

Studies on Phosphorous **dynamics** in soils (: A Review=delete)

Abstract:

More than two-thirds of the world's soils are afflicted with phosphorus shortage (P). Agricultural output suffers because phosphorus is unavailable to all but a tiny percentage of plants. To lessen the severity of the environmental impact, phosphorus-deficient soils have traditionally been fertilized. Improving agricultural management practices for sustainable crop output, including **lowering phosphorus runoff, is the emphasis of this study.** Investigating chemical fractionation mechanisms to distinguish between **occluded P, acid-extractable calcium-bonded P, and non-occluded P** is necessary for increasing **inorganic phosphorous, Pi** in mung bean (*Vigna radiata*) types and establishing net grazing systems. Phosphate (P) is known to be transferred between pools due to weathering, with the highest P retention rates found in clay-rich soils. Soil with a finer texture is better able to absorb and fix phosphorus, which means more nutrients and water can be made available to the plant when **Mung bean** varieties are inoculated into it. More photosynthesis means more accumulated dry stuff. The P-treated variety of **Mung bean** had the greatest yield index (13.28). Pods per plant (46.02), pods per crop (8.20), test weight (40.63 gm), pod weight (8.0 g), and seed weight (1.0 g) were all significantly **increased to (1.83 g) when fertilized with Diammonium phosphate, DAP.** An increase in nodule count, leaf area index, plant height, grain yield, total chlorophyll content, and straw output of up to 40% was observed at higher P₂O₅ concentrations (2988 kilograms per hectare). Each plant developed the maximum number of nodules after receiving injections of **Phosphate solubilizing bacteria, PSB** and *Aspergillus awamori*. The highest levels of chlorophyll were found in its leaves, and its plants were the largest and most productive overall. Liming improved mungbean yields in acid piedmont soil by raising the pH and altering other chemical characteristics. They claim that increasing India's output of pulses is necessary for the country to attain food security in the future. If farmers and extension workers want to see higher quality pulse yields and

greater long-term profitability, they need to incorporate P nutrition into their balanced nutrient management **programs**.

Keywords: Phosphatic Fertilizer, grain yield, photosynthetic yield; plant productivity

Introduction:

Phosphorus (P) is a macronutrient required by plants for its growth. It accounts for between 0.2 and 0.8 percent of the dry weight of a plant. Phosphorus is necessary for a wide variety of physiological and biochemical processes in plants. **Phospholipids, nucleic acids, coenzymes, enzymes, and nucleotides include phosphorus** (Mehrvaz *et al.*, 2008). In the early stages of plant growth, enough P availability is also necessary for the laying down of the plant's reproductive parts' primordia (Satyaprakash *et al.*, 2017; Kalayu, 2019). The typical amount of phosphorus found in soil is 0.05 percent (w/w). However, only 0.1 percent of this phosphorus is available for plant uptake (Zhu *et al.*, 2011). Phosphorus fertilizers have traditionally been used as a method for addressing phosphorus deficiency in the soil. To maintain consistent amounts of phosphorus in soil and plant systems, fertilizers in 52,3 billion **tons** are sprayed yearly (FAO, 2017). This tremendous quantity of phosphorus, which has been transformed into inorganic phosphorus and precipitated by metal cations in the soil such as iron, **aluminum**, magnesium, and calcium, is only used by plants at a rate of around 0.2 percent, (or 10 micrograms=???- delete) (Soumare *et al.*, 2020). The phosphate deposit in agricultural soils is sufficient to ensure that optimum crop production may be maintained **Worldwide** for about 100 years (Kalay, 2019). In addition, the use of fertilizers contributes to a variety of environmental problems, including the contamination of groundwater and the eutrophication of streams (Yu, *et al.*, 2011 and Alori *et al.*, 2017). **Agricultural management strategies are needed to develop and improve the efficiency of phosphorus fertilization for increasing crop productivity, and reducing environmental pollution caused by phosphorus loss from the soil. Also these goals will be achieved by reducing the amount of phosphorus loss from the soil (write how to achieve this task).....**

Forms of Phosphorous in soil system:

It is common for soils to have greater concentrations of total P than other key nutrients, such as N or K. More than 80% of P, on the other hand, is inert and so inaccessible to plants (Xu *et al.*, 2020). Figure 1 shows that inorganic and organic forms of P (Pi/Po) are found in

soil, and the quantities of each alter over time. Most soil P comes from biological tissues where phosphorus is found in organic substances such as nucleotides, phospholipids, phosphoproteins, and coenzymes (Cross *et al.*, 1995). It is also possible that soil nutrient cycle mechanisms (e.g., nitrogen cycling) are responsible for the dispersion of primary Pi into Po forms. Because of the widespread usage of Po-containing products, including fire retardants, plasticizers, insecticides, and antifoam agents, new Po sources have emerged in the environment, increasing soil Po concentrations and diversity (Adams *et al.*, 1992).

Due to its weak connections with the soil particles, Po is more quickly leached than the more stable forms of primary (variscite, apatite, and stringier) and secondary minerals (calcium, aluminum phosphates, and iron) found in soil (Gebrim *et al.*, 2010). Chemical fractionation processes were created in the early studies to determine the three forms of Pi: occluded P (containing 2nd NaOH extractable P and residual Pi), acid-extractable calcium-bonded P and non-occluded P. Fe and Al oxide coatings and concretions absorb occluded P during diffusive penetration and the development of the soil. Occluded P is more difficult to remove from Fe, and Al oxide surfaces than non-occluded P ions that make up non-occluded P. Soil Pi content decreases from 684 g/g in Entisols to 200 to 430 in Ultisol and Oxisols **respectively** as the soils develop. **Plants or soil bacteria may take up Pi and use it in the secondary Po cycle, or it can seep into streams and deposit phosphorus (P) in ocean sediments** (Maltais-Landry *et al.*, 2014). A variety of types and amounts of Pi may be found in the soil. Plant structures (such stems, roots, and leaves), **soil of animals**, and microbes gradually return the P they absorbed to the soil when mineralization has occurred. It is important to note that even though P has a wide range of biochemical activities in soil that are crucial to the health of the ecosystem, its cycles are slower than those of C, N, and S. A wide range of geochemical processes (weathering, precipitation/dissolution, and solid-phase transitions) have a long-term impact on soil phosphorus availability and distribution. Soil microorganisms immobilize and mineralize plant-available phosphorus (P) in soil organic matter over a short time (**10–2 to 100 years????**). However, very little is known about the function these biological activities play in limiting soil phosphorus availability due to geochemical processes (Hesterberg *et al.*, 1999; Prietzel *et al.*, 2016 and Hao *et al.*, 2020.)

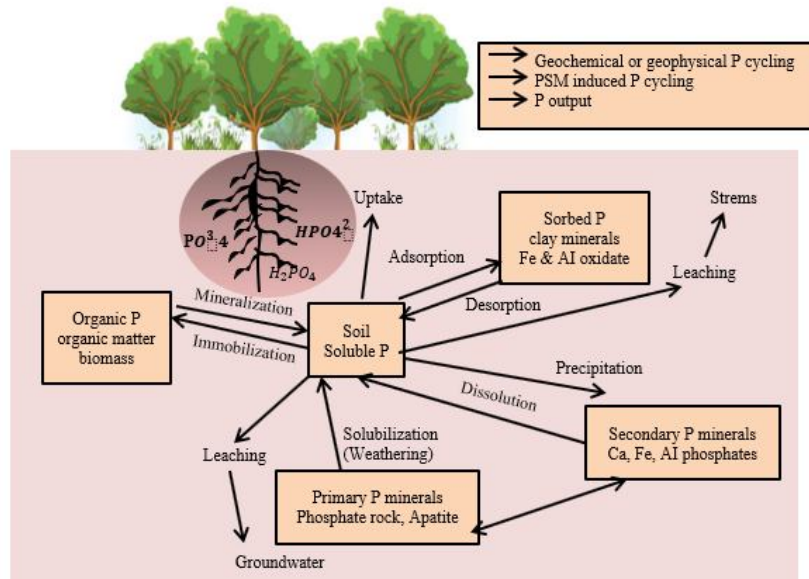


Figure 1: The Phosphorous cycle

Release of inorganic P in soil from fertilizer application:

Despite the fact that the soil contains a large amount of total P, only a little amount (less than 1%) gets dissolved in it (Bunemann *et al.*, 2015). There are many different forms of phosphorus, including organic (Po) which includes nucleic acids, phospholipids, and orthophosphate), Al/Fe-bound inorganic P (which is found in acidic soils), and Ca-bound inorganic **P which is found in alkaline soils**. Occluding-P and Fe/Al-P percentages rise with rising P concentrations in the soil (Bai *et al.* 2017). Progressively, these compounds release soluble P (0.1 percent of the total soil P), which is slowly released over time. Soluble P, labile soil P, and nonlabile soil P are all various types of P in the soil. For plants to flourish they must have both primary and secondary orthophosphate ($(\text{H}_2\text{PO}_4)^{-1}$ and HPO_4^{-2} forms of P (Zhu *et al.*, 2018). Acidic soils contain primary orthophosphates, whereas alkaline soils have secondary orthophosphates (Rashid *et al.*, 2005). Stunted growth, delayed maturity of crops, and poor seed development are all symptoms of a phosphorus deficit in plants. A shortage of total phosphorus in the soil, along with the fixation of extra phosphorus via the use of chemical fertilizers and organic sources such as manure, results in a deficiency of phosphorus in the soil. **The overall amount of phosphorus in soil may vary depending on several factors, including pH, organic matter content, particle size distribution, and mineralogy**

of the parent material (Havlin *et al.*, 2005). The most important controls on P dynamics in soil are the **sorption** and **desorption capacities of soil P**, **particle size distribution**, metal oxides, and soil pH. (Li *et al.*, 2016). It has been shown that weathering is a factor in the movement of P across various pools. The largest phosphate retention capacity is found in clay-rich soils, which subsequently adsorbs/fixes more P than coarse-textured soils (Brady *et al.*, 2008).

Phosphorus fertilizer is very necessary to take full use of the greatest production potentials offered by various crop plants. Since the ratio of phosphorus that is applied determines the amount of phosphorus that is fixed, the amount of phosphorus that is fixed by fertilizer that is broadcast is significantly higher than the amount of phosphorus that is fixed by fertilizer that is applied in bands due to the narrow soil to fertilizer ratio in the latter situation (Rashid *et al.*, 1993). One of the primary reasons for the low output in crops is the poor performance **of phosphatic fertilizer(edit the typical analytical data for phosphate fertilizers to identify the quality of the fertilizers, may they include uranium as an impurity).** **Fertilization** resulted in a more significant reaction than broadcast and banding. Mungbean fertigation yielded the maximum mungbean production, as well as P absorption, P recovery, and agronomic efficiency, compared to broadcasting methods. Phosphate fertilizer costs have risen worldwide due to the rising demand for greater yields, which has placed a strain on land resources (Weber *et al.*, 2014; Chowdhury *et al.*, 2017). Soil P has been replenished by using phosphate fertilizers. Rock phosphate mining is the primary source for making phosphate fertilizers. Chemical methods must dissolve the phosphate that can only be obtained from mined rock phosphate (Khan *et al.*, 2009). Because crops have a high capacity to fix phosphate and a low ability to utilize it, the rock phosphate is applied to the soil each cropping season (Withers *et al.*, 2001). Phosphate fertilizers that have been applied indiscriminately have created a fixed P pool that may be explored and used by various bacteria, enzymes, organic acids, manures, and humic substances. Organic acids and enzymatic activity released by phosphate solubilizing microorganisms aid plant absorption (Dash *et al.*, 2017). However, soil type, parent material, organic matter concentration, soil pH, soil salinity, and metal oxides all influence phosphate dynamics in the soil.

3. Dynamics of Phosphorous in the soil affected by organic sources:

Various phosphorus pools in the soil, such as Al/Fe₂ bound P or Ca-bound P, replenish the P that plants take up from the soil solution (Saleque *et al.*, 2004). With mass flow taking over from diffusion, the function of P in the process is diminished (Hinsinger, 2001). The availability of Fe, Al, and Ca, which aid in the fixation of P, is also influenced by soil pH. Sodium phosphate is reduced in alkaline soils by forming several insoluble phosphate derivatives. Phosphorus is depleted from acidic soils by precipitates of hydroxyl phosphates generated by the interaction of iron and aluminum with phosphate ions (Harris *et al.*, 2006). Cation and anion concentrations at the exchange sites influence soil adsorption capacity for phosphate, although divalent cations (Ca²⁺) are significantly more attracted to P adsorption in clays than monovalent cations (Na⁺) are in soils.

