

1 **Phenylpropanoids from the immature fruits of black nightshade (*Solanum***  
2 ***nigrum* L.)**

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4

5 **Abstract**

6 Black nightshade (*Solanum nigrum* L.) has been traditionally used as indigenous  
7 Chinese medicine. Phytochemical investigations of the immature fruits of *S. nigrum*  
8 have been carried out and nine phenylpropanoids (**1-9**) were obtained. Their structures  
9 were elucidated on the basis of spectroscopic and chemical methods, including MS,  
10 <sup>1</sup>H and <sup>13</sup>C-NMR. They were identified as (7*S*, 8*R*)-4-[3-hydroxymethyl-5-(3-  
11 hydroxypropyl)-2,3-dihydrobenzofuran-2-yl]-2-methoxyphenol (**1**), (7*S*, 8*R*)-  
12 dihydrodehydroconifery alcohol (**2**), massonianoside A (**3**), butane-2,3-diol 2-O-  
13 (6-O-caffeoyl)-β-D-glucopyranoside (**4**), 4-[(6-O-(*E*)-caffeoyl)-β-D-glucopyranosyl]  
14 vanillic acid (**5**), (+)-isolariciresinol (**6**), *trans*-cinnamic acid (**7**), ferulic acid (**8**), and  
15 4-hydroxy cinnamic acid (**9**). This study enriched the chemical constituents of black  
16 nightshade.

17 .

18 **Keywords:** Black nightshade, *Solanum nigrum* L., immature fruits, chemical  
19 investigation, phenylpropanoids

## 20 **1. INTRODUCTION**

21 Black nightshade (*Solanum nigrum* L.) belongs to the family of Solanaceae and is  
22 widely distributed all over the world [1]. In China, it has been used as a common  
23 traditional Chinese medicine, which has the effects of clearing heat and detoxification,  
24 activating blood circulation and removing blood stasis, promoting water and swelling,  
25 and is mainly used to treat cold and fever, toothache, and cancers. Modern  
26 pharmacological studies indicated that *S. nigrum* exhibited a variety of biological  
27 activities including antiproliferative, anti-inflammatory [2,3], antiviral,  
28 hepatoprotective, and antioxidative activities [4-6].

29 *S. nigrum* contains steroidal saponins, steroidal alkaloid glycosides and phenolic  
30 compounds in the previous chemical studies [7-10]. Steroidal alkaloid glycosides are  
31 the main chemical components of *Solanum* species, which possess various  
32 pharmacological activities such as antiproliferative and anti-inflammatory properties  
33 [11,12]. Currently, most of the chemical investigations on *S. nigrum* were focused on  
34 the aerial parts, while the bioactive components of its unripe berries are still unclear.

35 In this study, the phytochemicals from the immature fruits of *S. nigrum* has been  
36 carried out and nine phenylpropanoids were obtained.

37

## 38 **2. MATERIALS AND METHODS**

### 39 **2.1. General experimental procedures**

40 MS spectra were obtained on an Acquity UPLC-Q-TOF Microsystem (Waters Co.,  
41 Milford, MA). NMR spectra were taken on a Bruker Avance III 500 MHz  
42 spectrometer (Bruker, Switzerland). ODS packed column (40–60 µm, Merck KGaA,  
43 Darastadt, Germany) and column chromatography was employed on silica gel (Anhui  
44 Liangchen Silicon Source Material Co. Ltd, Lu'an, China). All other analytical

45 chemicals and reagents were purchased from Sinopharm Chemical Reagent Co. Ltd.  
46 (Shanghai, China).

47

## 48 **2.2. Plant materials**

49 The immature fruits of *S. nigrum* were purchased from Haerbin (Heilongjiang  
50 province, China) and dried at room temperature in the shade. The voucher specimen  
51 has been deposited in the School of Pharmacy, Guangdong Pharmaceutical University,  
52 China.

