

1 Original Research Article
2 Climatic factors affecting water quality under
3 natural conditions: a field survey of a local reservoir

26
27
28

ABSTRACT

The global water cycle is closely related to climate change, and fresh water is a precious resource. Therefore, the effects of climate change on fresh water quality are of major importance. Worldwide, shallow lakes and ponds are the most abundant reservoir types. However, there have been few studies about ponds despite their large number. It is commonly accepted that wind-driven currents and thermal stratification mainly affect water circulation and oxygen diffusion in lakes. The presented research aims to verify whether this accepted view would be observed in a pond (~ 1 m depth and $\sim 5,600$ m² area) under natural conditions accompanying changes in temperature and wind. A field survey performed over 7 months in Japan has demonstrated that (i) the temperature variations in the air and the pond water were negatively correlated with the dissolved oxygen concentration; and (ii) the wind variation shows weak negative correlation with the dissolved oxygen level in the bottom layer. A simple concept of the link between temperature and dissolved oxygen is established through these findings – the oxygen solubility dependent on temperature is important rather than thermal stratification and wind in terms of discussing the climate change effects on pond water quality.

29
30
31
32
33
34
35
36

Keywords: Dissolved oxygen, Pond, Shallow lake, Temperature, Wind

1. INTRODUCTION

Global climate change is often thought of as something that will happen in the future, but it is an ongoing process [1]. Changes in Earth's climate, driven by increased greenhouse gases, are already having widespread effects on the environment such as droughts, wildfires, and

37 torrential downpours [2]. The global water cycle is closely related to climate change, land
 38 use and the environment through complex interactions [3] — water is fundamental to all life
 39 on Earth; for example, water makes up about 70% of the human body by weight, so a loss of
 40 only 4% of water from the body leads to dehydration and a loss of 15% results in death [4 &
 41 5].

42 Regarding the sources of water indispensable to life, the main ones include fresh surface
 43 water, which accounts for just 1/10,000 of the total water available on the planet [6], and on
 44 a global scale the amount of fresh surface water is almost constant over time, being
 45 replenished by water precipitation previously evaporated from the ocean ($\sim 350,000 \text{ km}^3$) and
 46 land areas ($\sim 70,000 \text{ km}^3$). Most precipitation falls back into the ocean, and only $\sim 110,000$
 47 km^3 falls on the land; that means, fresh water is a scarce resource. About 4 billion people
 48 currently experience water shortages for at least one month of the year [7], and these water
 49 crises place third in a list of impact risks [8]. Modification in the distribution of groundwater
 50 recharge and river flows over space and time are determined by changes in temperature,
 51 evaporation and, mainly, precipitation [9]. Therefore, the effects of climate change on fresh
 52 water quality seem to be an urgent issue that confronts not only human beings but also
 53 many other species. This paper attempts to elucidate the interaction between climate
 54 change and water quality on the basis of a field survey of a local reservoir, to assess
 55 dissolved oxygen differences in ponds.

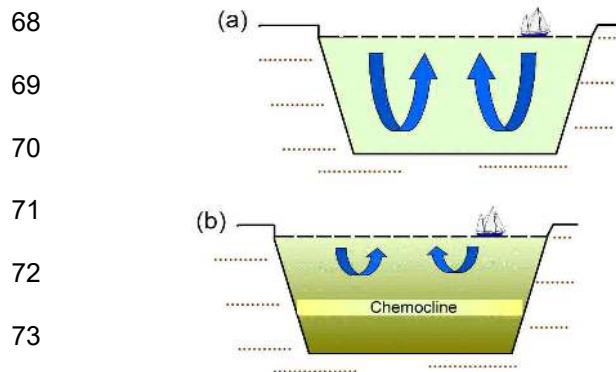
56 2. BASIC INFORMATION

57 Multidisciplinary knowledge may be required to understand the relation between climate
 58 change and water quality in reservoirs such as lakes and ponds. The overall aim of this
 59 paper is to present a comprehensive case study on the relation of water quality with climate
 60 change; therefore, basic information about key factors is briefly reviewed first, followed by a
 61 description of the main discussion.