Alkaline calcareous soils have a high concentration of calcium phosphate minerals (CaP) that plants may take up, although most of the P is fixed as CaP. In addition to supplying P, clay minerals in soil play a significant role (Mehdi *et al.*, 2007). Both the fixation and transformation of phosphorus are intricate processes, and it may be challenging to pin down the appropriate kinetics for them. Phosphate may be in the soil in several different forms, including solution P, active P, as well as fixed P.

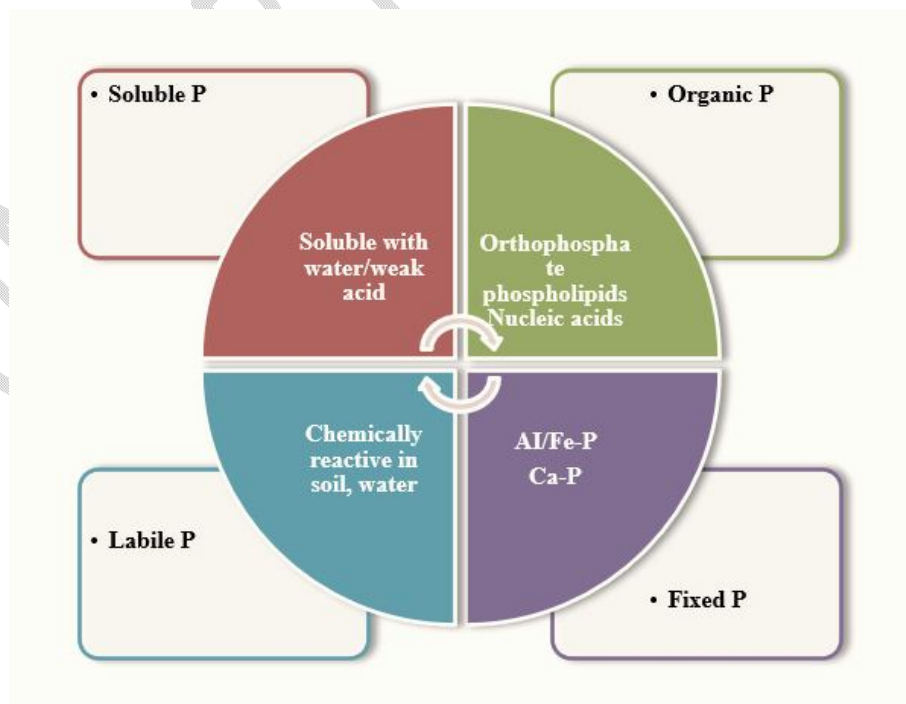


Figure 2: Various types of total Phosphorous in the soil matrix.

Phospholipids, inositol phosphates, and nucleic acids are the three different kinds of organic P compounds. Phospholipids are the most common kind. Inositol phosphates include mono- to hexaphosphate esters as their primary building blocks (Dash *et al.*, 2017). One of the most common organic phosphates (Ca–Mg salts of phytic acid) is also phytin. There are three kinds of enzymes that can break down organic P, which include phosphatases and phytases as well as phosphatases (Ingle and Padole, 2017).

As a result of using goat dung in combination with low-rates of P fertilizers, the amount of the less-labile NaOH-Pi fraction in the soil may be reduced, ensuring that the soil's labile P levels are maintained at an adequate level. P needs for Smallholder farmers may be met with this cost-effective technique for enhancing fertilizer P usage efficiency (Gichangi *et al.*, 2009).