53

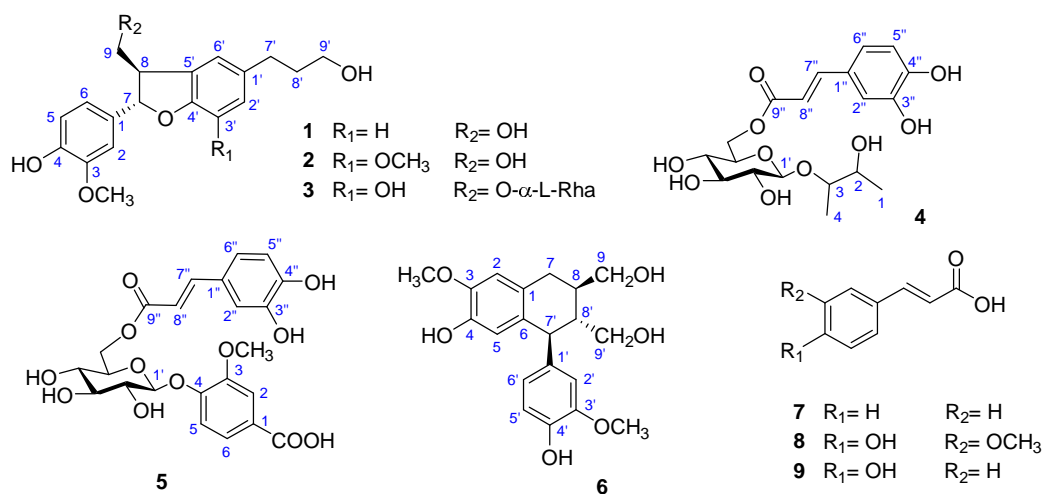
## 54 **2.3. Extraction and isolation**

55 Dried immature fruits of *S. nigrum* (2.5 kg) were extracted with 70% EtOH (v/v, 15  
56 L × 3) for 2 h. The ethanol-free suspension was subjected to a D101 macroporus resin  
57 column (80 × 1100 mm), and eluted with H<sub>2</sub>O, 10% MeOH, 30% MeOH, 50% MeOH,  
58 70% MeOH and MeOH to give six fractions (I-VI). The 70% MeOH elution (fraction  
59 V, 45.1 g) was separated by a silica gel chromatography column (200-300 mesh, 1100  
60 g) into 15 fractions (V-1 to V-15) with a CHCl<sub>3</sub>-MeOH gradient (100:1 to 0:1, v/v).

61 Compounds **7** (16.2 mg), **8** (10.1 mg), and **9** (16.0 mg) were obtained from fraction  
62 V-8 followed by an ODS MPLC, gradiently eluted with MeOH-H<sub>2</sub>O (1:9 to 10:0, v/v).  
63 Fraction V-6 was applied to an ODS MPLC eluted with a gradient of MeOH-H<sub>2</sub>O (1:9  
64 to 10:0, v/v) to afford eight subfractions (V-6-1 to V-6-8). Subfraction V-6-5 was  
65 further purified by a semi-preparative HPLC to obtained compounds **1** (8.2 mg) and **3**  
66 (7.3 mg). Fraction V-7 was subjected to an ODS MPLC and eluted with MeOH-H<sub>2</sub>O  
67 gradient (1:9 to 10:0, v/v) to afford ten fractions (V-7-1 to IV-7-10). Subfraction V-7-6  
68 was further separated by an ODS MPLC and semi-preparative HPLC to obtained  
69 compounds **2** (10.7 mg), **4** (6.3 mg), **5** (9.0 mg), and **6** (5.6 mg), respectively.

### 70 3. RESULTS AND DISCUSSION

71 The seventy percentage ethanol extract of the immature fruits of *S. nigrum* was  
72 separated successively by column chromatography on D101 macroporus resin, silica  
73 gel, ODS MPLC, and preparative HPLC, to afford nine phenylpropanoids (**1-9**) (Fig.  
74 1). Their structures were elucidated on the basis of spectroscopic data, including MS,  
75 and  $^1\text{H}$  and  $^{13}\text{C}$ -NMR.



76

77 **Fig. 1** Structures of the phenylpropanoids **1-9**

78 Compound **1**, brownish oil, was a blue fluorescence under 365 nm after TLC  
79 development. Its molecular formula was determined as C<sub>19</sub>H<sub>22</sub>O<sub>5</sub> based on its ESI-MS  
80 with the ion  $m/z$  353 [M+Na]<sup>+</sup>,  $^1\text{H}$  and  $^{13}\text{C}$ -NMR.