62 2.1 Holomictic and meromictic

63
 64 A lake can be basically classified as holomictic or meromictic in terms of mixing [10].
 65 Holomictic lakes follow a seasonal cycle of stratification and complete mixing (Fig. 1a), but
 66 meromixis is a condition in which a lake does not mix completely [11] (Fig. 1b).

67



75 **Fig. 1. Conceptual pattern of water circulation in lakes (redrawn from [12 & 13]):** (a)
 76 *holomictic lake — physical circulation (i.e. mixing) occurs between the surface and the deep waters;*

77 *and (b) meromictic lake — circulation is possible only within a layer, so turnovers from top to bottom do*
 78 *not occur. A chemocline (i.e. transition zone) is commonly formed and separates the upper and lower*
 79 *layers.*

80 In holomictic lakes, the water body circulates at least once a year due to homothermal
 81 conditions, and mixing is complete or partial. The circulation homogenizes oxygen and
 82 nutrient concentrations throughout the water mass [14].

83 In meromictic lakes, the lack of circulation between layers creates radically different
 84 environments for organisms to live in: among the consequences of this stratification, or
 85 stable layering, of lake waters is that the bottom layer receives little oxygen from the
 86 atmosphere, hence becoming depleted of oxygen. While the surface layer may have 10 mg/l
 87 or more dissolved oxygen in summer, the depths of a meromictic lake can have less than 1
 88 mg/l [15].

89 Most lakes on Earth are holomictic, whereas meromictic lakes are rare [14]. However, in the
 90 case of Lake Biwa (the largest lake in Japan, see also Fig. 2) having about 670 km²
 91 surface area and 41 m of mean depth, for the first time in recorded history, full circulation
 92 was not observed in 2018, and this phenomenon continued in 2019 [12]. A warmer than
 93 usual temperature continued at that time [12]. Hence it can be considered that the lake
 94 turnover did not occur completely because the water density did not change due to the
 95 insufficient function of temperature. As Lake Biwa is a main source of drinking water for 14
 96 million people in the Kansai region of Japan [16], the quality degradation of the lake water
 97 became a potential threat to the neighboring population, agricultural irrigation and regional
 98 ecosystems [17, 18 & 19].

99

100 **2.2. Shallow lakes and ponds**

101

102 Worldwide, shallow lakes and ponds are the most abundant water reservoirs on land, and
 103 these reservoirs supply lots of ecosystem services, goods and materials [20]. Many shallow
 104 lakes and ponds have been created by humans after millennia of landscape modification,
 105 such as stream and river impoundment. It is considered that the historical undercounting of
 106 small lakes and/or ponds has led to a significant underestimation of the world's lake and
 107 pond area [20].

108 As stated above, there are many shallow lakes and ponds throughout the world. They are
 109 usually wind-exposed [11]. After thermal stratification develops during the daytime, full
 110 circulation takes place at night due to wind-induced mixing and convection from surface
 111 cooling [21]. Regarding the diurnal mixed layer in a shallow lake, the relative buoyancy ($\epsilon \cdot g$)
 112 caused by solar radiation can be expressed as follows (cf. [21 & 22]):

113

114
$$\epsilon \cdot g = (\Delta\rho/\rho_0) \cdot g = \epsilon_0 \cdot g \cdot \exp(-z/H) \text{ — (I)}$$

115

116 where ρ_0 is the reference density of water, $\Delta\rho$ is the density variation caused by solar
 117 radiation, g is the gravitational acceleration, ϵ_0 is the value ϵ in the water surface, z is the
 118 water depth, and H is the center of buoyancy.

