help in saccharification by releasing cellulosic glucose.**

Morphological alterations of straw by pretreatments were examined via SEM images. Evaluating SEM descriptions of treated and untreated rice husk illustrates remarkable alterations in the surface configuration and porosity. Untreated rice husk had a filled structure covered by a silica layer. However, the silica layer was unwrapped as an outcome of treatment, leading to amplified reachable surface area for the enzyme penetration. Samples treated with HNO₃ and NaOH had a demolished structure and the swelled shape after steam explosion. Hence the SEM examination proved that the rice husk morphology altered significantly post being treated in alkaline conditions. Both the inner and outer surface of rice husk were affected as an outcome of elimination of silica

(inorganic component), hemicellulose and lignin (organic component) and the diminution in cellulosic crystallization. This pretreatment ensures the further utility of rice husk to harness left over cellulosic sugar (glucose) by cellulase treatment.

Declarations

Ethical Approval and Consent to participate - All authors have consent for participation and there are no ethical issues associated with this manuscript.

Consent for publication – All authors have consent for the publication of this manuscript.

Availability of supporting data- All data are provided in the manuscript.

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