81 In the  $^1\text{H}$ -NMR, protons at  $\delta_{\text{H}}$  7.01 (1H, d,  $J$ =1.8 Hz, H-2), 6.87 (1H, dd,  $J$ =8.2, 1.8  
82 Hz, H-6) and 6.80 (1H, d,  $J$ =8.2 Hz, H-5) consisted an ABX coupling system of in a  
83 1,3,4-trisubstituted benzene ring. Signals at  $\delta_{\text{H}}$  7.13 (1H, s, H-2'), 7.00 (1H, d,  $J$ =8.1  
84 Hz, H-6') and 6.70 (1H, d,  $J$ =8.1 Hz, H-5') came from another ABX coupling system  
85 benzene ring in the molecule.  $\delta_{\text{H}}$  5.48 (1H, d,  $J$ =6.4 Hz, H-7) was a proton linked to  
86 oxygenated carbon.  $\delta_{\text{H}}$  3.80 (3H, s, 3-OCH<sub>3</sub>) was a methoxy group.

87  $^{13}\text{C}$ -NMR gave 19 carbon signals (Table 1). In aromatic region ( $\delta_{\text{C}}$  159.1 to 109.5),  
88 there were 12 carbon signals, which were two benzene ring units. Carbon signal at  $\delta_{\text{C}}$

89 56.4 in high field region is a methoxy carbon signal, which was consistent with the  
90 information given by  $^1\text{H-NMR}$ .  $\delta_{\text{C}}$  87.8, 54.8 and 64.8 were deduced as the carbon  
91 signals on the furan ring, and  $\delta_{\text{C}}$  36.1, 32.4 and 61.9 were form a hydroxypropyl  
92 group. Taken together, compound **1** was speculated a benzodihydrofuran lignin. Based  
93 on above analysis and the literature [13], compound **1** was identified as (7*S*,  
94 8*R*)-4-[3-hydroxymethyl-5-(3-hydroxypropyl)-2,3-dihydrobenzofuran-2-yl]-2-methox  
95 yphenol.

96 The molecular formula of compound **2** was determined as  $\text{C}_{20}\text{H}_{24}\text{O}_6$  based on its  
97 ESI-MS with the ion  $m/z$  383  $[\text{M}+\text{Na}]^+$ ,  $^1\text{H}$  and  $^{13}\text{C-NMR}$ . In the  $^1\text{H-NMR}$ , protons at  
98  $\delta_{\text{H}}$  7.03 (1H, d,  $J=1.8$  Hz, H-2), 6.88 (1H, dd,  $J=8.1, 1.8$  Hz, H-6) and 6.81 (1H, d,  
99  $J=8.1$  Hz, H-5) consisted an ABX coupling system of in a 1,3,4-trisubstituted benzene  
100 ring. Signals at  $\delta_{\text{H}}$  6.74 (1H, s, H-2') and 6.76 (1H, d,  $J=8.1$  Hz, H-6') came from  
101 another benzene ring in the molecule.  $\delta_{\text{H}}$  5.50 (1H, d,  $J=6.4$  Hz, H-7) was a proton  
102 linked to oxygenated carbon.  $\delta_{\text{H}}$  3.82 (3H, s, 3-OCH<sub>3</sub>) and 3.81 (3H, s, 3'-OCH<sub>3</sub>) were  
103 two methoxy groups.

104 In the  $^{13}\text{C-NMR}$  (Table 1), there were 20 carbon signals, including 12 carbon  
105 signals in aromatic region ( $\delta_{\text{C}}$  148.4 to 110.6), which were two benzene ring units.  
106 Carbon signals at  $\delta_{\text{C}}$  56.4 and 56.3 in high field region were two methoxy carbons,  
107 which were consistent with the information given by  $^1\text{H-NMR}$ .  $\delta_{\text{C}}$  88.3, 55.2 and 64.8  
108 were deduced as the carbon signals on the furan ring, and  $\delta_{\text{C}}$  36.1, 32.8 and 61.8 were  
109 form a hydroxypropyl group. Therefore, compound **2** was speculated a  
110 benzodihydrofuran lignin. Based on above analysis and the literature [14], compound  
111 **2** was identified as (7*S*, 8*R*)-dihydrodehydroconifery alcohol.