119 Assuming that the wind blows at the point where the relative buoyancy is formed, the
 120 integrated value of buoyancy B within the interval $z=0$ to $z=h$ is denoted as follows:

121

122
$$B = \int \epsilon_0 \cdot g \cdot \exp(-z/H) dz = \epsilon_0 \cdot g \cdot H \cdot [1 - \exp(-h/H)] \text{ — (II)}$$

123

124 The value H is $h/2$ where the cline (e.g. a layer in which the water property varies) exists in
 125 $z=h$, and the increment of potential energy P (cf. turbulent kinetic energy) represented by the
 126 value B is expressed by the following equation:

127

128
$$P = B \cdot (h/2 - z_0) = \epsilon_0 \cdot g \cdot H \cdot h \cdot [\frac{1}{2} \{1 + \exp(-h/H)\}] - H/h \{1 - \exp(-h/H)\} \text{ — (III)}$$

129

130

3. FIELD SURVEY

131

132

133

134

135

136

137

3.1. Survey area

138

139

140

141

142

143

144

145

146

147

148

149

150

151

152

153

154

155

156

157

158

159

160

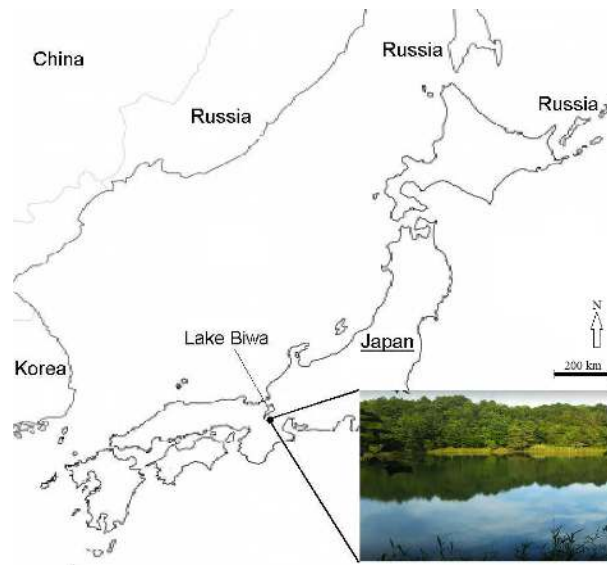
161

162

163

164

165



166

167

Fig. 2. Location of the study area (34°97'N and 135°93'E). *The picture shows a view of Nagao Pond.*

168

169

170

171

172

173

174

175

176

177

178

A typical pond was selected as the research target. The study focuses on Nagao Pond (Fig. 2), a kind of artificial water reservoir located on the north side (34°97'N and 135°93'E) of Lake Biwa (section 2.1) in western Japan. The area encompasses about 5,600 m² and the mean depth is about 1.0 m. The survey region belongs to a temperate zone with four distinct seasons.

According to meteorological data in the research region [25 & 26], the snow period lasts for 1.5 months from January to mid-February, and the month with the most snow is February with an average snowfall of 30 mm/month; the period of the summer season is about 3 months from mid-June to mid-September, and the maximum temperature is 26°C and the average diurnal difference between the maximum and the minimum temperatures is 10°C; the period of the winter season lasts for about 3.5 months from December to mid-March,

179 with average monthly maximum and minimum temperatures of 12°C and 0°C, respectively.
180 Wind speed and the predominant wind direction vary throughout the year – winds often blow
181 from the north at an average speed of 1.8 m/s during February-June and September-
182 November, from the south at an average speed of 1.6 m/s during July-August, and from the
183 west at an average speed of 2.2 m/s during December-January.

184

185 **3.2. Materials and methods**

186

187 The spot information on wind speed and direction was gathered through a local climate
188 database by JMA [25], and the procedures used in this survey are summarized below.

189

190 **3.2.1. Water sampling**

191

192 Five points were chosen for water sampling, including the center section: the northeast
193 section, the northwest section, the southeast section, the southwest section, and the center
194 of Nagao Pond. Using a Heyroth sampler, 250 ml of water were collected from the surface
195 layer and the bottom layer of each point, respectively. The sampling was carried out in the
196 early afternoon once every 2 weeks, and two samples per point were collected.