After two and 16 weeks of composting in labile pools, it was shown that the Pi percentage soil was more likely to be dominated by compost P. However, the amount of labile P (51-58 percent), as well as total Pi (61-98 percent) in soil, was the same or nearly identical to that obtained from inorganic KH_2PO_4 alone. Compost from potato trash had a higher proportion of organic P (Po) than those from the other composts investigated when Po was extracted using NaHCO_3 . While this study indicated that Pi was the most necessary form of compost P, it is still hard to assess the availability of compost P after it is applied to soil (Gagnon *et al.*, 2012).

Additionally, it was shown that the soil formed from Basalt might lose some of its ability to bind P when organic molecules were introduced through the treatments. The intensity of P adsorption decreased when chicken dung as compost was added to soils originating from granite and river alluvial deposits but rose following the application of organic fertilizer. The three adsorption isotherm equations for basalt-derived soils all fit well. Langmuir >Freundlich >Temkin adsorption models were used for soils generated from granite and alluvial deposits as parent materials. It was shown that organic fertilizer and chicken manure had no effect on soils generated from basalt's MPBC or SPR, however organic fertilizer and

incubation time had a significant impact on these parameters in soils formed from granite and alluvial deposits (Yu *et al.*, 2013).

As in both soils, a sandy loam acidic soil (Aquepts) and a silt loam alkaline soil, extraction efficiency was in the sequence of Bray and Kurtz-II>Mehlich-3> Kelowna>Olsen amended with diverse organic sources, such as CM, PM, CW, and triple superphosphate (TSP). As evaluated by NaHCO_3 , organic additions improve the effectiveness of the additional P, but the efficiency of Triple superphosphate with days of incubation decreases (Ara *et al.*, 2018).

Nitrogen (N), as well as Leaf litter, had an effect on the bacterial action **and it** led to an increase in phosphorus orthophosphate (Po), which resulted in a rise in the NaOHPI fractions, as a result, the quantity of Po detected in the NaOHPI portion increased overall (Kunito *et al.*, 2018). To increase P availability, organic matter may reduce P-selectivity and the maximum phosphate buffering capacity. At a SOM concentration of 75.3 grams per kilogram (g/kg), there is a growing tendency toward greater availability.

The relationship between different sources and sinks of phosphorus by using an intense cropping system that included rice, maize, and green gram (residual) was determined. Inorganic P fractions were for 71.9–86.0 percent of total P in surface soil, whereas organic P fractions accounted for 14.0–28.1 percent. The percentages in the subsoil ranged from 75.9–81 percent and 18.1–24.1 percent. The following is the order of the pools: Up to the first 15 centimetres of the soil, the order is FeP> RS-P > Al >Ca > Occluded >Saloid P; however, the lowest 15–30 centimetres of soil have the following order: "Occluded P" = "Al-P" > "saloid P" in the RS-P Both fractions provide a large quantity of phosphorus to the surface layer of the soil. In terms of inorganic phosphorus forms, there is a positive association between surface soil P absorption and Fe-P as well as a significant relation between the subsurface soil P uptake and Al-P (Chauhan *et al.*, 2012).

Conclusion

Phosphorus plays an important role in pulses nutrition mainly for mungbean production. Most of the Indian soils are deficient in phosphorus leading to affect the nutrition of pulse crop as pulses occupy an important position in food and nutritional security in India. India produces over 280 million tonnes of foodgrains every year with an increase of four folds

since independence. Increased efforts to produce more food have resulted in tremendous shift in cropping systems towards cereal-cereal based systems. Accordingly, vision of Indian Institute of Pulse Research, 2030, the projected pulse requirement by the year 2030 would be 32 million tons with an anticipated required growth rate of 4.2%. The growing Indian population enhanced the pulses demand. The burgeoning human population in India needs higher pulses production for fulfilling the dietary protein requirement. It requires to mitigating the demand of protein. In this regard, balanced fertilization with NPK alongwith biofertilizers has been proved beneficial in pulses both under rainfed and irrigated conditions. Hence, phosphorous needs to be taken care of while balanced fertilization in pulses. A concerted effort by farmers, researchers, development agencies, and government are needed to ensure that India becomes self-sufficient in pulses in the future. Therefore, farmers and extension functionaries must recognize that it is highly needed to include P nutrition in their nutrient management practices in balanced manner in present day intensive farming to harvest higher pulse yields of superior quality earning maximum sustainable profitability.

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