112

113

**Table 1**  $^{13}\text{C}$  NMR data of compounds **1-6** ( $\delta$  in ppm and  $J$  in Hz)

NO.	<b>1</b> <sup>a</sup>	<b>2</b> <sup>a</sup>	<b>3</b> <sup>a</sup>	<b>4</b> <sup>b</sup>	<b>5</b> <sup>b</sup>	<b>6</b> <sup>a</sup>
1	134.9	134.8	132.6	15.1	123.7	128.5
2	110.5	110.6	110.2	78.6	112.7	111.8
3	148.5	147.4	146.4	68.9	149.1	146.5
4	147.2	148.4	147.7	19.5	151.6	145.3
5	115.6	115.7	114.8		114.9	116.9
6	119.6	119.6	118.6		123.4	138.5
7	87.8	88.3	87.2		167.9	33.8
8	54.8	55.2	50.9			40.4
9	64.8	64.8	68.9			65.9
1'	135.4	130.1	128.3	101.3	100.5	133.9
2'	125.8	113.9	115.4	73.4	73.2	113.6
3'	129.1	147.2	140.8	76.6	75.8	148.3
4'	159.1	144.9	144.7	70.5	70.3	145.9
5'	109.5	136.4	135.2	73.8	73.7	115.6
6'	129.3	117.7	116.2	63.8	64.0	122.8
7'	32.4	36.1	34.8			48.1
8'	36.1	32.8	31.5			48.4
9'	61.9	61.8	60.3			62.1
1''			100.2	125.6	125.9	
2''			70.9	113.9	115.3	
3''			72.0	145.8	145.3	
4''			70.6	148.7	147.4	
5''			68.6	115.9	116.3	
6''			18.1	121.5	120.5	
7''				145.4	144.1	
8''				114.9	115.2	
9''				166.5	165.6	
3-OCH <sub>3</sub>	56.4	56.4	55.5		55.6	56.2
3'-OCH <sub>3</sub>		56.3				56.3

115 <sup>a</sup> 151 MHz for  $^{13}\text{C}$  NMR in Acetone- $d_6$ . <sup>b</sup> 101 MHz for  $^{13}\text{C}$  NMR in DMSO- $d_6$ .

116 Compound **3**, brownish oil, was a blue fluorescence under 365 nm after TLC  
 117 development. Its molecular formula was determined as  $\text{C}_{25}\text{H}_{32}\text{O}_{10}$  based on its  
 118 ESI-MS with the ion  $m/z$  515  $[\text{M}+\text{Na}]^+$ ,  $^1\text{H}$  and  $^{13}\text{C}$ -NMR.

119 In the  $^1\text{H}$ -NMR, protons at  $\delta_{\text{H}}$  6.93 (1H, d,  $J=1.3$  Hz, H-2), 6.86 (1H, dd,  $J=8.2$ ,  
 120 1.3 Hz, H-6) and 6.81 (1H, d,  $J=8.2$  Hz, H-5) consisted an ABX coupling system of in  
 121 a 1,3,4-trisubstituted benzene ring. Signals at  $\delta_{\text{H}}$  6.50 (2H, br.s, H-2', 6') came from

122 another benzene ring in the molecule.  $\delta_{\text{H}}$  5.37 (1H, d,  $J=6.6$  Hz, H-7) was a proton  
123 linked to oxygenated carbon.  $\delta_{\text{H}}$  3.76 (3H, s, 3-OCH<sub>3</sub>) and 1.13 (3H, d,  $J=6.2$  Hz,  
124 H-6") were a methoxy and methyl groups in the higher field.  $\delta_{\text{H}}$  4.62 (1H, d,  $J=0.7$  Hz,  
125 H-1 ") was deduced as a terminal proton signal of sugar.