197

198 **3.2.2. Measurements and analysis**

199

200 The following parameters were measured in the field: (i) the water depth was measured
201 using an ultrasonic echo sounder (Hondex PS7); (ii) the surface water temperature and the
202 deep water temperature were measured using a digital multimeter (Hanna HI98129N); and
203 (iii) the concentration of dissolved oxygen was determined using a digital oxygen meter with
204 polarographic probe (Lutro PDO-520). In addition, a portion of each sample was promptly
205 transported to a laboratory for the quantification of (iv) total phosphorus (TP) through
206 molybdate colorimetry at 880 nm wavelength (cf. Japan Industrial Standard K0102 46.1.1);
207 and (v
208) ammonia nitrogen ($\text{NH}_4\text{-N}$) by indophenol blue colorimetry at 630 nm wavelength (cf. Japan
209 Industrial Standard K0102 42.6).

210

211 **3.2.3. Data processing**

212

213 Using numerical analysis with R software, the correlation coefficient was applied to evaluate
214 the relation between the temperature and the other parameters. The correlation coefficients
215 interpreted in this paper are as follows: $0 \leq |r| < 0.2$ indicates little or no association; $0.2 \leq |r| < 0.4$
216 indicates weak association; $0.4 \leq |r| < 0.7$ indicates moderate association; and $0.7 \leq |r| \leq 1.0$
217 indicates strong association

218

219 **4. RESULTS AND DISCUSSION**

220

221 The data taken in this survey are summarized in Table 1. All data are shown as average
222 values, which are considered to be representative. Variations in parentheses represent the
223 minimum and maximum values measured.

224

225 **4.1. Water depth**

226

227 There is a clear difference between the least (0.4 m) and greatest values (1.0 m) of water
228 depth, but the average depth varied slightly from 0.7 m to 0.8 m (Table 1). Viewing the
229 detailed data for each survey point, the variations in water depth were similar (0.78 ± 0.05 m)
at all sampling points except the northeast point; on the other hand, the water depth varied

230 from 0.4 m to 0.5 m at the northeast point; generally, the water depth at the northeast point
 231 is about one half of that at the other points.
 232 When considering all five sampling sites, the dissolved oxygen level does not vary
 233 statistically from the surface layer to the bottom layer ($p > 0.05$). Considering that dissolved
 234 oxygen enters water through the air or as a plant byproduct (review in [27]), the following
 235 hypothesis is proposed to interpret this specific phenomenon: oxygen originating in the air
 236 can diffuse across the water's surface and be naturally mixed in a shallow pond, that is, the
 237 diffused oxygen easily reaches equilibrium to some extent throughout the whole pond; and
 238 even if a large portion of photosynthesis (e.g. phytoplankton) takes place underwater,
 239 sunlight can penetrate shallow water and reach the bottom, meaning that the phytoplankton
 240 is not likely to suppress the photosynthesis linked with oxygen supply.
 241 Although Nagao Pond is man-made, it has an uneven bottom contour – i.e. deep and
 242 shallow parts. It is reported that there are about 160 thousand artificial ponds and these
 243 ponds are generally characterized by a depth of below 1.0 m (cf. section 3.1). Doubt remains
 244 as to whether most of the other ponds have uneven bottom contours similar to Nagao Pond.

245 **Table 1. Average values of parameters measured at five sampling points in Nagao**
 246 **Pond over the study period**

	Date	Water depth (m)	Wind (m/s)	Temperature (°C)			Dissolved oxygen (mg/l)		TP (mg/l)	NH ₄ -N (mg/l)
				Air	Surface	Bottom	Surface	Bottom	Bottom	Bottom
June	2nd week	0.7 (0.5 - 0.9)	1.5 NW	24.2	29.4	28.5	4.8 (4.4 - 5.1)	4.8 (4.1 - 5.4)	NA	NA
	4th week	0.8 (0.5 - 1.0)	1.8 ESE	23.7	29.9	28.1	5.4 (4.8 - 5.9)	5.1 (4.9 - 5.1)	NA	NA
July	2nd week	NA	1.4 WNW	25.7	NA	NA	NA	NA	NA	NA
	4th week	0.8 (0.5 - 1.0)	1.6 NW	28.4	32.5	31.2	5.4 (5.1 - 5.8)	5.9 (5.1 - 6.2)	NA	NA
Aug.	2nd week	0.8 (0.4 - 0.9)	1.3 WNW	24.0	34.6	33.7	5.0 (3.7 - 6.0)	5.0 (4.3 - 5.4)	NA	NA
	4th week	0.7 (0.4 - 0.8)	1.4 ENE	28.0	30.3	29.9	5.8 (5.5 - 6.7)	5.9 (4.7 - 7.2)	NA	NA
Sept.	2nd week	0.7 (0.5 - 0.9)	1.0 NW	22.5	26.9	26.8	5.4 (5.5 - 6.1)	5.5 (5.2 - 6.3)	NA	NA
	4th week	0.7 (0.4 - 0.9)	1.8 ESE	23.8	25.1	24.8	5.8 (5.7 - 6.0)	5.9 (5.7 - 6.4)	NA	NA
Oct.	2nd week	0.7 (0.4 - 0.9)	1.5 NE	20.0	26.0	26.0	5.6 (5.2 - 6.1)	5.9 (5.2 - 6.5)	0.022	NA
	4th week	0.7 (0.4 - 1.0)	1.2 WNW	15.0	18.2	18.0	7.0 (6.7 - 7.3)	7.2 (6.8 - 7.5)	0.014	0.015
Nov.	2nd week	0.7 (0.4 - 0.9)	1.1 W	12.1	17.7	17.6	7.2 (6.8 - 7.7)	7.3 (6.9 - 7.8)	0.015	0.290
	4th week	0.7 (0.4 - 1.0)	1.4 E	9.6	16.7	16.3	7.6 (7.3 - 8.2)	7.6 (7.3 - 8.0)	0.016	0.180
Dec.	2nd week	0.7 (0.4 - 0.9)	1.3 W	7.7	13.4	13.4	8.0 (7.4 - 8.6)	8.1 (7.5 - 8.6)	0.009	0.425

TP - total phosphorus; NA- not available

247
 248
 249

4.2. Eutrophication

250 Eutrophication is the nutrient enrichment of waters that stimulates an array of symptomatic
 251 changes, including increased phytoplankton and rooted aquatic plant production, fisheries
 252 and water quality deterioration, and other undesirable changes that interfere with water uses
 253 [28]. Two primary nutrient cycles, phosphorus (P) and nitrogen (N), are generally focused on
 254 anthropogenic perturbations and their cumulative effects [29]. Some threshold values for
 255 eutrophication management range from 0.01 to 0.09 mg/l TP and 0.15 to 1.30 mg/l NH₄-N
 256 [29]. As seen in Table 1, the measured values are comparatively lower than the threshold

257 ones. On the other hand, Lake Kasumigaura, the largest shallow lake (about 4 m depth) in
 258 Japan, continues to show typical signs of eutrophication [30]. This is due to the increased
 259 nutrient loadings from urbanization, agricultural development and fishing culture [31].
 260 Although the Environmental Quality Standard of TP is set to 0.03 mg/l, the TP value
 261 increased from 0.04 mg/l to 0.10 mg/l in Lake Kasumigaura over the past 30 years;
 262 meanwhile, the total nitrogen concentration also increased from 0.8 mg/l to 1.3 mg/l [32].
 263 Continuous measurements of water quality parameters should be performed in Nagao Pond
 264 to infer the actual/potential occurrence of eutrophication.