126 In the <sup>13</sup>C-NMR (Table 1), there were 25 carbon signals, including 12 carbon  
127 signals in aromatic region ( $\delta_{\text{C}}$  144.7 to 110.2), which were two benzene ring units.  
128 Carbon signal at  $\delta_{\text{C}}$  55.5 in high field region was a methoxy carbon, which was  
129 consistent with the information given by <sup>1</sup>H-NMR. Compared with the NMR data of  
130 compound 2, it is speculated that compound 3 is a benzodihydrofuran lignin with  
131 rhamnose substitution at C-9 position. Based on above analysis and the literature [15],  
132 compound 3 was identified as massonioside A.

133 Compound 4 was brownish oil and exhibited a blue fluorescence under 365 nm  
134 after TLC development. Its molecular formula was determined as C<sub>19</sub>H<sub>26</sub>O<sub>10</sub> based on  
135 its ESI-MS with the ion  $m/z$  437 [M+Na]<sup>+</sup>, <sup>1</sup>H and <sup>13</sup>C-NMR.

136 In the <sup>1</sup>H-NMR spectrum, signals at  $\delta_{\text{H}}$  7.50 (1H, d,  $J=15.6$  Hz, H-7") and 6.26 (1H,  
137 d,  $J=15.6$  Hz, H-8") were speculated to be the proton signals on the trans  
138 carbon-carbon double bond. The aromatic region had three proton signals at  $\delta_{\text{H}}$  7.04  
139 (1H, s, H-2 "), 6.99 (1H, d,  $J=7.8$  Hz, H-5 ") and 6.76 (1H, d,  $J=7.4$  Hz, H-6 "). Signal  
140 at  $\delta_{\text{H}}$  4.26 (1H, d,  $J=7.7$  Hz, H-1') was the terminal proton signal of sugar.  $\delta_{\text{H}}$   
141 4.40~2.98 has 8 proton signals, which were proton signals on oxymethylene or  
142 oxymethylene.  $\delta_{\text{H}}$  1.07 (3H, d,  $J=6.0$  Hz, H-1) and 0.99 (3H, d,  $J=6.0$  Hz, H-4) were  
143 two methyls in the high field.

144 <sup>13</sup>C-NMR spectrum indicated 19 carbon signals, and  $\delta_{\text{C}}$  166.5 was an ester carbonyl  
145 in the lower field (Table 1). Signals at  $\delta_{\text{C}}$  166.5, 148.7, 145.8, 145.4, 125.6, 121.5,  
146 115.9, 114.9 and 113.9 consisted of a caffeic acyl group. Meanwhile, compound 4 has

147 a glucose unit, which the carbon signals at  $\delta_C$  101.3, 73.4, 76.6, 70.5, 73.8 and 63.8.  
148 The remaining four carbon signals at  $\delta_C$  78.6, 68.9, 19.5 and 15.1 formed a  
149 3-hydroxy-2-butoxy group. Based on above analysis and the literature [16],  
150 compound **4** was identified as butane-2,3-diol 2-O-(6-O-caffeoyl)- $\beta$ -D-  
151 glucopyranoside.

152 Compound **5** was brownish oil and exhibited a blue fluorescence under 365 nm  
153 after TLC development. Its molecular formula was determined as  $C_{23}H_{24}O_{12}$  based on  
154 its ESI-MS with the ion  $m/z$  491  $[M-H]^-$ ,  $^1H$  and  $^{13}C$ -NMR.

155 In the  $^1H$ -NMR spectrum, signals at  $\delta_H$  7.48 (1H, d,  $J=15.7$  Hz, H-7'') and 6.36 (1H,  
156 d,  $J=15.7$  Hz, H-8'') were speculated to be the proton signals on the *trans*  
157 carbon-carbon double bond. The aromatic region had six proton signals at  $\delta_H$  7.40 (1H,  
158 dd,  $J=8.2, 1.7$  Hz, H-6), 7.38 (1H, d,  $J=1.7$  Hz, H-2), 7.37 (1H, d,  $J=1.8$  Hz, H-2''),  
159 7.18 (1H, dd,  $J=8.2, 1.8$  Hz, H-6''), 6.84 (1H, d,  $J=8.2$  Hz, H-5) and 6.77 (1H, d,  
160  $J=8.2$  Hz, H-5''), which speculated that the compound had two 1,3,4-trisubstituted  
161 benzene rings. Signal at  $\delta_H$  5.08 (1H, d,  $J=7.3$  Hz, H-1') was the terminal proton signal  
162 of sugar.  $\delta_H$  3.73 (3H, s, 3-OCH<sub>3</sub>) was a methoxyl group in the high field.