265
 266 **4.3. Climatic effect on dissolved oxygen**

267
 268 As stated in sections 2.2 and 3, mainly wind-driven currents and temperature variations
 269 affect water circulation in shallow lakes and ponds. It is therefore assessed whether such
 270 climatic conditions are linked with the dissolved oxygen (DO) level near the bottom layer. As
 271 seen in Table 2, the temperature variations in both air and water are negatively correlated
 272 with the variation in bottom dissolved oxygen ($r \approx -0.9$), and the temperature difference
 273 between the surface layer and the bottom layer also shows moderate negative correlation (r
 274 ≈ -0.6); by contrast, the wind variation shows weak negative correlation ($r \approx -0.4$).
 275 It can therefore be concluded that both the air temperature and the water surface
 276 temperature dominantly affect the dissolved oxygen level in the bottom layer of a shallow
 277 pond under natural conditions accompanying changes in temperature and wind.

278
 279 **Table 2. Correlation coefficients (r) of parameters measured in Nagao Pond (p < 0.05).**

	Air temperature	Water surface temperature	Temperature difference between pond surface and bottom	Wind
- Dissolved oxygen in the pond bottom	-0.870	-0.922	-0.581	-0.369
- Correlation strength	Strong	Strong	Moderate	Weak

280
 281 As stated in section 2.2, pond stratification leads to the bottom layer receiving little oxygen
 282 from the atmosphere, which may become depleted of oxygen. However, it is possible to
 283 consider that stratification associated with temperature variation hardly occurs in a shallow
 284 pond because there is no statistic evidence ($p = 0.84$) showing a temperature difference
 285 between the surface layer and the bottom layer. This may have resulted in a weak
 286 correlation of the surface-bottom temperature difference with the dissolved oxygen level.
 287 Other studies developed in a 1.8 m-deep pond also show no clear gradient of temperature to
 288 a depth of 1.2 m [32]. Thus, it can be considered that a shallow pond is not thermally
 289 stratified.

290
 291 **5. CONCLUDING REMARKS**

292 The global water cycle is related closely to climate change. Fresh surface water accounts for
 293 just 1/10,000 of the total water available on the planet (section 1), and shallow lakes and
 294 ponds are worldwide the most abundant in lake types (section 2.2). It is accepted that wind-
 295 driven currents and thermal stratification mainly affect oxygen diffusion and its dissolved
 296 concentration in lakes (section 3). The present research aimed to verify whether this
 297 accepted view would be really observed. Our field survey in a 1 m-deep pond has
 298 demonstrated that (i) the temperature variations in the air and the pond water were
 299 negatively correlated with the dissolved oxygen level in the bottom layer (section 4.3); (ii)
 300 there was no statistic evidence showing a temperature difference between the surface layer
 301 and the bottom layer (section 4.3.); (iii) there was no statistic evidence for differences in
 302 dissolved oxygen concentration between the surface and bottom layers (section 4.1).

303 As to shallow reservoirs like ponds, a simple concept of the link between temperature and
304 dissolved oxygen is established through these findings – the oxygen solubility in water
305 decreases as the temperature increases [34].
306 As stated above, fresh water is a precious resource worldwide, and shallow lakes and ponds
307 are abundant worldwide. However, there have been few reports about ponds despite their
308 large number (section 3.1). As the presented research was often restricted by the global
309 pandemic, long-term observation is necessary to collect reliable data for underpinning the
310 proper management of water reservoirs.

323 REFERENCES

- 324
325 1. NOAA. Climate change impacts. Accessed 10 September 2022. Available:
326 <https://www.noaa.gov/education/resource-collections/climate/climate-change-impacts>.
327 2. IPCC - Intergovernmental Panel on Climate Change. Climate Change 2021: *The Physical*
328 *Science Basis, the Working Group I contribution to the Sixth Assessment Report*.
329 Cambridge: Cambridge University Press; 2021.
330 3. Stanford, B. D. *Water quality impacts of extreme weather-related events* (report No.
331 4C10PDF). Denver: Water Research Foundation; 2013.
332 4. Sargen, M. Biological Role of Water. Accessed 20 September 2022. Available:
333 [https://www.usgs.gov/special-topics/water-science-school/science/water-you-water-and-](https://www.usgs.gov/special-topics/water-science-school/science/water-you-water-and-human-body)
334 [human-body](https://www.usgs.gov/special-topics/water-science-school/science/water-you-water-and-human-body).
335 5. Mitchell HH, Hamilton TS, Steggerda FR, Bean HW. (1945). The chemical composition of
336 the adult human body and its bearing on the biochemistry of growth. *J Biol Chem*. 1945;
337 158 (3): 625-637.
338 6. Baumgartner DJ. Surface water pollution. In: Pepper I, Gerba C, Brusseau M, editors.
339 *Pollution Science*. San Diego (CA): Academic Press; 1996.
340 7. Mekonnen M, Hoekstra A. (2016). Four Billion People Facing Severe Water Scarcity. *Sci*
341 *Adv*. 2016; 2 (2): 1-6.
342 8. GCRT - Global Competitiveness and Risks Team. *The Global Risks Report 2017* (12th
343 edition). Geneva: World Economic Forum; 2017.
344 9. Chiew FH. Estimation of rainfall elasticity of streamflow in Australia. *Hydrol Sci J*. 2006; 51
345 (4): 613-625.
346 10. Lewis, W.M. (1983). A revised classification of lakes based on mixing. *Can J Fish Aquat*
347 *Sci*. 1983; 40 (10): 1779-1787.
348 11. Stewart KM, Walker KF, Likens GE. (2009). Meromictic lakes. In: Likens GE. Editor.
349 *Encyclopedia of Inland Waters*. Oxford: Academic Press; 2009.
350 12. LBERI - Lake Biwa Environmental Research Institute. Circulation and influence in Lake
351 Biwa. Accessed 15 July 2022. Available: <https://www.lberi.jp/learn/layers>.
352 13. Hakala A. *Paleoenvironmental and paleoclimatic studies on the sediments of Lake Vähä-*
353 *Pitkusta and observations of meromixis*. PhD thesis. Helsinki: University of Helsinki;
354 2005.