163 The  $^{13}C$ -NMR spectrum gives 23 carbon signals (Table 1). Compared with the  
164 NMR data of compound **4**, compound **5** contained a caffeic acyl segment and a  
165 glucose unit. The chemical shifts of C-1' and C-6' of the glucose shifted to the low  
166 field, which speculated that there were substituted on C-1' and C-6'. Meanwhile,  
167 compound **5** had a vanillic acid fragment with substitutions at position 4. Based on  
168 above analysis and the literature [17], compound **5** was identified as  
169 4-[(6-O-(*E*-caffeoyl)- $\beta$ -D-glucopyranosyl] vanillic acid.

170 Compound **6** was brownish oil and its molecular formula was determined as  
171  $C_{20}H_{24}O_6$  based on its ESI-MS with the ion  $m/z$  359  $[M-H]^-$ ,  $^1H$  and  $^{13}C$ -NMR. In the

172 <sup>1</sup>H-NMR spectrum, signals in aromatic region at  $\delta_{\text{H}}$  6.78 (1H, s, H-5'), 6.76 (1H, s,  
173 H-2'), 6.66 (1H, s, H-2) and 6.22 (1H, s, H-5) were protons on the benzene ring. In  
174 the high field,  $\delta_{\text{H}}$  3.78 (3H, s, 3-OCH<sub>3</sub>) and 3.81 (3H, s, 3'-OCH<sub>3</sub>) were two methoxy  
175 signals.

176 The <sup>13</sup>C-NMR spectrum gave 20 carbon signals (Table 1), and there had 12 carbon  
177 signals in the aromatic region, which were two benzene ring units.  $\delta_{\text{C}}$  56.2 and 56.3  
178 were two methoxy groups, which were consistent with the information given by  
179 <sup>1</sup>H-NMR spectrum. The remaining six carbon signals at  $\delta_{\text{C}}$  65.9, 62.1, 48.4, 48.1, 40.4  
180 and 33.78 were speculated that it was the alkyl carbon signal on the C<sub>3</sub> skeleton of  
181 phenylpropanoid. Based on above analysis and the literature [18], compound **6** was  
182 identified as (+)-isolariciresinol.

183 Compounds **7** to **9** were identified as *trans*-cinnamic acid (**7**) [19], ferulic acid (**8**)  
184 [20], and 4-hydroxy cinnamic acid (**9**) [21], respectively, based on their  
185 spectroscopic analysis.

186

#### 187 4. CONCLUSIONS

188 Further chemical investigation of the immature fruits of *S. nigrum* led to the  
189 isolation of nine phenylpropanoids. Their structures were elucidated on the basis of  
190 spectroscopic and chemical methods. They were identified as compound **1** was  
191 identified as (7*S*, 8*R*)-4-[3-hydroxymethyl-5-(3-hydroxypropyl)-  
192 -2,3-dihydrobenzofuran-2-yl]-2-methoxyphenol, (7*S*, 8*R*)-dihydrodehydroconifery  
193 alcohol, massonioside A (**3**), butane-2,3-diol 2-O-(6-O-caffeoyl)- $\beta$ -D-  
194 glucopyranoside, 4-[(6-O-(*E*)-caffeoyl)- $\beta$ -D-glucopyranosyl] vanillic acid,  
195 (+)-isolariciresinol, *trans*-cinnamic acid, ferulic acid, and 4-hydroxy cinnamic acid.

196

197 **CONSENT**

198 It is not applicable.

199

200 **ETHICAL APPROVAL**

201 It is not applicable.

202

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207

208 **COMPETING INTERESTS**

209 Authors have declared that no competing interests exist.

210

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