- 355 14. Hakala A. Meromixis as a part of lake evolution – observations and a revised
356 classification of true meromictic lakes in Finland. *Boreal Environ Res.* 2004; 9: 37-53.
- 357 15. Lampert W, Sommer U. *Limnoecology*. Oxford: Oxford University Press; 2007.
- 358 16. Global Nature Fund. Lake Biwa- Japan. Accessed 15 September 2022. Available:
359 <https://www.globalnature.org/en/living-lakes/asia/lake-biwa>.
- 360 17. Nakanishi M, Sekino T. Recent drastic changes in Lake Biwa bio-communities, with
361 special attention to exploitation of the littoral zone. *GeoJournal*. 1996; 40 (1); 63-67.
- 362 18. Hosono T, Nakano T, Igeta A, Tayasu I, Tanaka T, Yachi S. (2007). Impact of fertilizer
363 on a small watershed of Lake Biwa: Use of sulfur and strontium isotopes in environmental
364 diagnosis. *Sci Total Environ.* 2007; 384 (1-3): 342-354.
- 365 19. Chen X, Chen Y, Shimizu T, Niu J, Nakagami K, Qian X, Jia B, Nakajima J, Han J, Li J.
366 Water resources management in the urban agglomeration of the Lake Biwa region,
367 Japan: An ecosystem services-based sustainability assessment. *Sci Total Environ.* 2017;
368 586: 174-187.
- 369 20. Climate Policy Watcher. Lake ecosystem. Accessed 10 July 2022. Available:
370 [https://www.climate-policy-watcher.org/lake-ecosystems/definition-of-shallow-lakes-and-](https://www.climate-policy-watcher.org/lake-ecosystems/definition-of-shallow-lakes-and-ponds-and-world-distribution.html)
371 [ponds-and-world-distribution.html](https://www.climate-policy-watcher.org/lake-ecosystems/definition-of-shallow-lakes-and-ponds-and-world-distribution.html).
- 372 21. Macintyre, S. and Melack, J. M. (1995). Vertical and horizontal transport in lakes: linking
373 littoral, benthic, and pelagic habitats. *J North Am Benthol Soc.* 1995; 14: 599-615.
- 374 22. Spigel RH, Imberger J, Rayner KN. (1986). Modeling the diurnal mixed layer, *Limnol*
375 *Oceanogr.* 1986; 31: 533-556.
- 376 23. Rural Development Bureau (2021). *Tameike* [water reservoir], Tokyo: Ministry of
377 Agriculture, Forestry and Fisheries; 2021. Japanese.
- 378 24. Yamagishi T. Effective measures for water quality improvement in an eutrophic pond of
379 urban park, *J Japan Biol Soc Water Waste.* 2015; 51 (1): 19-28.
- 380 25. JMA - Japan Metrological Agency. Climate data. Accessed 5 June 2022. Available:
381 [https://www.data.jma.go.jp/obd/stats/etrn/view/daily_a1.php?prec_no=60&block_no=058](https://www.data.jma.go.jp/obd/stats/etrn/view/daily_a1.php?prec_no=60&block_no=0586&year=2022&month=09&day=&view=p1)
382 [6&year=2022&month=09&day=&view=p1](https://www.data.jma.go.jp/obd/stats/etrn/view/daily_a1.php?prec_no=60&block_no=0586&year=2022&month=09&day=&view=p1)
- 383 26. World Weather & Climate Information. Climate in Otsu. Accessed 10 June 2022.
384 Available: [https://weather-and-climate.com/average-monthly-Rainfall-Temperature-](https://weather-and-climate.com/average-monthly-Rainfall-Temperature-Sunshine,otsu-shiga-jp)
385 [Sunshine,otsu-shiga-jp](https://weather-and-climate.com/average-monthly-Rainfall-Temperature-Sunshine,otsu-shiga-jp).
- 386 27. EPA - Environmental Protection Agency (2022). Dissolved Oxygen and Biochemical
387 Oxygen Demand. Accessed 11 November 2022. Available:
388 <http://water.epa.gov/type/rsl/monitoring/vms52.cfm>.
- 389 28. Prepas EE, Charette T. (2003). Worldwide eutrophication of water bodies: causes,
390 concerns, controls. *Treatise Geochem.* 2003; 9: 311-331.
- 391 29. Morishita I. *Ecology in dam lake*. Tokyo: Sankaido Publishing; 1983.
- 392 30. Nakayama T, Watanabe M. Re-evaluation of groundwater dynamics about water and
393 nutrient budgets in Lake Kasumigaura, *Proc Hydraul Eng.* 2005; 49: 1231-1236.
- 394 31. Matsuoka Y, Goda T, Naito M. (1986). An eutrophication model of Lake Kasumigaura,
395 *Ecol Modell.* 1986; 31 (1-4): 201-219.
- 396 32. Ibaraki Prefecture. Water quality in Kasumigaura. Accessed 10 October 2022. Available:
397 [https://www.pref.ibaraki.jp/soshiki/seikatsukankyo/kasumigauraesc/18_foreignlanguage/d](https://www.pref.ibaraki.jp/soshiki/seikatsukankyo/kasumigauraesc/18_foreignlanguage/documents/02_water%20quality%20in%20lake%20kasumigaura.pdf)
398 [ocuments/02_water%20quality%20in%20lake%20kasumigaura.pdf](https://www.pref.ibaraki.jp/soshiki/seikatsukankyo/kasumigauraesc/18_foreignlanguage/documents/02_water%20quality%20in%20lake%20kasumigaura.pdf).
- 399 33. Ushiroda, T. and Hashimoto, T. (2011). Analysis of dissolved oxygen deficiency and
400 eutrophication in shallow pond, *Bull Hiroshima Tech Res Inst.* 2011; 19: 27-36.
- 401 34. Wetzel RG. *Limnology: Lake and River Ecosystems*. 3rd ed. San Diego (CA): Academic
402 Press; 2001.